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RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

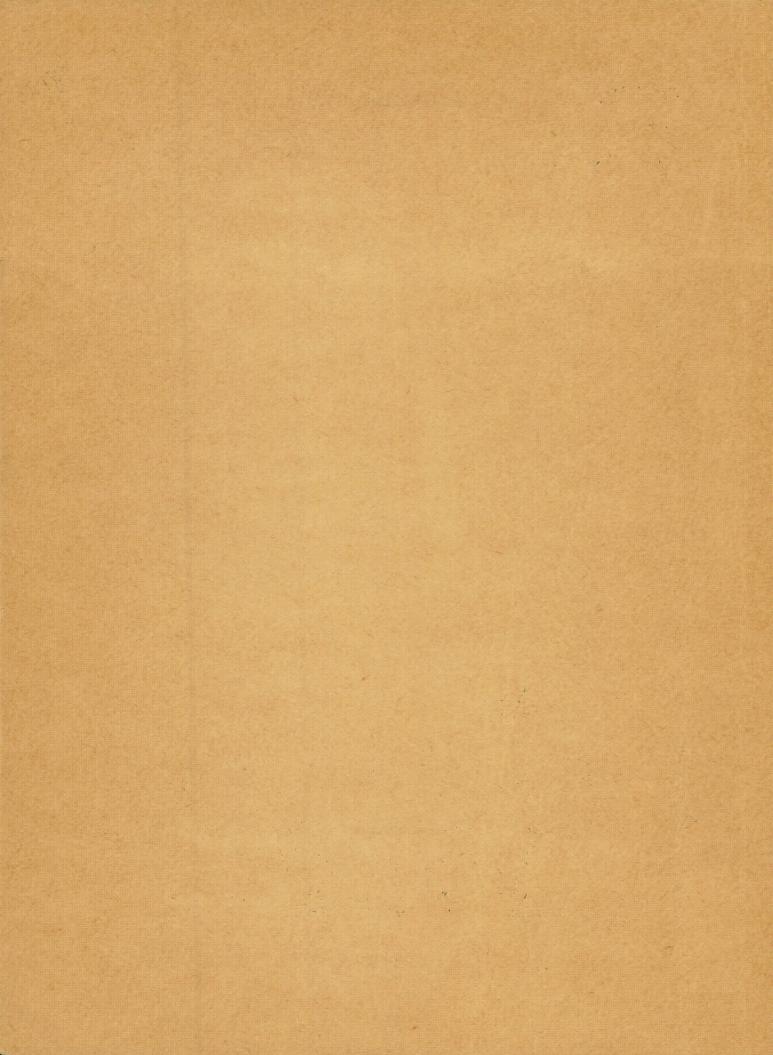
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REPORT 129

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

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AUG 2 3 1972
Dallas, Texas

Donald K. Leifeste, James F. Blakey, and Leon S. Hughes

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Prepared by the U.S. Geological Survey in cooperation with the Texas Water Development Board

TEXAS WATER DEVELOPMENT BOARD

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RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

ABSTRACT

The Red River, from its point of origin in eastern New Mexico to the northeast corner of Texas, drains an area of about 48,000 square miles. The total area in Texas is 24,500 square miles. From west to east the topography changes from the nearly flat surface of the High Plains, to a gently eastward-sloping plain dissected by prominent systems of drainage in the Osage Plains, to the low relief and gently gulfward slope of the West Gulf Coastal Plain.

The climate of the basin ranges from semiarid to humid; mean annual precipitation is less than 18 inches in the far western part and more than 46 inches in the extreme eastern part. Runoff increases from about 50 acre-feet per square mile at the 100th meridian to more than 800 acre-feet per square mile in the northeast corner of the State.

The dissolved-mineral content and chemical character of waters in the Red River basin vary widely from place to place and from time to time. Geologic factors, runoff and streamflow characteristics, and activities of man largely determine the nature and amount of dissolved material transported by the Red River and its tributaries. In the semiarid western part of the basin, base flow is usually nonexistent. However,

numerous seeps and springs in Permian rocks that crop out in this part of the basin account for much of the salt load in the Red River above Lake Texoma. The water quality of the main stem has been further degraded by oil-field brines. The eastern part of the basin is in an area of high rainfall and well-leached rocks and soils. Ground-water effluent is generally low in dissolved minerals, and the dissolved-solids content of streamflow varies only slightly with discharge.

The highly mineralized waters from salt sources in the western part of the basin cause the water of the Red River to be undesirable for public supply throughout most of its reach in Texas. Storage of good-quality water in existing and proposed reservoirs on tributaries to the Red River will increase degradation of water quality in the main stem, especially above Lake Texoma. Even if releases are made from tributary impoundments in the western part of the basin, evaporation during impoundment and waste water from various uses of the reservoir waters will degrade the tributary waters entering the main stem. For any plan to be effective in the improvement of water quality of Red River throughout its reach in Texas, large amounts of natural brine must be prevented from entering water courses of the basin.

RECONNAISSANCE OF THE CHEMICAL QUALITY OF SURFACE WATERS OF THE RED RIVER BASIN, TEXAS

INTRODUCTION

The investigation of the chemical quality of the surface waters of the Red River basin, Texas, is part of a statewide reconnaissance. Each major river basin in the State is being studied and a report is being prepared to present the results of the study and to summarize the available chemical-quality data. Reports that have been published are included in the list of references.

The purpose of this report is to summarize information on the quality of surface water in the Red River basin, and to present it in a form that will aid in the proper development, control, and use of water resources of the area. In the study, the following items were considered: the nature and amounts of mineral constituents in solution; the geologic, hydrologic, and cultural influences that determine water quality; the amount and probable source of the salt transported by streams; and the suitability of the water for domestic, industrial, and agricultural uses. Data for the Oklahoma part of the Red River basin are included to show the effect of runoff from Oklahoma on the chemical quality of water in the mainstem Red River.

A network of daily chemical-quality stations on principal streams in Texas is operated by the U.S. Geological Survey in cooperation with the Texas Water Development Board and with federal and local agencies. This network has not been adequate to inventory completely the chemical quality of the surface waters of the State. To supplement the information being obtained by the network, a cooperative statewide reconnaissance by the U.S. Geological Survey and Texas Water Development Board was begun in September 1961. In this reconnaissance, samples for chemical analyses have been collected periodically at numerous sites throughout the State so that some quality-of-water information would be available for locations where water-development projects are likely. These data aid in the delineation of areas having water-quality problems and in the identification of probable sources of pollution, and thus indicate areas where more detailed investigations are needed.

During the period September 1961 to September 1967, water-quality data were collected on the principal streams, on the major reservoirs, at a number of potential reservoir sites, and on many tributaries in the basin.

Quality-of-water information for the Oklahoma part of the Red River basin was collected by the U.S. Geological Survey in cooperation with the Oklahoma Water Resources Board. Water-quality data in Texas and Oklahoma have been collected also by the U.S. Public Health Service and the Federal Water Quality Administration.

Agencies that have cooperated in the collection of water-quality and streamflow data include the U.S. Army Corps of Engineers, the Texas State Department of Health, and the city of Wichita Falls.

RED RIVER DRAINAGE BASIN

General Description

The Red River basin in Texas is bounded on the north by the Canadian River basin and on the south by the Brazos, Trinity, and Sulphur River basins. (See Figure 1).

The headwater stream in the Red River basin, Tierra Blanca Creek, rises in the High Plains of eastern New Mexico about 40 miles west of the Texas-New Mexico boundary at an elevation of about 4,800 feet above mean sea level. Tierra Blanca Creek flows eastward across the Texas High Plains and becomes the Prairie Dog Town Fork Red River in eastern Randall County. The Prairie Dog Town Fork Red River flows eastward to the southeast corner of the Texas Panhandle where it becomes the Red River. The Red River flows eastward as the Texas-Oklahoma boundary, then becomes the Texas-Arkansas boundary for about 30 miles before leaving the State. At the northeast corner of Texas, the streambed elevation is about 250 feet.

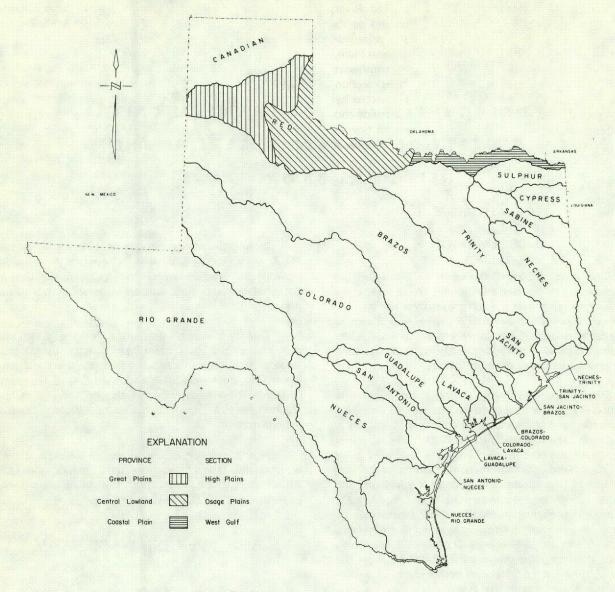


Figure 1.—Drainage Basins in Texas

The Red River has many tributaries in Oklahoma and Texas. The Washita River, Sweetwater Creek, and the North and Salt Forks Red River rise in the Texas Panhandle and flow into Oklahoma before joining the Red River from the north. The major all-Texas tributaries are the Pease, Wichita, and Little Wichita Rivers. Downstream from the Little Wichita River, the Texas part of the basin is narrow and is drained by numerous small streams. The major all-Oklahoma tributaries are Muddy Boggy Creek and the Kiamichi River.

The total area drained by the Red River usptream from the northeast corner of Texas is approximately 48,000 square miles, of which about 5,900 square miles is considered as noncontributing to streamflow. The total area in Texas draining to the Red River is approximately 24,500 square miles of which about 5,300 square miles is considered noncontributing.

The Red River basin in Texas is in three physiographic sections—the High Plains section of the Great Plains province, the Osage Plains section of the Central Lowlands province, and the West Gulf Coastal Plain section of the Coastal Plain province. The physiographic sections are shown on Figure 1.

The High Plains section within the Red River basin is characterized by a nearly flat surface sloping gently southeastward about 10 feet per mile. Among the few and generally insignificant features of relief are saucerlike depressions, ranging in diameter from several tens of feet to about 1 mile, and ranging in depth from a few inches to about 60 feet. The eastern margin of the High Plains is marked by a prominent escarpment or "break of the plains."

The Osage Plains section within the Red River basin adjoins the High Plains section and has as its eastern boundary the western margin of the gulfward-dipping Cretaceous rocks of the West Gulf Coastal Plain, which extends diagonally from northeast to southwest across Montague County. The Osage Plains section generally is a gentle eastward-sloping plain dissected by prominent systems of drainage. The valleys are wide and bounded by abrupt escarpments, and the streams flow in broad, shallow channels. Much of the surface area has a definite reddish color.

The West Gulf Coastal Plain section extends from the edge of the Osage Plains section eastward throughout the remainder of the report area. Low relief and a gentle gulfward slope of the land surface characterizes this section. Local topographic features are irregular, rolling, and hilly uplands, and flat flood plains and terraces. The streams have wide, nearly flat flood plains bounded by a series of terraces, which may be more than 100 feet higher than the stream channels.

The climate of the basin ranges from semiarid to humid (Thornthwaite, 1952, p. 32). Thornthwaite's classification, which is based on a moisture index, compares potential evapotranspiration with precipitation. Where precipitation is exactly the same as potential evapotranspiration and water is available just as needed, water is neither deficient nor in excess, and the climate is neither moist or dry. As water deficiency becomes larger with respect to potential evapotranspiration, the climate becomes more arid; conversely, as water surplus becomes larger, the climate becomes more humid.

East of a north-south line near the Cooke-Montague County line, the basin has surplus moisture and is characterized by a moist subhumid to humid climate. West of this line the area is deficient in moisture and has a dry subhumid to semiarid climate.

Precipitation ranges from an annual mean of less than 18 inches in the far western part of the basin to more than 46 inches in the extreme eastern part.

Figure 2 shows the average monthly precipitation at Amarillo, Wichita Falls, and Sherman, Texas, and McAlester, Oklahoma; it also shows the annual precipitation for 1937-65 at Altus, Oklahoma. In general, precipitation is greatest during the spring and summer months and least during the winter. However, precipitation is more evenly distributed throughout the year in the eastern part of the basin than in the western part.

Runoff is that part of precipitation that appears in surface streams. It is the same as streamflow unaffected by artificial diversions, storage, or other works of man in or on stream channels (Langbein and Iseri, 1960, p. 17). However, the terms are not synonymous for regulated flow. The Red River is regulated by Lake Texoma, and some of the tributary streams are regulated by reservoirs,

floodwater-retarding structures, and farm ponds. However, many streams in the Red River basin are not regulated by reservoirs of appreciable size.

The 28-year record of the Salt Fork Red River at Mangum, Oklahoma, is the only long-term record of flow from west of the 100th meridian. Runoff varies widely from year to year, and at the Mangum station has varied from a maximum of 200,400 acre-feet in 1941 to a minimum of 8,930 acre-feet in 1940. Annual average runoff is 0.9 inch (50 acre-feet per square mile) at this station. Runoff at Mangum is indicative of runoff from the area west of the 100th meridian.

Runoff in the Red River basin in Texas increases more or less uniformly from west to east, and averages more than 15 inches per year (800 acre-feet per square mile) at the northeast corner of the State. For the period 1944-65, the average runoff was 15.8 inches per year at the U.S. Geological Survey stream-gaging station Boggy Creek near Daingerfield in nearby Cypress Creek basin. Average annual runoff in inches per year, as computed from streamflow records for the period 1938-66, is given for seven stations on Figure 2. Also shown on Figure 2 is annual runoff expressed as mean discharge in cubic feet per second and inches per year for the gaging stations Salt Fork Red River at Mangum, Oklahoma; Red River near Gainesville, Texas; and Red River at Index, Arkansas.

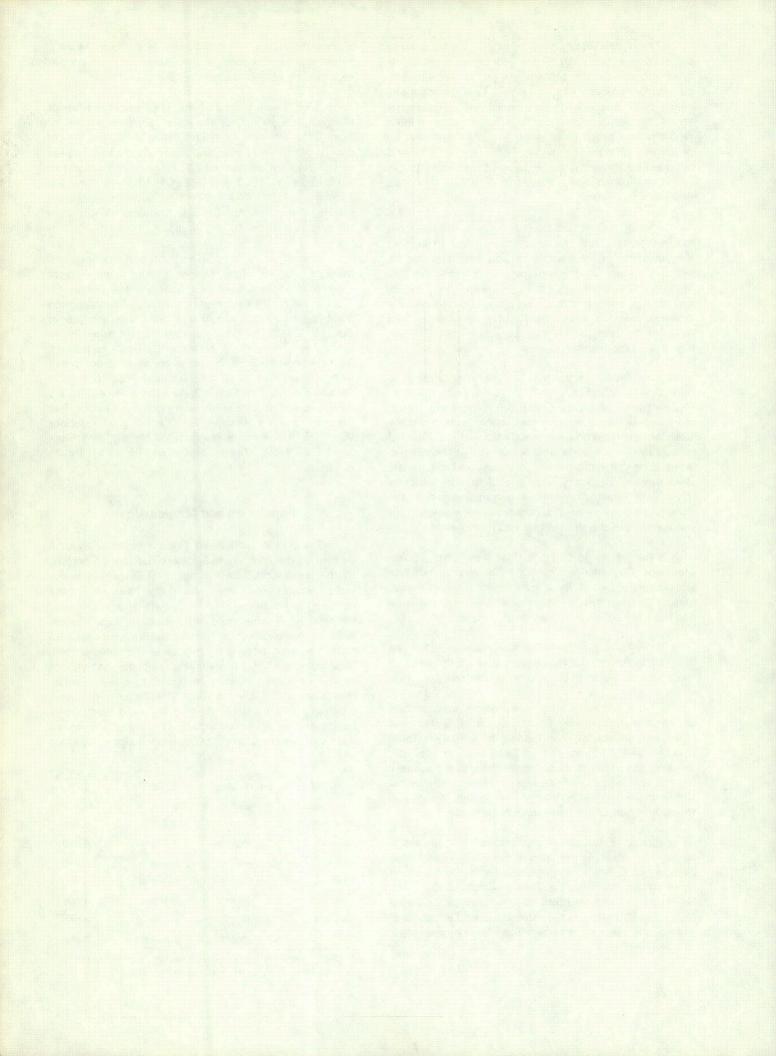
Population and Municipalities

The Red River basin in Texas constitutes about 9 percent of the area of the State and has about 4 percent of the population. Much of the land is sparsely populated. Population changes within the basin reflect the national trend of rural area decline and urban area increase. As in other areas of the country, these changes are due to the reduction of farm employment opportunities resulting from the development of mechanized, large-scale agricultural methods, and the consequent exodus of surplus farm labor to cities, as well as other migration and social factors. The larger cities have continued to grow while the population of small towns has remained fairly constant. The cities and towns having populations over 2,500 are listed in the following table.

CITY	*POPULA- TION	CITY	*POPULA- TION
Amarillo1/	164,770	Bonham	7,600
Wichita Falls	113,800	Tulia	6,690
Sherman	27,100	Childress	6,420
Paris	24,000	Iowa Park	5,410
Denison	23,400	Shamrock	3,420
Vernon	13,980	Nocona	3,360
Hereford	12,570	Henrietta	3,200
Burkburnett	8,490	Whitesboro	2,980

^{* 1967} population estimates (Dallas Morning News, 1967).

^{1/}Amarillo is partly in the Canadian River basin.



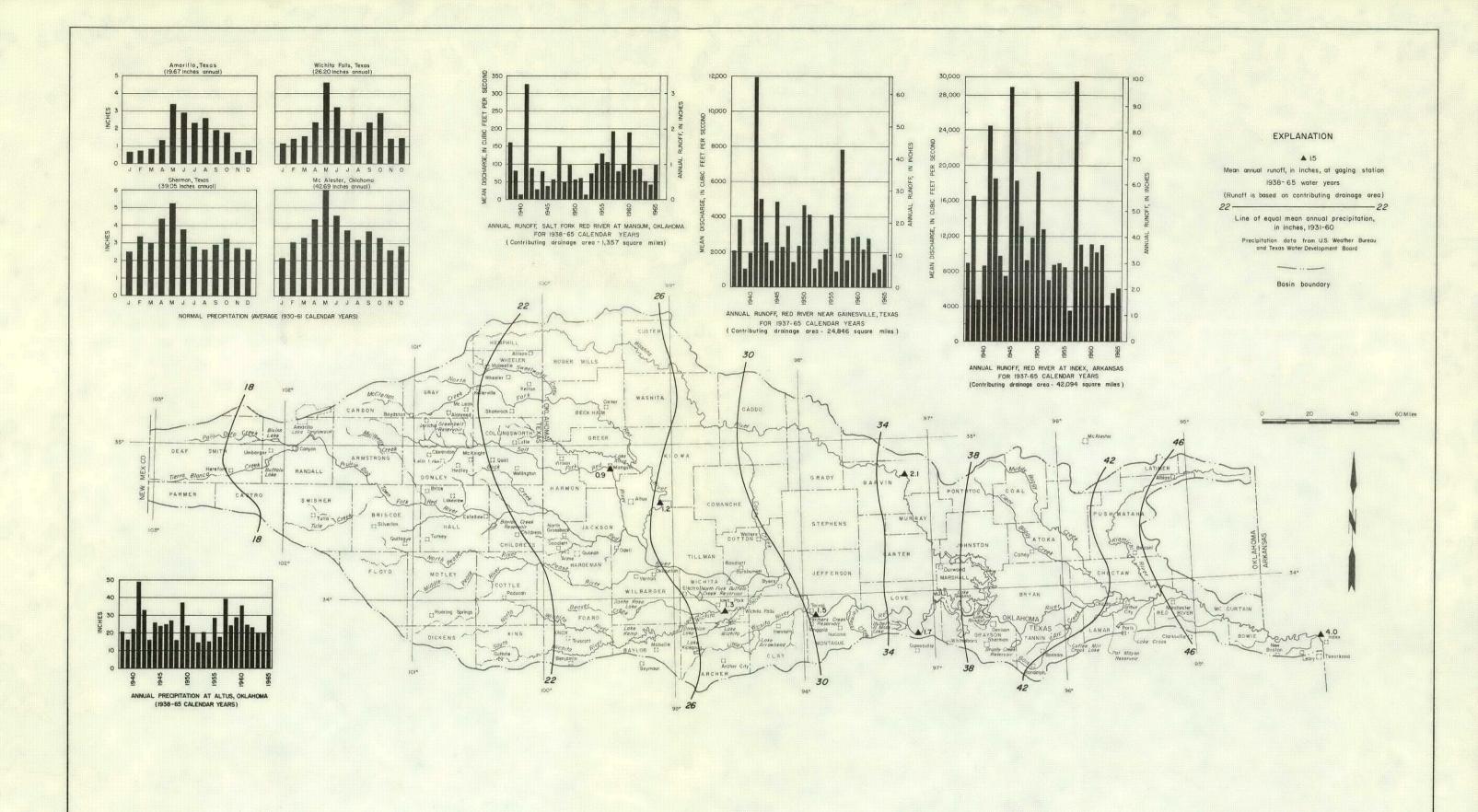
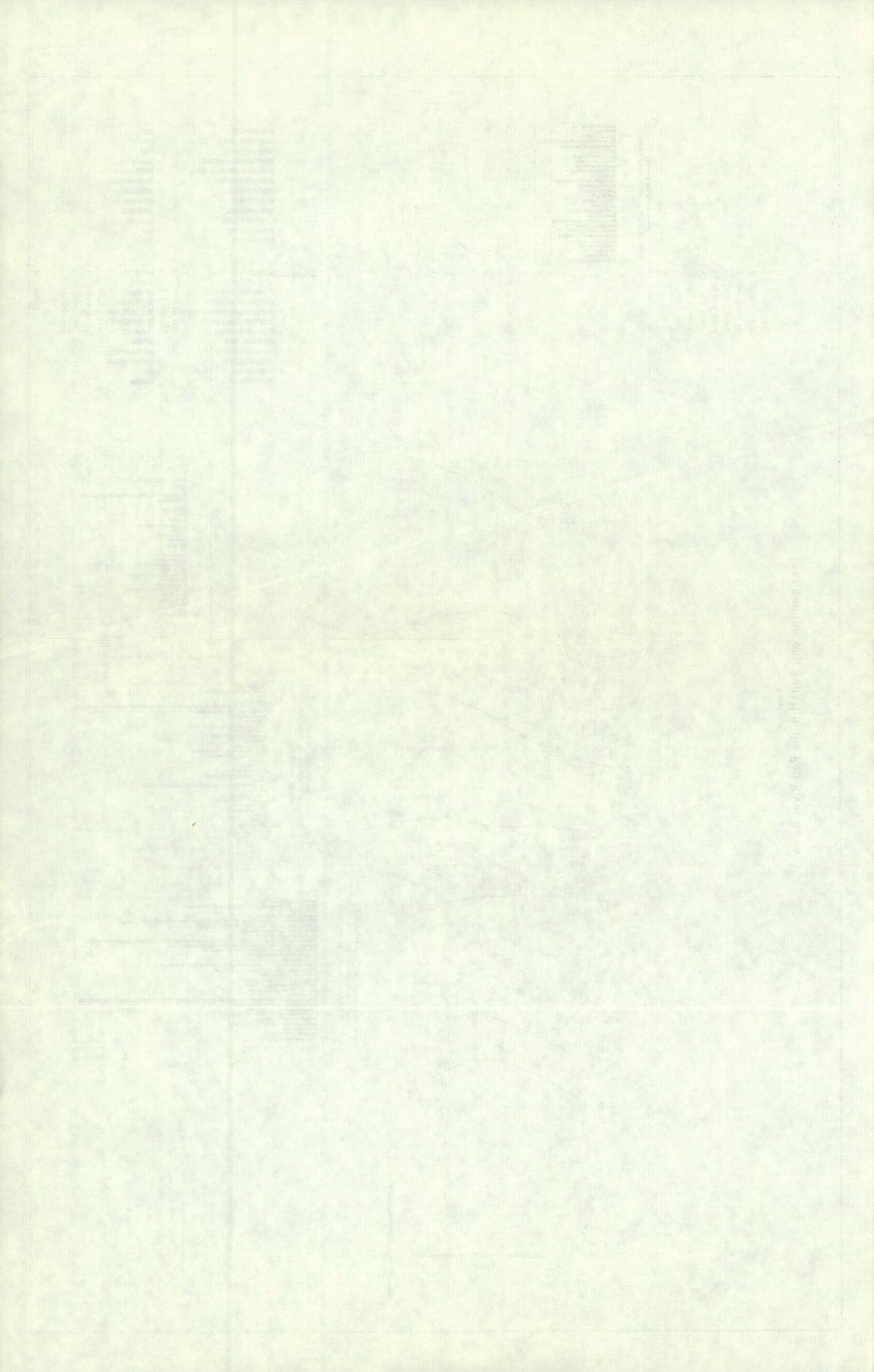


Figure 2
Precipitation and Runoff in the Red River Basin



Agricultural and Industrial Development

Agriculture has contributed substantially to the economic growth of the Red River basin. Farming, livestock raising, and dairying are successful because of the fertile soils and generally favorable climate. The availability of ground water for irrigation and the advent of mechanized farm equipment have been largely responsible for the success of farming in the drier western part of the basin. In the eastern part, where rainfall is greater, supplemental irrigation insures good crop yields. Cotton, grain sorghums, and wheat are the principal crops in the western part, and cotton, corn, and vegetables predominate in the eastern part.

The processing of local farm products is one of the major industries in the basin; processing plants are located close to areas of agricultural production. Oil and gas production constitutes another substantial income-producing segment of the economy. Much of the industrial development of the basin is related to the production of oil and gas. The development of irrigation in places has been greatly facilitated by the abundant supply of natural gas for power. Industries in the area that depend on the production of oil and gas include synthetic rubber, carbon black, oil refining, petrochemical, and pipeline equipment. Lumber mills, plants related to timber production, power plants, machinery, and furniture manufacturers are also located in the basin.

Development of Surface-Water Resources

The only reservoir on the Red River is Lake Texoma, which was built for flood control and hydroelectric power generation. Because of the poor quality of the water of the main stem, most of the water development projects in the basin are on tributary streams. As of December 31, 1967, nineteen reservoirs in the Texas part of the basin had capacities of 5,000 acre-feet or more. The capacity, ownership, and use of these reservoirs are listed in Table 1; the locations are shown on Figure 11.

CHEMICAL QUALITY OF THE WATER

Chemical-Quality Records

Although the U.S. Geological Survey has collected chemical-quality records in the Red River basin, Texas, since 1942, very few long-term daily records are available. In 1942, a daily sampling station was established on the Pease River near Crowell, but it was discontinued in 1943. Daily chemical-quality records of more than 10 years are available for the stations at Red River near Gainesville and Red River at Denison Dam. Since 1942, the U.S. Geological Survey has collected daily

chemical-quality data for varying periods at 12 stations either on the main stem or on Texas tributaries. In addition, miscellaneous chemical-quality data are available for numerous sites.

The periods of record at all data-collection sites in Texas are given in Table 4 and the locations are shown on Figure 11. The chemical-quality data for the daily stations are summarized in Table 5, and the complete records are published in an annual series of U.S. Geological Survey Water-Supply Papers and in reports of the Texas Water Development Board and predecessor agencies. Results of all the miscellaneous analyses are given in Table 6. Chemical analyses from selected stations in Oklahoma are given in Table 7. Complete records of all chemical-quality data available for surface water in Oklahoma are published in the annual series of U.S. Geological Survey Water-Supply Papers and in reports of the Oklahoma Water Resources Board. See list of references.

Chemical-quality records, including continuous specific conductance data, were collected by the U.S. Public Health Service (1964) at 27 sites in the Red River basin in Texas and Oklahoma during 1961-62. Public Health Service sampling sites in the Red River basin in Texas are identified in Table 4.

The Texas State Department of Health has made available to the Geological Survey the data collected in its former statewide stream-sampling program. The former data-collection sites are listed in the following table. Some of them are at U.S. Geological Survey stream-gaging stations. The numbers refer to sites shown on Figure 11.

SITE NO. FORMER TEXAS STATE DEPART-MENT OF HEALTH DATA-COLLECTION SITES

- 4 Palo Duro Creek at Park Road 5 near Canyon, Texas.
- 13 Prairie Dog Town Fork Red River at State Highway 70 near Brice, Texas.
- Prairie Dog Town Fork Red River at U.S. Highway 83 near Childress, Texas.
- 24 Red River at State Highway 283 near Quanah, Texas.
- Red River at U.S. Highway 183 near Oklaunion, Texas.
- 68 Red River at U.S. Highway 281 near Burkburnett, Texas.
- Red River at State Highway 79 near Byers, Texas.

SITE NO. FORMER TEXAS STATE DEPART-MENT OF HEALTH DATA-COLLECTION SITES

- 97 Red River at U.S. Highway 81 near Terral, Oklahoma.
- 99 Red River near Gainesville, Texas.
- Red River near Denison, Texas.
- Red River at State Highway 78 near Bonham, Texas.
- 104 Red River at U.S. Highway 271 near Arthur City, Texas.
- Red River at State Highway 37 at Albion, Texas.
- Red River at U.S. Highway 59 near Texarkana, Texas.

Streamflow Records

Streamflow records in the Red River basin date from the 1890's, when the U.S. Weather Bureau began collecting gage-height records on the Red River at Arthur City in 1891 and on the Red River near Colbert, Oklahoma (near Denison, Texas) in 1892. The first Geological Survey gaging station was established on the Wichita River at Wichita Falls in 1900. Discharge records are available for more than 50 stations on the Red River and its tributaries in Texas; 11 stations have 15 years or more of record and several others have more than 5 years of record.

In 1966 the Geological Survey operated 26 streamflow stations, five reservoir content stations, and five partial-record stations in the Red River basin, Texas. During this reconnaissance, discharge measurements were made at other sites where water samples were collected for chemical analyses.

Records of discharge, stage of streams, and contents and stages of reservoirs from 1900 to 1960 have been published in the annual series of the U.S. Geological Survey Water-Supply Papers. Beginning with the 1961 water year, streamflow records have been released by the Geological Survey in annual reports for each state (U.S. Geological Survey, 1961, 1962, 1963, 1964b, 1965, 1966). Summaries of discharge records giving monthly and annual totals have been published (U.S. Geological Survey 1955, 1964a; Texas Board of Water Engineers, 1958).

Environmental Factors and Their Effects on the Chemical Quality of the Water

Water from natural sources contains mineral constituents dissolved from the rocks and soils of the

earth's crust. The kind and quantities of dissolved minerals in surface water depend upon a number of environmental factors, some of the most important of which are geology, streamflow characteristics, and the activities of man.

Geology

The amounts and kinds of minerals dissolved in water that drains from areas where municipal and industrial influences are small depend principally on the chemical composition and physical structure of the rocks and soils traversed by the water. The length of time the water is in contact with the soil and rocks is also important. The amount of minerals in the soils and rocks available for solution is decreased by leaching; therefore, in areas of high rainfall, rocks that originally contained large quantities of readily soluble minerals have been leached by circulating water until the mantle rock and residual soil contain relatively small amounts of readily soluble materials. These rocks usually yield water of low mineralization. However, in arid or semiarid regions most soils, and the rocks from which they originated, are incompletely leached and still contain large amounts of readily soluble material.

In the semiarid western part of the Red River basin, some rocks and soils contain large quantities of halite, gypsum, limestone, and dolomite. Water of streams draining these areas usually is highly mineralized. In the eastern part of the basin, where precipitation is more abundant, the well-leached rocks usually yield waters of low mineralization.

The geology of the Red River basin, Texas, has been described by Baker and others (1963, p. 18-26). Rocks exposed in the Texas part of the basin consist of a thick series of sedimentary strata that range in age from Pennsylvanian to Quaternary. The outcrop areas of the geologic units are shown on Figure 3.

Chemical analyses of selected low-flow samples are represented diagrammatically (Stiff, 1951) on Figure 3 to relate chemical composition of surface waters to geology. The shape of the diagram indicates the relative concentrations of the principal chemical constituents of the water (in milliequivalents per liter) and the size of the diagram indicates roughly the relative degree of mineralization.

The headwater stream of the Red River rises in the Ogallala Formation of Tertiary age. The Ogallala consists of clay, silt, sand, gravel, and caliche. Some of the sand, gravel, and silt are unconsolidated; but some cementation occurs, chiefly by calcium carbonate. The principal chemical constituents in water from the Ogallala are sodium, calcium, magnesium, and bicarbonate. Base flow is generally nonexistent in streams that drain the Ogallala outcrop, and runoff occurs only after heavy rains.

Table 1.—Reservoirs in the Red River Basin in Texas Having Capacities of 5,000 Acre-Feet or More. 1

(The purpose for which the impounded waters are used is indicated by the following symbols: M, municipal; I, industrial; Ir, irrigation; P, hydroelectric power; F, flood control; R, recreation.)

	DATE		*CAPACITY			
RESERVOIR	COMPLETED	STREAM	(AC-FT)	OWNER	COUNTY	USE
Buffalo Lake	1938	Tierra Blanca Creek	18,150	Fish and Wildlife Service, U.S. Department of Interior	Randall	R
Bivins Lake	1927	Palo Duro Creek	5,120	City of Amarillo	Randall	М
Baylor Creek	1950	Baylor Creek	9,220	City of Childress	Childress	М
Greenbelt	1966	Salt Fork Red River	59,800	Greenbelt Municipal and Industrial Water Authority	Donley	М, І
Lake Kemp	1923	Wichita River	461,800	City of Wichita Falls and Wichita County Water Improvement District No. 2	Baylor	l, ir
Diversion Lake	1924	do	40,000	do	Baylor, Archer	I, Ir
Santa Rose Lake	1929	Beaver Creek	11,570	W. T. Waggoner Estate	Wilbarger	l, Ir
North Fork Buffalo Creek	1964	North Fork Buffalo Creek	15,400	Wichita County Water Control and Improvement District		
				No. 3	Wichita	M
Lake Wichita	1901	Holliday Creek	14,000	City of Wichita Falls	Wichita, Archer	M
Lake Kickapoo	1945	North Fork Little Wichita River	106,000	do	Archer	M
Lake Arrowhead	1966	Little Wichita River	228,000	do	Archer, Clay	М, І
Farmers Creek	1960	Farmers Creek	25,400	North Montague County Water Supply District	Montague	M, I
Hubert H. Moss Lake	1966	Fish Creek	23,200	City of Gainesville	Cooke	М, І
Lake Texoma	1943	Red River	5,393,000	U.S. Army Corps of Engineers	Cooke, Grayson	P, F
Lake Randall	1909	Shawnee Creek	5,400	City of Denison	Grayson	M
Brushy Creek	1961	Brushy Creek	16,800	Texas Power and Light Co.	Fannin, Grayson	P
Coffee Mill Creek Lake	1938	Coffee Mill Creek	8,000	U.S. Forest Service	Fannin	R
Pat Mayse	1967	Sanders Creek	124,500	U.S. Army Corps of Engineers	Lamar	M, I, F
Lake Crook	1923	Pine Creek	9,960	City of Paris	Lamar	М

^{1/} Existing or under construction as of December 31, 1967.

^{*} Total capacity is that capacity below the lowest uncontrolled outlet or spillway and is based on the most recent reservoir survey available.

Downstream from the Ogallala outcrop, the drainage area of the Prairie Dog Town Fork Red River is underlain by rocks of Triassic and Permian age. The Dockum Group of Triassic age consists of shale and sandy shale, crossbedded sandstone, and conglomerate. The chemical quality of the water from the Dockum Group varies with local conditions, but it is generally unsuitable for irrigation or public supply.

Rocks of Permian age crop out over much of the basin east of the High Plains Escarpment and west of the eastern boundary of Montague County. The Permian rocks consist predominantly of shale, anhydrite, gypsum, limestonė, dolomite, and sandstone. The chemical composition of water contributed to streams by these rocks varies. During periods of sustained low flow, water of the Prairie Dog Town Fork Red River, and the North and Salt Forks Red River is of a highly mineralized calcium sulfate type. Water of the Pease River and North and South Forks Wichita River is of a highly mineralized sodium chloride type. Significant natural brine emission areas as identified by the U.S. Public Health Service (1964) are shown on Figure 4. Numerous small alluvial deposits of Quaternary age are present in this area, and ground-water flow from them probably causes some of the variations in chemical composition and dissolved-solids concentration of surface waters.

Downstream from Lake Kemp, the Wichita River drains rocks of the Wichita Group of Permian age. The Wichita Group consists of shale, sandstone, and limestone. Ground-water effluent in this reach is generally of a mixed chemical type having sodium, sulfate, and chloride as the predominant ions. Dissolved-solids concentrations are much less than those of waters above Lake Kemp.

The drainage area of the Little Wichita River is underlain by rocks of the Cisco and Wichita Groups. The Cisco Group of Pennsylvanian age is composed of shale, limestone, sandstone, and conglomerate. Ground-water effluent in Little Wichita River watershed is generally of a mixed chemical type, having sodium, bicarbonate, and chloride as the predominant ions; the water is relatively low in dissolved solids. However, oil-field brine pollution in some reaches in the Wichita and Little Wichita Rivers has in the past altered the composition of water to a sodium chloride type.

Downstream from the Montague-Cooke County line, the streams drain rocks of Cretaceous age. The principal outcrops are the Eagle Ford Shale consisting of shale, limestone, and sand; rocks of Austin age consisting of chalk, marl, and sand; rocks of Taylor age consisting of marl, chalk, and sandy marl; and the Washita and Fredericksburg Groups undifferentiated consisting of limestone, marl, and clay. Waters draining these rocks, although varying slightly in composition from one formation to another, are low in dissolved-solids content

and usually contain calcium and bicarbonate as the predominant ions. Quaternary alluvium is exposed from Lake Texoma eastward to the northeast corner of Texas. However, these alluvial deposits are present along the river in the form of terraces which hold water in bank storage. The alluvium is recharged during high flow, but it releases the water to the river when the high flow subsides.

Streamflow

For many streams not regulated by upstream reservoirs, the concentrations of dissolved minerals vary inversely with the water discharge. The minimum concentrations usually occur during periods of high flow because most of the water is surface runoff that has been in contact with rocks and soils for a relatively short time. The maximum concentrations usually occur during periods of low flow when the water is predominantly ground water that has been in contact with the rocks and soils for a sufficient time to dissolve part of their soluble minerals.

In the western part of the Red River basin, dissolved-solids content and water discharge are not related in a predictable manner. Many of the streams are dry or almost dry much of the time, and salt deposits accumulate on the beds and banks of the streams. Subsequent runoff dissolves these deposits causing erratic variation in the salt content of the runoff.

In the eastern part of the basin, the dissolvedsolids content of ground-water effluent is generally low, and therefore the dissolved-solids content of streams varies only slightly with discharge. Consequently, the dissolved solids-water discharge relationship is poorly defined for streams in the Red River basin in Texas.

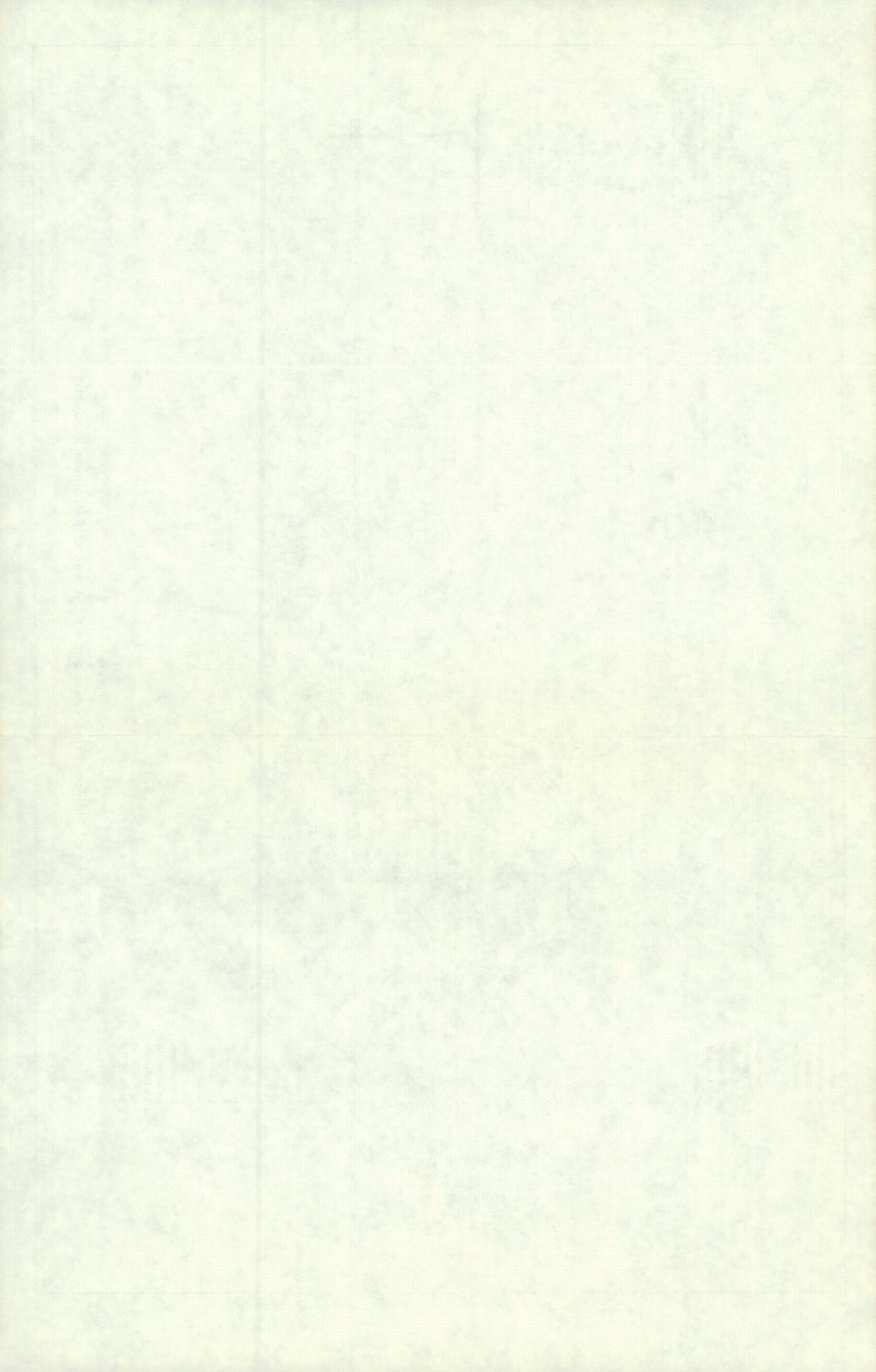
Activities of Man

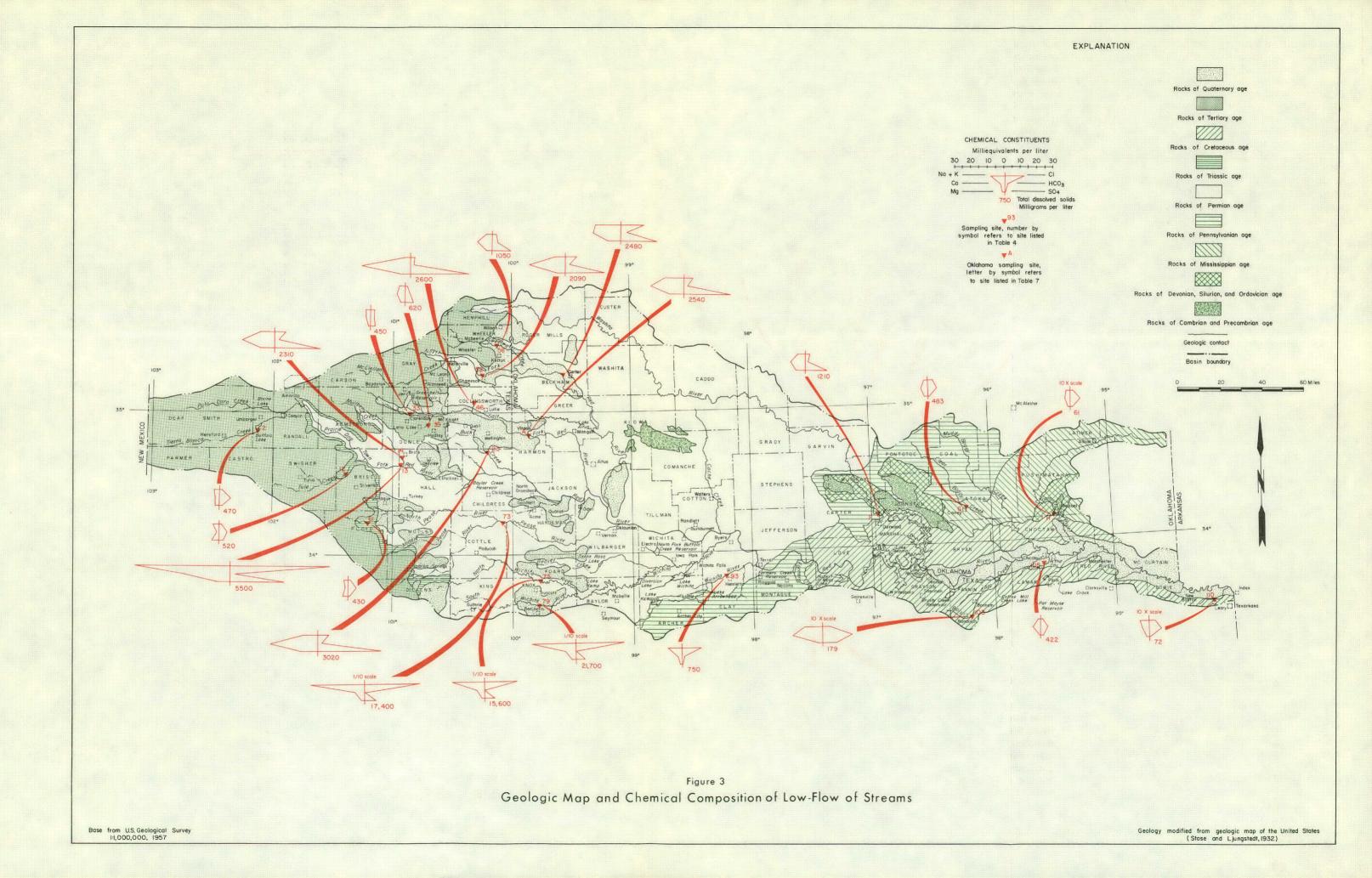
The activities of man often degrade the chemical quality of surface water. Depletion of flow by diversion and by consumptive use, increased evaporation from impoundment, and return flow from irrigation increase the dissolved-solids concentration of water in streams. Also, the discharge of municipal and industrial wastes into a stream degrades the chemical quality of water.

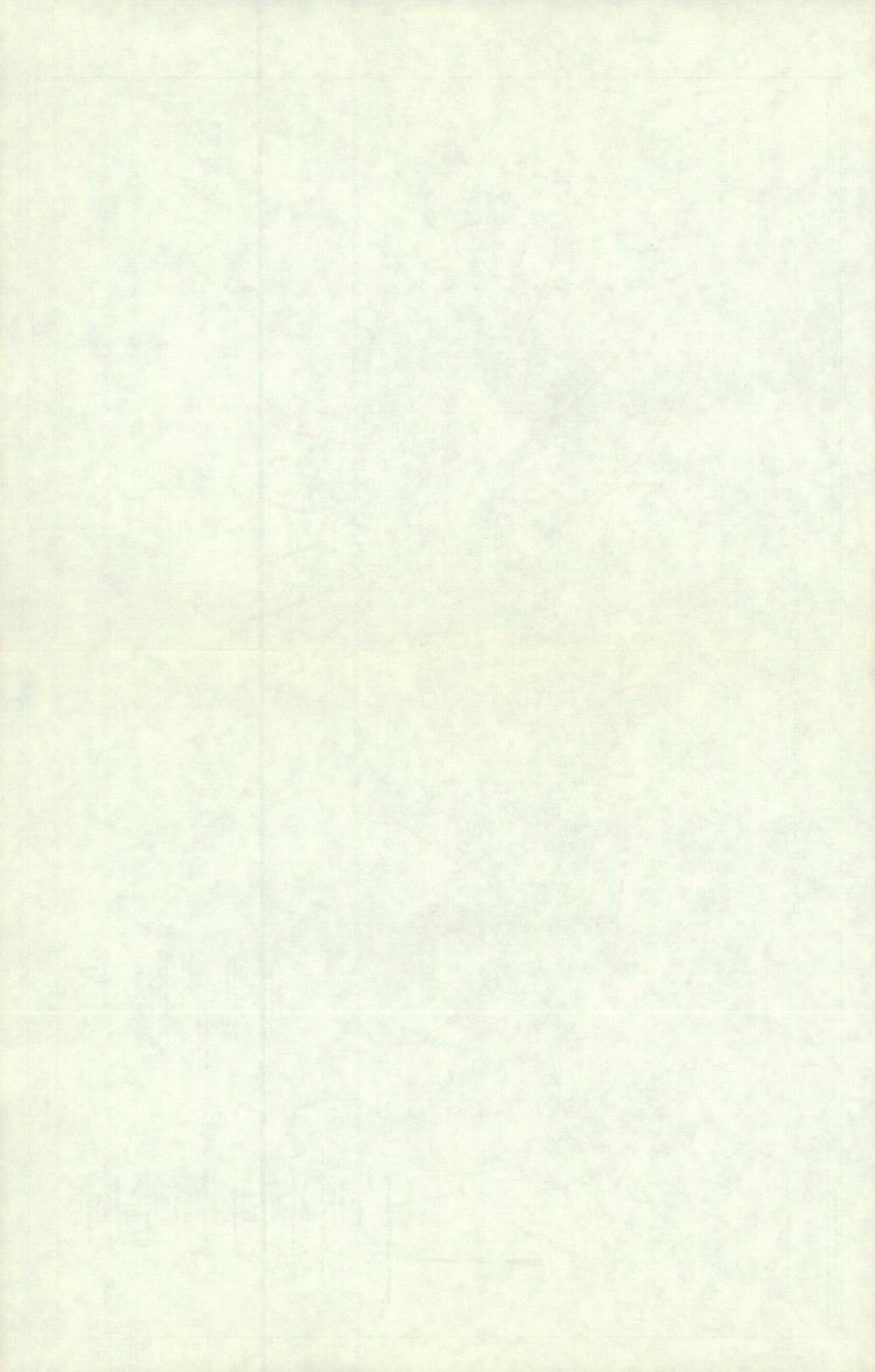
Eighteen reservoirs presently impound water of Texas tributaries to the Red River (Table 1). All of these except Lake Kemp, Diversion Lake, and Lake Wichita contain waters of good to excellent quality. Because these waters are stored and prevented from reaching the river, the quality of water of the main stem has been degraded to some extent. Much of the lake water eventually returns to the river system, but its quality has been affected by evaporation and other hydrologic changes due to impoundment. Most often the water

EXPLANATION Contribution of significant natural brine emission areas General area of petroleum and Load-tons/day Estimated natural gas production brine flow Area No. CI NaCI Cfs Area of intensive petroleum production 175 Significant natural brine emission area 150 250 375 620 Secondary natural brine emission area 450 740 150 250 250 410 350 580 200 330 50 80 10 200 330 WISHER CHILDRESS Goodlett

Figure 4
Location of Natural Brine Emissions and Areas of Petroleum Production







diverted from the lakes returns to the river system as waste water from municipal, industrial, and agricultural uses. As the needs for more water and number of reservoirs increase, the quality-of-water problems of the main stem may increase.

At present, degradation of water quality by return flow from irrigation in the Red River basin, Texas is considered minor and localized. Although 1.4 million acres in the basin was irrigated with over 2 million acre-feet of water in 1964, more than 95 percent of both the land irrigated and the water used was in the Texas Panhandle (Gillett and Janca, 1965). Nearly all of the Panhandle supply is from ground water; the Ogallala Formation supplied about 1.5 million acre-feet of water to irrigate approximately a million acres in the High Plains. Irrigation return flow contributes very little to streamflow in the Panhandle, and any flow that does reach a stream probably will enhance rather than degrade the quality of the natural saline waters in most areas. The use of surface water for irrigation is limited primarily to the Wichita Falls area and the area along the Red River north of Texarkana. Minor, localized degradation of small tributaries to the Red River may be occuring in these areas.

Oil is produced in many areas in the Red River basin (Figure 4). Brine is produced in nearly all oil fields and, if improperly handled, eventually reaches surface streams. According to an inventory by the Texas Railroad Commission in 1961, more than 95 percent of the salt water produced in oil fields of the Red River basin, Texas, was injected underground (Texas Water Commission and Texas Water Pollution Control Board, 1963). The remainder of the salt water was disposed of in open surface pits, most of which were unlined. From these surface pits, much of the brine has seeped into the ground and eventually reaches the streams, or it is washed by surface runoff directly into the streams. Also, brine from abandoned wells and unplugged or improperly plugged test holes may reach streams.

The composition of oil-field brine varies, but the principal chemical constituents, in order of magnitude of their concentrations, are chloride, sodium, calcium, and sulfate. Generally, an erratic variation of the sodium chloride content of water in streams draining areas where oil fields are located is evidence that oil-field brine pollution is occurring. Because of widespread contamination of streams in the Red River basin by naturally occurring sodium chloride brines, distinction between natural contamination and manmade pollution is difficult. However, the saline waters of several streams in the central part of the basin have contained salts from both natural sources and oil fields. Chemical-quality records indicate that several streams, not affected by naturally occurring brines, have periodically shown effects of oil-field drainage. Generally, these streams are in the Beaver Creek, Buffalo Creek, and Little Wichita River sub-basins.

Relation of Quality of Water to Use

Quality-of-water studies usually are concerned with determining the suitability of water—judged by the chemical, physical, and biological characteristics—for a proposed use. In the Red River basin, surface water is used for municipal and industrial supplies and for irrigation. This report considers only the chemical character of the water and its relation to these principal uses.

Most of the mineral matter dissolved in water is dissociated into charged particles, or ions. Principal cations (positive charged) in natural water are calcium (Ca), magnesium (Mg), sodium (Na), potassium (K), and iron (Fe). The principal anions (negative charged) are carbonate (CO₃), bicarbonate (HCO₃), sulfate (SO₄), chloride (Cl), fluoride (F), and nitrate (NO₃). Other constituents and properties are often determined to help define the chemical and physical quality of water. Table 2 lists the constituents and properties commonly determined by the U.S. Geological Survey, and includes a résume of their source and significance.

Domestic Use

Because of differences in individuals, varying amounts of water consumed, and other factors, it is difficult to define the safe limits for the mineral constituents usually found in water. The limits usually accepted in the United States for drinking water are the drinking-water standards established by the United States Public Health Service. Originally established in 1914 to control the quality of water used on interstate carriers for drinking and culinary purposes, these standards have been revised several times. The latest revision was in 1962 (U.S. Public Health Service, 1962). These standards have been accepted by the American Water Works Association and by many state departments of public health as minimum standards for all public water supplies.

The maximum concentrations permitted by the standards are given for selected constituents in the following table:

CONSTITUENTS	MAXIMUM CONCEN- TRATION (MILLI- GRAMS PER LITER)
Sulfate	250
Chloride	250
Nitrate	45
Fluoride	<u>a</u> /0.9
Dissolved solids	500

a/Recommended limits based on the average of maximum daily air temperatures. Concentration cited is the optimum based on temperature records for lowa Park.

CONSTITUENT OR PROPERTY	SOURCE OR CAUSE	SIGNIFICANCE
Silica (SiO ₂)	Dissolved from practically all rocks and soils, commonly less than 30 mg/l. High concentrations, as much as 100 mg/l, generally occur in highly alkaline waters.	Forms hard scale in pipes and boilers. Carried over in steam of high pressure boilers to form deposits on blades of turbines. Inhibits deterioration of zeolite-type water softeners.
Iron (Fe)	Dissolved from practically all rocks and soils. May also be derived from iron pipes, pumps, and other equipment. More than 1 or 2 mg/l of iron in surface waters generally indicates acid wastes from mine drainage or other sources.	On exposure to air, iron in ground water oxidizes to reddish- brown precipitate. More than about 0.3 mg/l stains laundry and utensils reddish-brown. Objectionable for food processing, tex- tile processing, beverages, ice manufacture, brewing, and other processes. U.S. Public Health Service (1962) drinking-water standards state that iron should not exceed 0.3 mg/l. Larger quantities cause unpleasant taste and favor growth of iron bacteria.
Calcium (Ca) and magnesium (Mg)	Dissolved from practically all soils and rocks, but especially from limestone, dolomite, and gypsum. Calcium and magnesium are found in large quantities in some brines. Magnesium is present in large quantities in sea water.	Cause most of the hardness and scale-forming properties of water; soap consuming (see hardness). Waters low in calcium and magnesium desired in electroplating, tanning, dyeing, and in textile manufacturing.
Sodium (Na) and potassium (K)	Dissolved from practically all rocks and soils. Found also in ancient brines, sea water, industrial brines, and sewage.	Large amounts, in combination with chloride, give a salty taste. Moderate quantities have little effect on the usefulness of water for most purposes. Sodium salts may cause foaming in steam boilers and a high sodium content may limit the use of water for irrigation.
Bicarbonate (HCO3) and carbonate (CO3)	Action of carbon dioxide in water on carbonate rocks such as lime- stone and dolomite.	Bicarbonate and carbonate produce alkalinity. Bicarbonates of calcium and magnesium decompose in steam boilers and hot water facilities to form scale and release corrosive carbon dioxide gas. In combination with calcium and magnesium, cause carbonate hardness.
Sulfate (SO ₄)	Dissolved from rocks and soils containing gypsum, iron sulfides, and other sulfur compounds. Commonly present in mine waters and in some industrial wastes.	Sulfate in water containing calcium forms hard scale in steam boilers. In large amounts, sulfate in combination with other ions gives bitter taste to water. Some calcium sulfate is considered beneficial in the brewing process. U.S. Public Health Service (1962) drinking-water standards recommend that the sulfate content should not exceed 250 mg/l.
Chloride (CI)	Dissolved from rocks and soils. Present in sewage and found in large amounts in ancient brines, sea water, and industrial brines.	In large amounts in combination with sodium, gives salty taste to drinking water. In large quantities, increases the corrosiveness of water. U.S. Public Health Service (1962) drinking-water standards recommend that the chloride content should not exceed 250 mg/l.
Fluoride (F)	Dissolved in small to minute quantities from most rocks and soils. Added to many waters by fluoridation of municipal supplies.	Fluoride in drinking water reduces the incidence of tooth decay when the water is consumed during the period of enamel calcification. However, it may cause mottling of the teeth, depending on the concentration of fluoride, the age of the child, amount of drinking water consumed, and susceptbility of the individual. (Maier, 1950)
Nitrate (NO ₃)	Decaying organic matter, sewage, fertilizers, and nitrates in soil.	Concentration much greater than the local average may suggest pollution. U.S. Public Health Service (1962) drinking-water standards suggest a limit of 45 mg/l. Waters of high nitrate content have been reported to be the cause of methemoglobinemia (an often fatal disease in infants) and therefore should not be used in infant feeding. Nitrate has been shown to be helpful in reducing inter-crystalline cracking of boiler steel. It encourages growth of algae and other organisms which produce undesirable tastes and odors.
Dissolved solids	Chiefly mineral constituents dis- solved from rocks and soils. Includes some water of crystalli- zation.	U.S. Public Health Service (1962) drinking-water standards recommend that waters containing more than 500 mg/l dissolved solids not be used if other less mineralized supplies are available. Waters containing more than 1000 mg/l dissolved solids are unsuitable for many purposes.
Hardness as CaCO ₃	In most waters nearly all the hardness is due to calcium and magnesium. All the metallic cations other than the alkali metals also cause hardness.	Consumes soap before a lather will form. Deposits soap curd on bathtubs. Hard water forms scale in boilers, water heaters, and pipes. Hardness equivalent to the bicarbonate and carbonate is called carbonate hardness. Any hardness in excess of this is called non-carbonate hardness. Waters of hardness as much as 60 ppm are considered soft; 61 to 120 mg/l, moderately hard; 121 to 180 mg/l, hard; more than 180 mg/l, very hard.
Specific conductance (micromhos at 25 ^o C)	Mineral content of the water,	Indicates degree of mineralization. Specific conductance is a measure of the capacity of the water to conduct an electric current. Varies with concentration and degree of ionization of the constituents.
Hydrogen ion concentration (pH)	Acids, acid-generating salts, and free carbon dioxide lower the pH. Carbonates, bicarbonates, hydroxides, and phosphates, silicates, and borates raise the pH.	A pH of 7.0 indicates neutrality of a solution, Values higher than 7.0 denote increasing alkalinity; values lower than 7.0 indicate increasing acidity, pH is a measure of the activity of the hydrogen ions. Corrosiveness of water generally increases with decreasing pH. However, excessively alkaline waters may also attack metals.

Industrial Use

The quality requirements vary greatly for almost every industrial application, as is indicated by the water-quality tolerances given in Table 3. One requirement of most industries is that concentrations of the various constituents of the water remain relatively constant. When concentrations of undesirable substances in water vary, constant monitoring is required and operating expenses are increased.

Hardness is one of the more important properties of water that affects its utility for industrial purposes. Excessive hardness is objectionable because it contributes to the formation of scale in steam boilers, pipes, water-heaters, radiators, and various other equipment where water is heated, evaporated, or treated with alkaline materials. The accumulation of scale increases costs for fuel, labor, repairs and replacement, and lowers the quality of many wet-processed products. However, some calcium hardness may be desirable because calcium carbonate sometimes forms protective coatings on pipes and other equipment and reduces corrosion.

The corrosive property of water receives considerable attention in industrial water supplies. A high concentration of dissolved solids in a water may be closely associated with the corrosiveness, particularly if chloride is present in appreciable quantities. Water that contains a large concentration of magnesium chloride may be highly corrosive because the hydrolysis of this salt yields hydrochloric acid.

Irrigation

The chemical composition of a water is an important factor in determining its usefulness for irrigation, because the quality of the water should not adversely affect the productivity of the land irrigated. The extent to which chemical quality limits the suitability of a water for irrigation depends on many factors, such as the nature, composition, and drainage of the soil and subsoil; the amounts of water used and the methods of applying it; the kind of crops grown; and the climate of the region. Because these factors are highly variable, every method of classifying waters for irrigation is somewhat arbitrary.

The most important characteristics in determining the quality of irrigation water, according to the U.S. Salinity Laboratory Staff (1954, p. 69) are: (1) total concentration of soluble salts, (2) relative proportion of sodium to other cations, (3) concentration of boron or other elements that may be toxic, and (4) the excess of equivalents of bicarbonate over equivalents of calcium plus magnesium.

High concentrations of dissolved salts in irrigation water may cause a buildup of salts in the soil solution,

and may make the soil saline. The increased salinity of the soil may reduce crop yields by decreasing the ability of the plants to take up water and essential plant nutrients from the soil solution. The tendency of irrigation water to cause a buildup of salts in the soil is called the salinity hazard of the water. The specific conductance of the water is used as an index of the salinity hazard.

High concentrations of sodium (Na) relative to the concentrations of calcium (Ca) and magnesium (Mg) in irrigation water can adversely affect soil structure. Cations in the soil solution become fixed on the surface of the soil particles; calcium and magnesium tend to flocculate the particles, whereas sodium tends to deflocculate them. This adverse effect on soil structure caused by high sodium concentrations in an irrigation water is called the sodium hazard of the water. An index used for predicting the sodium hazard is the sodium-adsorption ratio (SAR), which is defined by the equation:

$$SAR = \frac{Na^{+}}{\sqrt{Ca^{++} + Mg^{++}}}$$

where the concentrations of the ions are expressed in milliequivalents per liter.

The U.S. Salinity Laboratory Staff has prepared a classification for irrigation waters in terms of salinity and sodium hazard. Empirical equations were used in developing a diagram reproduced in modified form as Figure 5, which uses SAR and specific conductance in classifying irrigation waters. This classification, although embodying both research and field observations, should be used only as a general guide because many additional factors also affect the suitability of water for irrigation. With respect to salinity and sodium hazards, waters are divided into four classes; low, medium, high, and very high. The classification range encompasses waters that can be used for irrigation of most crops on most soils as well as waters that are usually unsuitable for irrigation. The salinity and sodium hazards of water at selected sites in the Red River basin are given on Figure 5.

Geographic Variations in Water Quality

Variations in dissolved solids, hardness, and chloride in the Red River basin are shown in Figures 12, 13, and 14. These values are based on the discharge-weighted average concentrations, as calculated from chemical-quality data. The discharge-weighted average represents the chemical character of the water if all the water passing a point in the stream during a period were impounded in a reservoir and mixed, with no adjustments for rainfall, evaporation, or chemical changes that might occur during storage. For many of the streams

Table 3.-Water-Quality Tolerances for Industrial Applications 1

[Allowable Limits in Milligrams Per Liter Except as Indicated]

INDUSTRY	TUR- BID- ITY	COLOR	COLOR +02 CON- SUMED	DIS- SOLVED OXYGEN (m1/1)	ODOR	HARD - NESS	ALKA- LINITY (AS CaCO ₃)		TOTAL SOLIDS	Ca	Fe	Mn	Fe+ Mn	A1203	S10 ₂	Cu	F	co3	нс03	ОН	CaSO ₄	Na2S04 TO Na2S03 RATIO	GEN- ERAL ² /
Air Conditioning3											0.5	0.5	0.5										A,B.
Baking	10	10		H-		(4)					. 2	.2	.2	77							7-1		ć
Boiler feed:																							
0-150 psi	20	80	100	2		75		8.0+	3,000- 1,000				nin	5	40			200	50	50		1 to 1	1
150-250 psi	10	40	50	.2		40		8.5+	2,500-	2-				.5	20		•••	100	30	40		2 to 1	
250 psi and up	5	5	10	0		8	15 () () ()	9.0+	1,500-			1001		. 05	5			40	5	30	60°44	3 to 1	()
Brewing: 5																							
Light	10				Low		75	6.5-7.0	500	100-200	. 1	. 1	. 1				1				100-200		C,D
Dark	10				Low		150	7.0→	1,000	200-500	.1	.1	.1				1				200-500		C,D
Canning:																							
Legumes	10				Low	25 - 75			122		. 2	.2	. 2										C
General	10				Low				15		. 2	. 2	. 2				1						С
Carbonated bev-																							
erages ⁶	2	10	10		0	250	50		850		.2	. 2	.3				. 2						С
Confectionary			••		Low			(7)	100		. 2	. 2	. 2	345 - 									
Cooling 8	50			7.7	••	50					.5	.5	.5										A,B
· Food, general	10		••		Low			••		- 1	. 2	.2	.2	•									С
Ice (raw water) 9/	1-5	5					30-50		300		. 2	. 2	.2		10								C
Laundering						50					.2	. 2	. 2										
Plastics, clear,																							
undercolored	2	2							200		.02	.02	.02							••			••
Paper and pulp: 19																							
Groundwood	50	20			••	180			-	-	1.0	.5	1.0										A
Kraft pulp	25	15				100			300		. 2	.1	.2									[
Soda and sulfite	15	10	••			100			200	44	.1	. 05	.1	••	••				-				
Light paper, HL-Grade	5	5				50			200		.1	. 05	.1			4-				-4		3	В
Rayon (viscose)																							
pulp: Production	5	5				8	50		100		. 05	. 03	. 05	<8.0	<25	<5							
	.3					55		7.8-8.3		10 22	.0	.0	.0										
Manufacture Tanning 11	20	10-100	24		,	50-135	135	8.0			. 2	. 2	.2										
Tanning '																							
Tanning 'j Textiles:																							
Textiles:	5	20				20					.25	.25			442						9		
Textiles: General Dueing 12	5	20 5-20	-			20 20		==		1	.25	. 25	.25	T.			::		=				
Textiles:							==	Ξ	==					Ξ	=		::		==	:: ::	==		==

^{1/} American Water Works Association, 1950.

^{2/} A-No corrosiveness; B-No slime formation; C-Conformance to Federal drinking water standards necessary; D-NaCl, 275 mg/l.

Waters with algae and hydrogen sulfide odors are most unsuitable for air conditioning.

⁴ Some hardness desirable.

^{3/} Water for distilling must meet the same general requirements as for brewing (gin and spirits mashing water of light-beer quality; whiskey mashing water of dark-beer quality).

⁽Clear, odorless, sterile water for syrup and carbonization. Water consistent in character. Most high quality filtered municipal water not satisfactory for beverages.

That candy requires pH of 7.0 or greater, as low value favors inversion of sucrose, causing sticky product.

^{8/} Control of corrosiveness is necessary as is also control of organisms, such as sulfur and iron bacteria, which tend to form slimes.

⁹ Ca (HCO₃)₂ particularly troublesome. Mg(HCO₃)₂ tends to greenish color. CO₂ assists to prevent cracking. Sulfates and chlorides of Ca, Mg, Na should each be less than 300 mg/1

¹⁰ Uniformity of composition and temperature desirable. Iron objectionable as cellulose adsorbs iron from dilute solutions. Manganese very objectionable, clogs pipelines and is oxidized to permanganates by chlorine, causing reddish color.

¹⁾ Excessive iron, manganese, or turbidity creates spots and discoloration in tanning of hides and leather goods.

^{12/} Constant composition; residual alumina 0.5 mg/l.

^{13/} Calcium, magnesium, iron, manganese, suspended matter, and soluable organic matter may be objectionable.

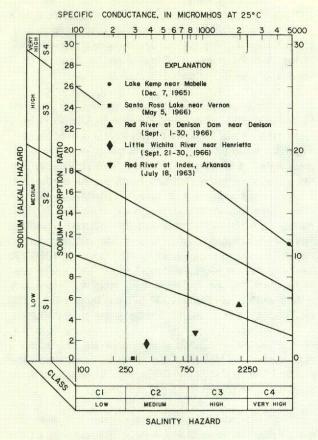


Figure 5.—Diagram for Classification of Irrigation Waters

chemical-quality data are limited, especially data on the chemical quality of flood flows; therefore, the sub-divisions shown on the maps should be considered as generalized. All the streams will at times have concentrations exceeding those shown, but the averages are indicative of the quality of water that would be stored in a hypothetical reservoir.

Dissolved Solids

The concentrations of dissolved solids in streams in the Red River basin range from several thousand to less than 250 mg/l (milligrams per liter) (Figure 3). Water from the outcrop areas of Tertiary age in the extreme western part of the basin usually have dissolved-solids concentrations less than 250 mg/l. Downstream from the Tertiary outcrop, rocks of Triassic and Permian age contribute water containing very high concentrations of dissolved solids; more than 10,000 mg/l is common in some areas. The highly concentrated water from the Prairie Dog Town, Salt, and North Forks Red River, Pease River, and North and South Wichita Rivers cause the mainstem Red River to contain more than 1,000 mg/l of dissolved solids throughout most of its reach in Texas. About midway between Lake Texoma and Index, Arkansas, good quality inflow from the tributaries is of sufficient quantity to cause the Red River to contain less than 1,000 mg/l of dissolved solids.

The discharge-weighted average concentrations of dissolved solids of the Red River near Gainesville for the periods 1944-46, 1953-63, and 1966-67 has ranged from a minimum of 891 mg/l in 1945 to a maximum of 1,950 mg/l in 1958. The discharge-weighted average concentration of dissolved solids of the Red River at Denison Dam for the period 1944-1967 has ranged from 486 mg/l in 1946 to 1,230 mg/l in 1961. The discharge-weighted average concentrations of dissolved solids at Index, Arkansas for 1961, 1962, and 1963 were 728 mg/l, 609 mg/l, and 538 mg/l, respectively. The analyses showing annual maximum and minimum dissolved-solids concentrations and the weighted averages for the stations are given in Table 5. Annual dissolved-solids averages for Red River at Denison Dam are shown on Figure 7.

Time-weighted averages represented by duration curves are usually higher than discharge-weighted averages. The duration curve for dissolved-solids concentrations for the Red River near Gainesville during 1953-62 is shown in Figure 6. Dissolved solids equaled or exceeded 3,560 mg/l 10 percent of the time, 3,040 mg/l 30 percent of the time, 2,620 mg/l 50 percent of the time, and 1,100 mg/l 90 percent of the time.

Downstream from the Wichita River, Texas tributaries drain Cretaceous rocks and contribute water containing less than 250 mg/l of dissolved solids. The discharge-weighted average concentration of dissolved solids of the Little Wichita River near Henrietta for the periods 1953-55 and 1959-66 ranged from 124 to 286 mg/l and averaged 211 mg/l; and the Little Wichita River near Ringgold for the 1959-62 period ranged from 151 to 187 mg/l and averaged 171 mg/l.

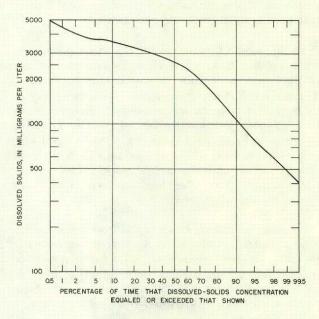


Figure 6.—Duration Curve of Dissolved Solids for Red River Near Gainesville, Texas, 1953-63

Chloride

The concentrations of chloride in surface water of the Red River basin vary from several thousand to less than 100 mg/l. Concentrations are generally less than 250 mg/l in all streams not affected by natural or oil-field brines. Brines in the drainage areas of the Prairie Dog Town, Salt, and North Forks Red River, Pease River, and North and South Wichita Rivers degrade the quality of the Red River throughout its reach in Texas. The annual weighted-average chloride concentration of the Red River near Gainesville (1944-46, 1953-63, 1966-67) has ranged from 283 to 717 mg/l, and at Denison Dam (1944-67), it has ranged from 139 to 431 mg/l. Chloride concentration of the main stem is generally more than 500 mg/l almost as far downstream as Lake Texoma, but it is less than 250 mg/l through the last 100-150 miles of its reach in Texas. Tributaries downstream from the Wichita River generally have chloride concentrations less than 50 mg/l.

Hardness

Surface water in the Red River basin generally ranges from moderately hard (61-120 mg/l) to very hard (more than 180 mg/l). Waters of streams in the western and central parts of the basin that contain high concentrations of dissolved solids are very hard, often having more than 500 mg/l hardness. The streams draining Cretaceous rocks in the eastern part of the basin generally contain waters that are moderately hard, even though they usually have a low dissolved-solids content.

Other Constituents

Other constituents of importance in the evaluation of the chemical quality of a water include silica, sodium, bicarbonate, sulfate, fluoride, and nitrate.

Silica concentrations in the Red River basin range from less than 10 to nearly 70 mg/l. In the western part of the basin, water from streams draining the Tertiary, Triassic, and Permian rocks usually contains more than 20 mg/l of silica. In the central part of the basin, streams draining rocks of Pennsylvanian age usually contain 10 to 15 mg/l of silica; and in the eastern part of the basin, water from rocks of Cretaceous age usually contains less than 10 mg/l of silica. The annual weighted-average silica concentration of the Red River at Denison Dam has usually been about 10 mg/l.

Sodium concentrations range from several thousand to less than 100 mg/l. Concentrations are generally less than 100 mg/l in streams unaffected by natural or oil-field brines. The sodium concentration of the Red River is usually more than 250 mg/l upstream from Lake Texoma and 100 to 250 mg/l from Lake Texoma to Index, Arkansas.

Bicarbonate concentrations are usually less than 250 mg/l in surface waters in the basin; bicarbonate is the principal anion in most waters unaffected by brines. The annual weighted-average bicarbonate concentrations of the Red River near Gainesville and at Denison Dam have usually been less than 150 mg/l.

Sulfate concentrations vary widely in the Red River basin. Streams draining Permian rocks north of the Priarie Dog Town Fork Red River have a sulfate content of several hundred mg/l. Sulfate is the principal anion in these waters. In Prairie Dog Town Fork Red River and in the streams draining Permian rocks south of the Red River, sulfate occurs in high concentrations; but chloride is the principal anion. Downstream from the Permian rocks, the tributaries contain less than 50 mg/l of sulfate. The annual weighted-average sulfate concentration of the Red River near Gainesville (1944-46, 1953-63, 1966-67) has ranged from 169 to 450 mg/l, but it has usually been more than 250 mg/l. The annual weighted-average of sulfate at Denison Dam (1944-67) has ranged from 100 to 297 mg/l, but it usually has been less than 250 mg/l.

Fluoride concentrations are generally less than 1.0 mg/l except in some of the streams that drain the Ogallala Formation. Tule Creek near Silverton at times contains more than 5.0 mg/l of fluoride.

Nitrate concentrations are usually less than 5.0 mg/l, except in some of the heavily irrigated areas of the High Plains where concentrations sometimes exceed 10 mg/l.

Water Quality in Reservoirs

Chemical analyses for most of the principal reservoirs in the Texas part of the Red River basin are given in Table 6. Most of the reservoirs are on tributaries where quality-of-water problems are less severe than on the main stem.

Buffalo Lake

When sampled in 1951, water in Buffalo Lake contained 472 mg/l dissolved solids, 27 mg/l of chloride and was very hard. Principal chemical constituents were sodium and bicarbonate.

Bivins Lake

Chemical analyses are not available for Bivins Lake, but analyses for downstream sites indicate that the stored water contains less than 500 mg/l of dissolved solids, is hard, and has calcium, sodium, bicarbonate, and sulfate as the principal chemical constituents.

Baylor Creek Reservoir

Very limited data indicate that Baylor Creek Reservoir impounds water containing between 500 and 1,000 mg/l of dissolved solids. The water is very hard and of a calcium sulfate type.

Greenbelt Reservoir

Greenbelt Reservoir was not impounding water during this study, but the chemical quality of its water can be inferred from analyses of Salt Fork Red River near Clarendon. Analyses of samples collected during low flow indicate that the dissolved-solids content seldom exceeds 500 mg/l. No data are available on the quality of water at high flow, but it is likely that the dissolved-solids content would be less than at low flow. Therefore, water impounded in the reservoir probably will contain less than 500 mg/l of dissolved solids, be very hard, and of a mixed chemical type.

Lake Kemp and Diversion Lake

Lake Kemp and Diversion Lake were constructed in 1923 and 1924, respectively, and are two of the oldest major reservoirs in the Red River basin. Water is released from Lake Kemp into Diversion Lake and then released or withdrawn for industrial use and irrigation. The reservoirs are downstream from natural salt-contributing areas. Sources of natural pollution of the Wichita River above Lake Kemp were investigated by Joerns (1961) and by the U.S. Public Health Service (1964). Chemical analyses indicate that since construction the impounded water has usually contained 2,000 to 3,000 mg/l dissolved solids. Calcium, sodium, sulfate, and chloride are the principal dissolved constituents. The water is suitable for irrigation for only highly salt-tolerant crops.

Santa Rosa Lake

Water stored in Santa Rosa Lake is low in dissolved solids, moderately hard, and of a calcium bicarbonate type.

North Fork Buffalo Creek Reservoir

Chemical analyses are not available for North Fork Buffalo Creek Reservoir, but analyses for North Fork Buffalo Creek near Iowa Park indicate that at high flow the water is of good quality; dissolved-solids content is less than 200 mg/l. Analyses of water at low flow, however, indicate oil-field brine pollution. The quality of the stored water will depend upon the extent to which brine reaches North Fork Buffalo Creek upstream from the reservoir.

Lake Wichita

Natural runoff into Lake Wichita is probably of good quality. However, the lake receives return flow from areas irrigated with water from Lake Kemp, and also is degraded with water from oil fields. The dissolved-solids content usually exceeds 1,000 mg/l. An analysis in June 1965 showed that the water in Lake Wichita contained 1,450 mg/l of dissolved solids; calcium, sodium, sulfate, and chloride are the principal chemical constituents.

Lake Kickapoo

Water stored in Lake Kickapoo usually contains less than 250 mg/l of dissolved solids and is moderately hard. Principal dissolved constituents are calcium, sodium, and bicarbonate.

Lake Arrowhead

Impoundment of water in Lake Arrowhead began in 1966, and no analyses of the stored water were available during the study period. The quality of the stored water can, however, be inferred from records for the daily sampling station, Little Wichita River near Henrietta. During the period of daily record (1952-55, 1959-66), the annual weighted-average dissolved-solids concentration has ranged from 124 to 286 mg/l, and averaged 218 mg/l. The water was of a sodium chloride type—probably because of oil-field brine reaching the stream.

Farmers Creek Lake

When sampled in 1967, water in Farmers Creek Lake contained 294 mg/l of dissolved solids and was hard. Principal chemical constituents were calcium, sodium, bicarbonate, and chloride.

Hubert H. Moss Lake

Chemical-quality data are not available for Hubert H. Moss Lake, but records from adjoining watersheds in the Trinity River basin indicate that the reservoir, when filled, will contain moderately hard water having a low dissolved-solids content.

Lake Texoma

Denison Dam which forms Lake Texoma was built in 1942 by the U.S. Army Corps of Engineers for flood control and hydroelectric power. Increasing needs for water have caused Lake Texoma to be considered as a source of water for public supply even though it has generally been too highly mineralized for this use. Water from Lake Texoma is pumped to Lake Randall to augment the municipal supply for the city of Denison. The city of Sherman has studied the practicability of damming off the Big Mineral Arm of the lake to obtain a municipal supply (Mendieta and Skinner, 1966).

Since 1965, the dissolved-solids content of water in Lake Texoma has ranged from 969 to 1,230 mg/l. Chloride has ranged from 325 to 442 mg/l, and sulfate from 228 to 296 mg/l. Although the dissolved-solids content of Lake Texoma water varies from year to year, 23 years of records collected since impoundment began in 1944 show a definite trend of increasing mineralization. Annual weighted-average concentrations of dissolved solids for the outflow station at Denison Dam are shown on Figure 7. The net quantity of water available annually at Denison Dam (expressed as annual

outflow plus change in storage, in thousands of acrefeet) is also shown on Figure 7. The net annual water available also represents the inflow to Lake Texoma minus losses due to evaporation, infiltration, and diversion. Along with the trend of increasing mineralization in Lake Texoma, Figure 7 shows a general trend toward less available water.

The mineralization of water of Lake Texoma is increasing because of (1) a continuous salt load reaching the reservoir from upstream natural and man-made brine sources, (2) decreasing inflows to the reservoir because of upstream impoundments, (3) increasing dissolved-solids concentrations of inflows because of upstream impoundments of good-quality water, and (4) increasing dissolved-solids concentrations because of evaporation from Lake Texoma and from impoundments upstream.

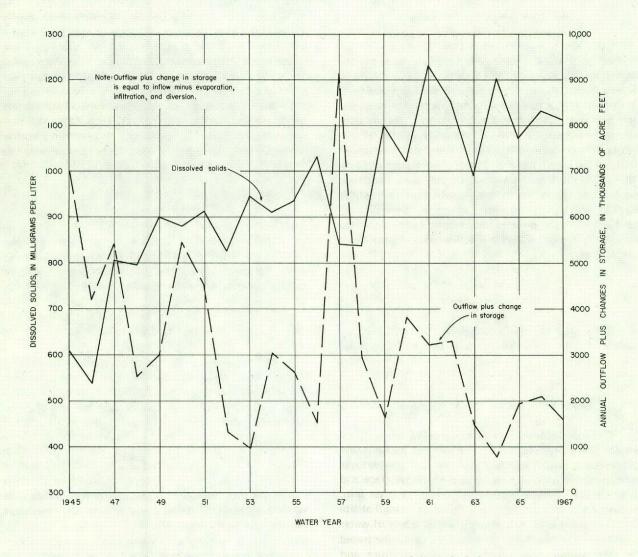


Figure 7.—Graph Showing Dissolved-Solids Content and Quantity of Water in Lake Texoma, 1945-67

Kane (1967, p. 17) shows average annual net lake-surface evaporation for Lake Texoma during the 1940-65 period to be 30 inches per year. At elevation 617 feet (top of power pool), Lake Texoma covers 89,000 acres. Therefore, evaporation losses from Lake Texoma may be more than 200,000 acre-feet per year. West of Lake Texoma average net annual evaporation losses increase rapidly to more than 50 inches in the High Plains. Waters that are released from impoundments above Lake Texoma after having been degraded by evaporation, further degrade the quality of inflows to the reservoir.

To aid in the evaluation of water quality in Lake Texoma, the Geological Survey made two reconnaissance-type surveys of the reservoir. The surveys were made in March and July 1967 to obtain data for different seasons of the year. Measurements of specific conductance, temperature, and dissolved oxygen were made at sites throughout the reservoir and at various depths at each site. Water samples were collected at selected sites for laboratory analysis. Figure 8 is a map of the reservoir showing the observation sites.

The first survey was made March 21-23. During this period, surface elevation remained almost constant at approximately 603 feet above mean sea level (contents, 1,710,000 acre-feet), and water was being released through the powerhouse. The reservoir was well mixed during this survey. Dissolved-solids content, estimated from specific conductance values and verified laboratory analyses (dissolved solids equals approximately 0.58 specific conductance), increased only slightly with depth at each site. In the Red River arm, dissolved-solids content was nearly uniform at about 1,200 mg/l from site 1C to site 24C, but increased upstream to about 1,700 mg/l at site 38C. In the Washita River arm, concentrations decreased in an upstream direction to about 1,000 mg/l at site 12C. Temperatures generally were about 1°C lower at the bottom of the lake than at the surface. Dissolved oxygen was nearly uniform throughout the vertical profile at each site-near saturation at the surface and only about 1 mg/l less near the bottom. Vertical profiles for sites 1C and 3C are shown on Figure 9.

During the second survey, made July 25-27, surface elevation was about 614 feet (contents, 2,480,000 acre-feet, an increase of 770,000 acre-feet since the March survey). The dissolved-solids content varied only slightly with depth at each site except at site 38C where the concentration was 1,090 mg/l at the surface and 2,290 mg/l at the bottom. At all the other sites on the Red River arm, dissolved-solids content was near 1,000 mg/l, usually slightly less at the surface and slightly more at the bottom. In the Washita River arm, the dissolved-solids content varied from 935 mg/l at site 7C to about 700 mg/l at site 12C. A thin layer of water about 40 feet below the surface was less concentrated than the water above and below. Temperature and oxygen stratification was evident in all areas of the

reservoir. At site 1C, temperature decreased from 25.8°C at the surface to 21.4°C at the bottom, and dissolved oxygen decreased from 7.2 mg/l at the surface to 0.0 mg/l at the bottom. At most sites, temperature and oxygen decreased slightly with depth through the top 50 feet, then decreased sharply through the next 10 feet, and was nearly uniform through the remaining depth. Vertical profiles for sites 1C and 3C are shown on Figure 9, and longitudinal profiles for the Red River and Washita River arms during the July 25-27 period are shown on Figure 10.

Lake Texoma, like many reservoirs in the southwest, undergoes thermal stratification in the summer and becomes almost completely mixed during the winter. During the summer, the more concentrated inflow from the Red River tends to flow along the bottom of the reservoir and the less concentrated inflow from the Washita River tends to seek an intermediate depth. Dissolved-solids content of the reservoir increases upstream in the Red River arm and decreases upstream in the Washita River arm. During the winter, oxygen is available at all depths throughout the reservoir, but during the summer is generally deficient at all depths greater than 50 feet.

Lake Randall

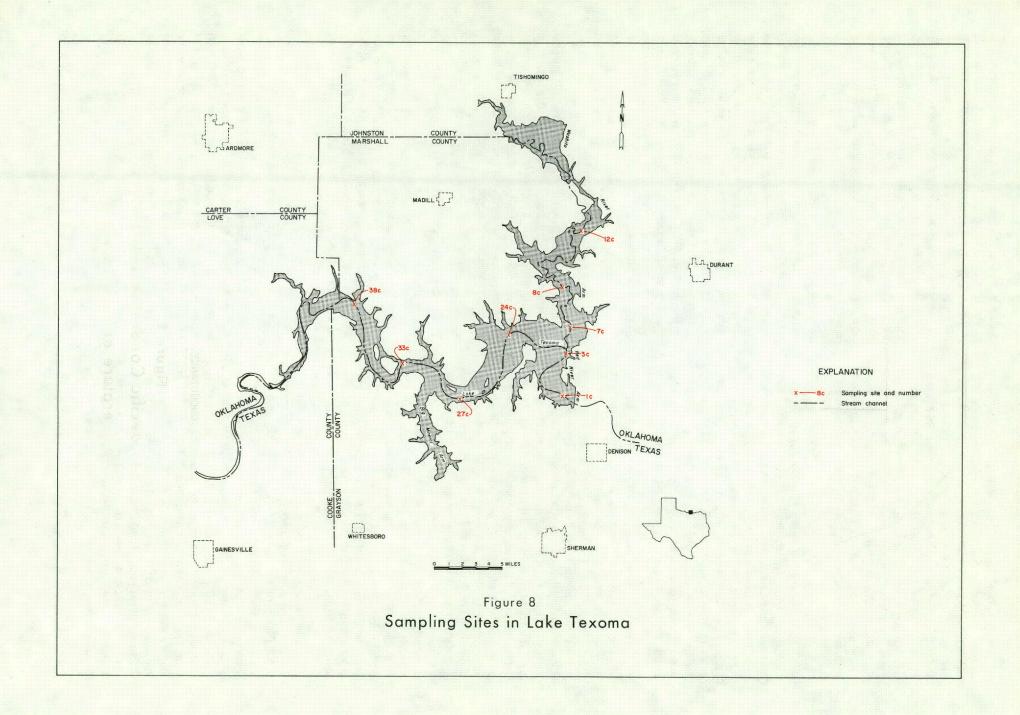
Lake Randall, a small reservoir owned by the city of Denison, is used as a municipal supply. Water is pumped from Lake Texoma to augment the normal yield, and quality of the water in Lake Randall is therefore determined by the proportion of water that is pumped from Lake Texoma.

Brushy Creek Reservoir and Coffee Mill Creek Lake

Although no chemical analyses are available for either of these reservoirs, the water quality can be inferred from records for nearby Bois d'Arc Creek and from records for watersheds in the adjacent Trinity River basin. Water in these areas is usually hard and of a calcium bicarbonate type. The dissolved-solids content averages less than 250 mg/l.

Pat Mayse Reservoir

Pat Mayse Reservoir was not impounding water during this study, but its quality can be inferred from analyses of Sanders Creek near Chicota (site 104). High flows in Sanders Creek contained less than 100 mg/l of dissolved solids, and the reservoir should store water containing less than 200 mg/l of dissolved solids. The water will be moderately hard and of a calcium bicarbonate type.



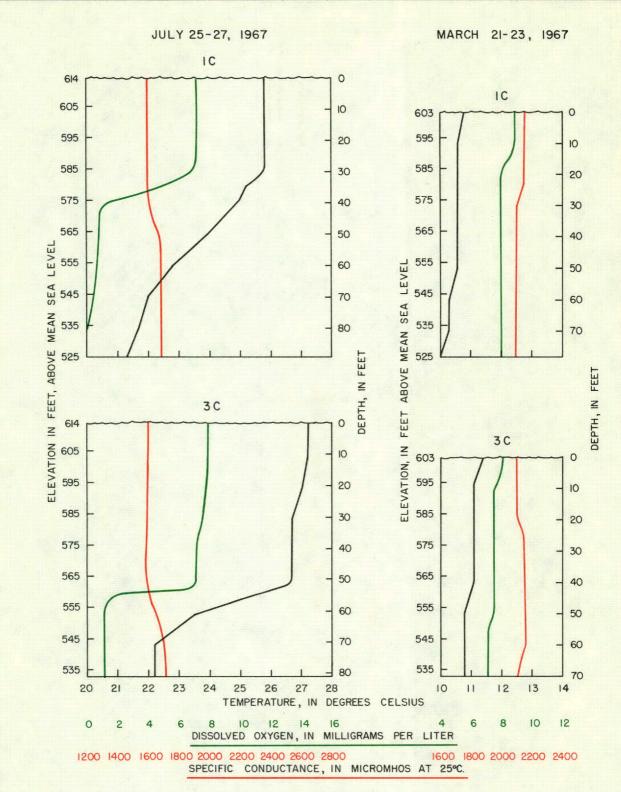
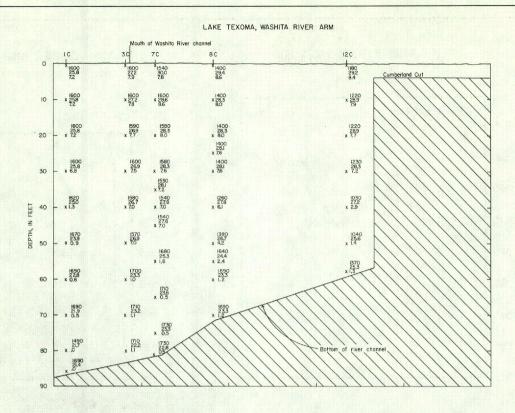
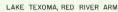
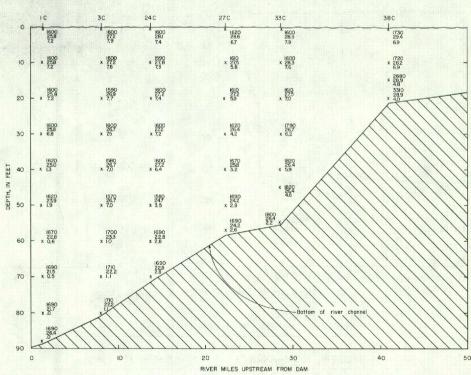


Figure 9

Vertical Profiles of Specific Conductance, Dissolved Oxygen,
and Temperature of Lake Texoma







EXPLANATION

25.8 7.2

3 C

Specific conductance, in micromhos at 25°C. Temperature, in degrees Celsius Dissolved oxygen, in milligrams per liter Sampling site

Figure 10

Longitudinal Profiles of Lake Texoma Showing Water Quality, July 25-27, 1967

Lake Crook

Lake Crook is owned and operated by the city of Paris for municipal water supply. When sampled in 1960, the reservoir water contained 70 mg/l of dissolved solids, was soft, and of a calcium bicarbonate type.

Water Quality at Potential Reservoir Sites

One of the principal objectives of this study was to appraise the quality of water available for storage at potential reservoir sites in the Red River basin. Several potential sites suggested by various agencies are shown on Figure 11. In the following discussion, evaluations of water quality are based on 1967 conditions and the names of potential reservoir sites are those in use as of December 31, 1967.

Mackenzie

A reservoir on Tule Creek at the Mackenzie site would impound water of good quality; the dissolved-solids content would be less than 250 mg/l. The water would be hard and have calcium and bicarbonate as its principal ions.

Buck Creek

Information is not available on the chemical quality of flood flow in Buck Creek. Low-flow samples show high concentrations of calcium and sulfate. A reservoir on Buck Creek would probably impound water containing more than 1,000 mg/l of dissolved solids.

Lelia Lake Creek

Although water samples have been collected periodically for several years, data on the chemical quality of flood flows in Lelia Lake Creek is lacking. Available data indicate that a reservoir on Lelia Lake Creek would impound water of mixed chemical composition containing about 500 mg/l dissolved solids. The water would undoubtedly contain more than 500 mg/l of dissolved solids at times.

Dozier Creek

A reservoir on the Salt Fork Red River at the Dozier site would probably store water containing more than 1,000 mg/l of dissolved solids. A 2-year (1953-54) daily record is available for a station near Wellington (site 47). The weighted-average dissolved-solids concentrations were 1,300 mg/l in 1953 and 1,100 mg/l in 1954.

Lower McClellan Creek

Limited chemical-quality data indicate that a reservoir on McClellan Creek would store water containing about 500 mg/l dissolved solids. The water would be of a mixed chemical type and would be very hard.

Sweetwater Creek

A reservoir on Sweetwater Creek would store water of acceptable quality for most uses. The water would be of a calcium sodium bicarbonate type and contain less than 500 mg/l dissolved solids.

Ringgold

Daily chemical-quality records for the Little Wichita River near Henrietta and Ringgold indicate that a reservoir near the Ringgold site would impound water of good quality. At the Henrietta station the annual weighted-average concentration of dissolved solids was less than 300 mg/l each year during the period of record (1953-66). Oil-field brine has reached the streams in the watershed, and has caused some deterioration of the otherwise excellent-quality water.

Timber Creek and Bois d'Arc Creek

The water available for storage at these sites is of a calcium and sodium bicarbonate type and is moderately hard. The dissolved-solids content should be less than 250 mg/l.

Big Pine

A reservoir on Big Pine Creek would impound water containing less than 150 mg/l dissolved solids. The water would be low in all dissolved constituents and would be soft.

Pecan Bayou

The water in Pecan Bayou is always low in dissolved constituents, therefore, water impounded at the Pecan Bayou site would contain less than 100 mg/l dissolved solids and would be soft.

Barkman Creek

A reservoir on Barkman Creek would impound water containing less than 100 mg/l dissolved solids. The water would be soft and low in all dissolved constituents.

Present and Future Water-Quality Problems

Natural and in the past, oil-field brines are the principal degrading influences on water-quality of the Red River. The highly mineralized waters from salt sources in the western part of the basin cause the water of the Red River to be undesirable for public supply throughout most of its reach in Texas. The salinity problems in the Red River basin have been intensively studied by various Federal agencies. The U.S. Public Health Service (1964) reported that there are 10 primary natural brine emission areas in the Red River basin. Figure 4 shows the locations of the primary sources and several secondary sources, and includes the average daily salt load that the Public Health Service calculated to be contributed by each primary source. A detailed description of each source is given in the report by the Public Health Service (1964).

The U.S. Army Corps of Engineer District at Tulsa, Oklahoma, has prepared a report (1966) on the feasibility of plans to control the major salt sources in the Arkansas and Red River basins and has constructed an experimental control project at Estelline Springs on Prairie Dog Town Fork Red River. Congress has authorized construction of additional salt-control projects on three tributaries of the Wichita River above Lake Kemp, and the Corps of Engineers has proposed five additional projects in the Red River basin, four in Texas and one in Oklahoma.

The plan for control of salt in the Wichita River consists of three low-water dams, pumping facilities, and pipelines for collecting highly mineralized water and moving it to storage basins for evaporation. The Corps estimates that the chloride load reaching Lake Kemp would be reduced by about 80 percent and the sulfate load by about 30 percent. Chloride concentrations would rarely exceed 200 mg/l and sulfate would usually be less than 500 mg/l.

Maximum control of both man-made and natural brine pollution throughout the upper Red River basin, as

proposed by the Corps of Engineers, would greatly improve the quality of the water impounded in Lake Texoma. The Corps estimates that chloride concentrations would be less than 110 mg/l 50 percent of the time, and would seldom exceed 150 mg/l. Dissolved solids would be reduced to an average of about 820 mg/l and sulfate to about 220 mg/l. With its quality thus improved, water of the lower Red River will be suitable for a wider variety of beneficial uses.

Impoundments on tributaries in Texas and Oklahoma and in Lake Texoma are also causing degradation of water quality in the main stem. Of the 18 existing reservoirs on tributary streams in Texas, 16 are impounding water of better quality than is carried by the Red River. Of the 11 potential reservoir sites discussed, 9 would impound water having better quality than water of the main stem.

The city of Sherman is considering damming off Big Mineral Arm of Lake Texoma which is less mineralized than the main body of the reservoir. Most existing and any future reservoirs on Oklahoma tributaries will impound waters of better quality than is carried by the mainstem Red River. Removal of the water of tributary streams leaves water of poorer quality in the Red River. Surface impoundments in the Red River basin will further degrade water of the basin because of evaporation losses. Figure 7 shows the trend of increasing mineralization of Lake Texoma waters, which must result in part from evaporation losses. According to Kane (1967, p. 17) average annual net lake surface evaporation rates vary from 10 inches near the Texas-Arkansas line to more than 50 inches in the High Plains.

The population of the Red River basin in Texas has doubled in the past 25 years. The population growth is expected to be greater in the next 25 years. Along with increasing demands for municipal supplies, more water will be needed for industry and irrigation. If a significant part of these supplies is to come from surface waters of the Red River basin, a maximum effort to improve water quality will be required.

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WATER YEAR	U.S.G.S. WATER-SUPPLY PAPER NO.	T.W.D.B.	OKLAHOMA WATER RESOURCES BOARD REPORT
1940-45	and the second	*1938-45	
1946	1050	*1946	+1946-49
1947	1102	*1947	+1946 49
1948	1133	*1948	+1946-49
1949	1163	*1949	+1946-49
1950	1188	*1950	+1950
1951	1199	*1951	+1951
1952	1252	*1952	+1952
1953	1292	*1953	+1953
1954	1352	*1954	+1954
1955	1402	*1955	+1955
1956	1452	Bull. 5905	+1956
1957	1522	Bull. 5915	+1957
1958	1573	Bull. 6104	+1958
1959	1644	Bull. 6205	+1959
1960	1744	Bull. 6215	+1960
1961	1884	Bull. 6304	+1961
1962	1944	Bull. 6501	+1962
1963	1950	Rept. 7	+1963
1964		# #	#
1965		#	#
1966		#	#

^{* &}quot;Chemical Composition of Texas Surface Waters" was designated only by water year from 1938 through 1955.

Published as U.S. Geological Survey open-file report.

^{+ &}quot;Chemical Character of Surface Water of Oklahoma" was designated only by water year from 1946 through 1963.

Table 4.--Index of Surface-Water Records for the Red River Basin, Texas

Refer-		Drainage			Type and peri	od of record		
ence no.	Stream and location	area (sq. miles)	Daily chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reservoir content	Water temperature
1	Tierra Bianca Creek above Buffalo Lake near Umbarger	2075		1938-54, 1966				1949-54, 196
2	Buffalo Lake near Umbarger	2075			1951		1938-54, 1966	
3	Tierra Blanca Creek below Buffalo Lake near Umbarger			1966				1966
4	Palo Duro Creek near Canyon	982		1942-54				1949-54
5	Prairie Dog Town Fork Red River near Canyon	3369		1937-49				
6	Prairie Dog Town Fork Red River above Stockton Dam near Canyon			1965		1961-65		1961-65
7	Lake Stockton near Canyon				1965-66			
8	Prairie Dog Town Fork Red River below Stockton Dam near Canyon					1961-65		1961-65
9	Prairie Dog Town Fork Red River above Palo Duro Park near Canyon				1961	1961-65		1961-65
10	Prairie Dog Town Fork Red River below Palo Duro Park near Canyon				1950, 1 <mark>96</mark> 1, 1964-65	1961-65		1961-65
11	North Tule Draw at Reservoir near Tulia	189		1938-66			1938-66	1949-66
12	Tule Creek near Silverton	1150		1964-66	1964-66			1964-66
13	Prairie Dog Town Fork Red River near Brice	5972	1950-51	1938-44, 1949-51, 1959-62				1949-51, 1959
14	Mulberry Creek near Brice	534	1950-51	1949-51				1949-51
15	Prairie Dog Town Fork Red River near *478 Lakeview	6792		1963-66				1963-66
16	Little Red River at State Highway 70 near Turkey				1959			
17	Prairie Dog Town Fork Red River near Estelline	7293		1937-47	1949-50			
18	Estelline Spring near Estelline				1959, 1962	1959, 1962		
19	Baylor Creek Reservoir near Childress				1949-50			
20	Baylor Creek near Childress				1948			
21	Salt Creek 12 miles northwest of Childress	THE PARTY IS			1959			
22	Prairie Dog Town Fork Red River near Childress	7725		1964-66	1948-49, 1963			1964-66
23	Buck Creek near Wellington	210			1945, 1947-48 1951-53, 1955-56 1959, 1962	1950-64		1950-64
24	Red River near Quanah *750	8321		1959-66	1959			1959-66
25	North Groesbeck Creek near North Groesbeck	150			1951-53 1957-58, 1961	1951-64		1951-64
26	South Groesbeck Creek near Goodlett				1962	1962-64		1962-64
27	South Groesbeck Creek near Acme	146			1951-53 1957-58, 1961	1951-64		1951-64
28	Groesbeck Creek at State Highway 283 near Quanah	303		1962-66	1950-53, 1957-58 1960, 1965-66			1962-66

Table 4.--Index of Surface-Water Records for the Red River Basin, Texas--Continued

Refer-	The second secon	Drainage			Type and peri	od of record		
no.	Stream and location	area (sq. miles)	Daily chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reservoir content	Water temperature
29	Wanderers Creek at Odell	199			1950-53 1957-58, 1960	1949-66		1949-66
30	Carroll Creek near Clarendon	177			1951-53	1948-60		1949-60
31	Kelly Creek near Clarendon					1961-65		1961-65
32	Greenbelt Reservoir near Clarendon						1966	
33	Salt Fork Red River near Clarendon	457			1950-53 1956, 1960, 1962	1950-60		1950-64
34	Salt Fork Red River above Saddlers Creek north of Lelia Lake				1951			
35	Barton Creek northeast of Clarendon				1959			
36	Saddlers Creek 8 miles northeast of Clarendon				1951			
37	Salt Fork Red River north of Lelia Lake				1951, 1959			
38	Lelia Lake Creek below Bell Creek near Hedley	74				1964-66		1964-66
39	Lelia Lake Creek near Hedley	79			1950-53 1957-58, 1964	1951-66		1951-66
40	Salt Fork Red River near Hedley	744	1957-61			1951, 1956-62		1951, 1956
41	Whitefish Creek near Alanreed				1962			
12	Whitefish Creek south of McLean				1951, 1962			
43	Whitefish Creek northeast of Hedley	57			1951, 1962			
44	Gyp Creek north of McKnight				1951			
45	Salt Fork Red River north of Quail				1959, 1963			
46	Dozier Creek near Wellington				1950-51, 1953 1955, 1960	1950-60		1950-60
47	Salt Fork Red River near Wellington *703	1222	1952-54	1962-66	0,000			1962-66
48	North Fork Red River west of Kellerville	1.00-			1959			
49	McClellan Creek at State Highway 70 near Boydston	14077			1950			
50	Lake McClellan near Jericho				1951			
51	McClellan Creek at State Highway 273 near McLean	- 4			1965	1965		
52	North Fork Red River near Shamrock	1082		1964-66	1951-53 1958-59, 1964-66	1951-63		1951-66
53	Sweetwater Creek at State Highway 152 west of Mobeetie				1951			
54	Sweetwater Creek at State Highway 152 southeast of Mobeetie				1951			
55	Sweetwater Creek near Wheeler	164			1951-53 1955, 1957-58	1951-64		1951-64
56	Sweetwater Creek near Kelton	287		1961-66	1962-66			1961-66
57	Elm Creek near Shamrock	 			1946-47, 1950-53 1955, 1958-59 1962	1947-65		1947-65
58	Elm Creek above Wolf Creek near Lutie				1962			

Table 4 .-- Index of Surface-Water Records for the Red River Basin, Texas--Continued

Refer-		Drainage			Type and peri	od of record		
no.	Stream and location	area (sq. miles)	Daily chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reservoir content	Water temperatur
59	Wolf Creek at mouth near Lutie				1962			
60	Elm Creek below Wolf Creek near Lutie				1959, 1962			
61	Quitaque Creek near Quitaque	293		1945-59	1945-46, 1950-51	1960-66		1960-66
62	Roaring Springs near Roaring Springs				1937, 1952-56 1958-60, 1962	1937, 1943-66		1949-66
63	Middle Pease River near Paducah				1950, 1959			
64	Salt Springs tributary to Middle Pease River 14 miles northeast of Paducah				1959			
65	Pease River near Childress *755	2747		1959-62	1959			1959-62
66	Pease River near Crowell	3037	1942-43	1924-47				
67	Pease River near Vernon *759	3488		1959-66	1942, 1951			1959-66
68	Red River near Burkburnett	20570		1959-66	1959	1924-25		1959-66
69	North Wichita River 11 miles south of Paducah				1951-54			
70	North Wichita River 10 miles southeast of Paducah				1951-54			
71	Salt Creek 4 miles southeast of Paducah				1951-52, 1958			
72	Salt Creek at mouth 8 miles southeast of Paducah				1939, 1951-54, 1956, 1958-59			
73	North Wichita River below Salt Creek 12 miles southeast of Paducah	44			1952, 1958			
74	North Wichita River near Paducah *771	540		1961-66	1958-59, 1965-66	1951-54		1961-66
75	North Wichita River near Truscott *757	937		1959-66	1954, 1956, 1959, 1965-66	1952-57		1959-66
76	South Wichita River at Guthrie	39			1950, 1958-59 1963			
77	South Wichita River tributary 6 miles east of Guthrie				1958			
78	South Wichita River 6.5 miles east of Guthrie	<u> </u>			1953-54, 1956 1958-59			
79	South Wichita River near Benjamin *756	584		1959-66	1949,1953-54,1956 1959, 1965-66	1952-57		1959-66
80	Wichita River near Seymour	1874		1959-66	1953-54, 1958, 1965-66			1959-66
81	Lake Kemp near Mabelle	2086			1939,1942,1946,195 1954-55, 1964-65	2	1922-66	
82	Wichita River near Mabelle *752	2086		1959-66	1965-66	1952-58		1959-66
83	Santa Rosa Lake near Vernon				1966			
84	Beaver Creek near Electra *751	652		1960-66	1966			1960-66
85	North Buffalo Creek near Iowa Park				1961-63	1961-63		1961-63
86	Buffalo Creek near Iowa Park	-			1964-65	1963-65		1963-65
87	Wichita River at Wichita Falls *503	3140		1900-1902, 1910-11, 1938-66	1951			1949-66

Table 4.--Index of Surface-Water Records for the Red River Basin, Texas--Continued

Refer-	Calle and resident to		Drainage			Type and peri	od of record		
ence no.	Stream and location		area	Daily chemical quality	Discharge	Periodic chemical quality	Periodic discharge measurements	Reservoir content	Water temperature
88	Lake Wichita at Wichita Falls					1944, 1946, 1952 1954, 1959, 1965			
89	Wichita River at Farm Road 171 near Byers					1949, 1951, 1958			
90	Lake Kickapoo near Archer City		275			1946, 1952, 1954 1957, 1964-65		1946-66	
91	Little Wichita River near Archer City		481	1953-55	1932-56				1949-56
92	Lake Creek near Henrietta					1959			
93	Little Wichita River near Henrietta	*704	1037	1953-55, 1959-66	1953-66				1953-66
94	Dry Fork Little Wichita River near Henrietta					1959			
95	East Fork Little Wichita River near Henrietta		178		1953-66	1959, 1964-66			1963-66
96	Little Wichita River near Ringgold		1350	1959-62	1959-65				1959-65
97	Red River near Terral, Oklahoma	*507	28723		1938-66				1949-66
98	Farmers Creek Reservoir near Nocona					1967			
99	Red River near Gainesville	*508	30782	1944-46, 1952-63 1966-67	1936-67				1944-46 1949-66
100	Washita River at Farm Road 2564 near Allison					1965	1965		
101	Lake Texoma near Denison		39719					1942-66	
102	Red River at Denison Dam near Denison	*522	39720	1944-1967	1923-67				
103	Bois d'Arc Creek near Randolph		72		1962-66	1966			1962-66
104	Sanders Creek near Chicota				1961-62 1965-66	1961-62, 1965-66			
105	Red River at Arthur City	*527	44531		1905-11, 1936-66	1961-63			
106	Lake Crook near Paris				1960				
107	Big Pine Creek near Manchester				1961-62	1961-62			
108	Pecan Bayou near Clarksville		100		1962-66				1962-66
109	Red River near New Boston		47555		1960-63				
110	Barkman Creek near Leary				1961-62	1961-62			
111	Red River at Index, Ark.	*529	48030	1960-63	1936-66				

^{*} U. S. Public Health Service has collected chemical quality records at this site. Number is Public Health Service site number.

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas

								Bi-							solved s			ness aCO,	50-	Specific	
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	1
		ii allaati	14 11 11 11			13.	PRAIR	IE DOG	TOWN	FORK RE	D RIVER N	EAR BI	RICE								
Water year 1950 Maximum, May 16-18,21,1950. Minimum, Oct. 10, 1949 Weighted average	10.2 283 69.3	26		650 397 327	134 42 56	296 20 77	5	114 82 110		1750 1100 930	4740 280 1190		2.5 2.9	10300 2090 3360	14.0 2.84 4.57	284 1600 629	2170 1160 1050	2080 1100 956		15200 2680	
Water year 1951 Maximum, Jan. 15-16, 1951 Minimum, May 17-20 Weighted average	10.1 6032 162	25 20 23		813 202 229	217 36 41	441 30 45	8	153 110 129		2270 583 669	7110 440 663		3.8	14900 1650 2140	20.3 2.24 2.92	406 26900 940	2920 652 752	2800 562 647		21400 2540 3370	8.
		P. Januar				388	1	4. MU	LBERRY	CREEK	NEAR BRIC	E		1000							
Water year 1950 Maximum, June 24, 1950 Minimum, July 24 Weikhted average	0.25 204 38.2	18	6585H	472 128 244	144 17 43	21 4 8	3	79 80 102		1730 334 697	270 49 111		0.5 1.8 1.7	2920 693 1260	3.97 .94 1.73	2.0 382 131	1770 391 786	1700 326 702		3480 918 1650	8.
Water year 1951 Maximum Mar. 2, 1951 Minimum, June 1-3 Weighted average	. 40 597 20.7	30 28 31		500 101 200	113 19 40	22 3 8	6	124 113 112		1680 235 566	255 49 115		.0 2.5 2.6	2870 526 1120	3.90 .75 1.52	3.1 893 63.0	1710 330 664	1610 238 572		3390 789 1480	8.
	100						40.	SALT	FORK F	ED RIVE	R NEAR HE	DLEY								ARE I	
Water year 1957 Maximum, Jan. 18, 1957 Minimum, Aug. 29		34 15	inne.	371 38	125	26 5 3		276 111		1260 57	328 25	1.0		2520 231	3.43		1440 126	1210 35		3260 382	7.8
Water year 1958 Maximum, Jan. 1, 3, 1958 Minimum, Oct. 15, 17-18, 1957		28 17		408 64	108 19	18		217 136		1280 166	250 125	. 6	.5	2370 575	3.22		1460 238	1280	2.1	2860 925	
Water year 1959 Maximum, Mar. 11-14, 16,20, 22, 25, 23, 1959 Minimum, Mar. 5		24 17		215 72	74 24	27 8	8	184 100		800 218	325 101	. 9	1.8	1810 563	2.46		841 278	590 196	4.2	2570 917	7.0
Water year 1960 Maximum, Nov. 3-9, 17-21, 23-30, 1959		46 13		130 45	49 8.	14 5 2		113 131		516 59	148 27		1.8	1090 27 0	1.48 .37		526 147	434 40	2.8	1510 413	
Water year 1961 Maximum, June 26-29, July 6, 1961 Minimum, Oct. 10-16, 1960		46 19		9 2 50	39 12	14 4		78 135		380 90	169 52	.8	.5 2.8	968 347	1.32 .47		390 174	326 64	3.1 3.8	1370 565	7.5

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

	4							Bi-							calcula		Hard as C:	ness aCO ₃	So-	Specific	
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car-	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (C1)	ride	Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
						4	7. S	ALT FOR	RK RED	RIVER	NEAR WELL	INGTO	N								
June to September 1952 Maximum, Sept. 21-30,1952 Minimum, June 22-24	3.22 51.4			558 199	101 47		51 15	135 136		1690 621	185 125	0.8		2780 1220	3.78 1.66	24.2 169	1810 690	1700 578		3240 1710	
Water year 1953 Maximum, Dec. 18-30, 1952	15.0	20		600	108	1	31	174		1650	255	. 8	5.8	2860	3.89	116	1940	1800	1.3	3370	7.8
Minimum, Aug. 6, 8, 18-20, 1953		26 29		141 238	25 47		51 02	125 128		359 681	65 134	. 5	1.5	730 1300	.99 1.77	382 194	455 788	352 682		1080 1730	
Water year 1954 Maximum, Aug. 12-20, 1954 Minimum, Oct. 21-24, 1953 Weighted average	414	29 14 25		518 127 188	141 21 40		13 53 02	114 131 141		1590 295 518	1030 71 141	. 4	3.5 3.0 3.0	3980 677 1100	5.41 .92 1.50	67.6 757 357	1870 404 634	296	6.2 1.1 1.8	5470 960 1550	7.7
							6	6. PE	ASE RI	VER NEAD	R CROWELL								4		
July to September 1942 Maximum, July 16, 1942 Minimum, June 10				1010 390	199 79	26 5	60 22	99 75		2910 1150	4260 835		13	11100 3010			3350 1300	==		16000 4460	
Water year 1943 Maximum, Dec. 24, 1942 Minimum, Apr. 17, 1943 Weighted average	1060	12 13		864 326 424	170 33 64		50 34 93	145 76 112		2300 847 1130	6480 208 1250		2.5 1.5 3.2	14200 1600 3740	18.9 2.18 5.09	1840 4580 1620	2860 949 1320	2740 886 1230		20400 2420 5540	
						91	. LI	TTLE W	ICHITA	RIVER	NEAR ARCH	ER CI	TY								
January to September 1953 Maximum, Aug. 15-17, 1953 Minimum, Aug. 20-28 Weighted average	21.1	18		72 22 27	25 5.9 8.0	9	62 53 84	121 113 114		14 7. 7.	675 L 65 L 128	0.8		1220 230 328	1.66 .31 .46	19.3 13.1 5.52	282 79 100	184 0 7		2360 405 622	8.0
Water year 1954 Maximum, Sept. 19, 1954 Minimum, Oct. 22-27, 1953 Weighted average	1542	10		14 19	3 5	1	 20 34	165 59 73		3.: 4.:		. 5	4.8	2340 137 168	 .19 .26	570 31.1	590 48 68		1.3 1.8	3730 192 303	8.0
Water year 1955 Maximum, Nov. 17-18, 1954 Minimum, Sept. 25-26, 1955. Weighted average	3890	8.6 6.4 9.7		129 12 23	37 2.4 5.5	1	42 18 36	86 48 94		13 7.8 5.0		. 5	4.0 4.0 4.0	1890 95 197	2.57 .13 .27	8.42 998 34.2	475 40 80		11 1.3 1.8	3550 156 337	7.3
		-				9	3. L	ITTLE V	WICHIT	A RIVER	NEAR HEN	RIETT	A								
Dec. 1952 - Sept. 1953 Maximum, Mar. 15-16 Minimum, Mar. 14, 17-18 Weighted average	137	14 9.4 12		94 13 23	31 4.: 7.:	3	14 20 73	81 59 90		15 5.0 6.4		 	3.5	1700 111 286	2.31 .15 .43	474 41.1 10.7	362 50 88	296 2 14		3290 205 542	
Water year 1954 Maximum, Oct. 6, 1953 Minimum, Oct. 22-24,26-29 Weighted average	3326	16 7.2 12		60 8. 14	18 2 2.3 4.8	7	15 11 25	93 38 58		14 2.0 3.0		. 7	7.5 3.0 3.0	1310 66 147	1.78 .09 .20	654 593 81.0	224 32 55		12 .8 1.5	2460 116 236	

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

	-57 -				34		D-	Bi-		i i						ssolved calcula		Hard as Ca		So-	Specific con-	
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	ride	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate		duct- ance (micro- mhos at 25°C)	рН
						93. LI	TTLE	WICHITA	RIVE	R NEAR	HENRIETTA	Con	tinued									
Maximum, Sept. 24, 1955 Minimum, May 19 Weighted average	1430	11 7.2 9.7		113 7.8 19	33 2.3 4.1		81 8.4 36	74 34 69		18 2 4.3	980 10 56	0.4			1670 57 166	2.27 .08 .23	415 220 51.7	418 29 64	358 1 8	10 .7 1.9	3250 97 306	7.4
March to September 1959 Maximum, May 12 Minimum, June 23 Weighted average	2280	9.6 6.4 8.9		78 6.8 21	23 3.4 6.1		45 10 50	128 39 69		25 4.4 6.6	642 10 85	.6 .2 .3	2.5		1190 63 218	1.62 .09 .30	1510 388 46.7	289 31 78	184 0 21	8.8 .8 2.5	2300 116 404	
Maximum, June 2, 1960 Minimum, Mar. 26 Weighted average	245	7.8 9.3		16 25	3.8 6.9		 18 64	58 59 70		6.8 7.7	2500 26 114	.2			4120 110 270	5.60 .15 .37	1280 72.8 45.7	1060 56 91	1010 7 33	1.0 2.9	7520 204 498	6.3
dater year 1961 Maximum, June 1-8, 1961 Minimum, Oct. 15-16, 1960 Weighted average	789	7.3 9.3		6.8 22	2.9 6.2		 8.8 59	94 34 71		4.0 6.6	1450 9.2 100	 .3 .3			2440 59 243	3.32 .08 .33	3.29 126 33.7	666 29 80	589 1 22	 .7 2.9	4590 104 458	7.6
Maximum, Nov. 4, 1961 Minimum, June 30, 1962 Weighted average	918	8.8		109	29 6.6		05 52	79 51 79		16 4.2 6.1	1000 10 89	. 4 	7.4 3.2		1710 76 234	2.33 .10 .31	679 188 66.3	392 34 87	327 0 21	11 2.2	3170 129 436	7.5
ater year 1963 Maximum, Feb. 1-4, 8-14, 16-19, 1963	.1 5.9 101	5.3		210 17	76 4.9		28 33	170 8 63		29 7.8	1440 8.0 53	==	1.0		2470 30 158	3.36 .04 .21	. 67 66. 0	836 8 63	697 1 12	9.4	4560 44 290	7.6
Maximum, Dec. 15-17, 1963 Minimum, Sept. 16-17, 1964. Weighted average	8.6 118 56.3	7.2 11 11		74 13 25	8.6 3.1 6.3		06 20 56	62 72 92		27 8.0 7.5	570 14 89	 •4 			1020 106 242	1.39 .14 .33	23.7 33.8 73.0	220 45 88	169 0 17	9.0 1.3 2.3	1990 182 455	7.6 6.8
ater year 1965 Maximum, Apr. 16-18, 1964 Minimum, Aug. 16-17 Weighted average	416	11 8.4 9.5		65 11 25	18 3.1 6.7		35 11 52	113 50 85		17 6.0 8.9	450 11 87	.2	3.2 1.5 1.0		855 77 232	1.16 .10 .32	191 86.5 48.0	236 40 89	144 0 19	6.7 .8 2.2	1600 134 439	
Maximum, July 12, 1966 Minimum, Apr. 26-30 Weighted average	3788	6.6 7./3 8.0		106 12 15	30 3.3 4.0		43 16 25	132 50 58		21 3.2 4.7	710 20 37	.5	. 5		1290 90 124	1.75 .12 .16	261 920 172	388 44 54	280 2 6	7.5 .9 1.2	2490 162 231	7.7
The state of the s						9	6. L	ITTLE V	VICHIT	'A RIVER	NEAR RIN	GGOLD				2011						
March to September 1959 Maximum, Mar. 16-18, 1959 Minimum, Sept. 4 Weighted average	32.1 383 130	7.8 7.6 9.6		205 5·2 15	63 1.5 4.3		83 4.3 33	150 22 55		34 2.6 5.1	1640 4.0 52	0.3 .1 .2			2810 38 151	3.82 .05 .21	244 39.3 53.0	770 19 55	648 1 10	12 .4 1.9	5200 60 279	8.1
Water year 1960 Maximum, June 3, 1960 Minimum, Oct. 3-4, 1959 Weighted average	173	10 10 11		302 4.5 17	96 2.9 5.4	12	60 6.3 40	60 33 63		62 1.4 6.0	2680 5.0 66	.5	1.0		4440 47 180	6.04 .06 .24	1590 173 52.5	1150 23 65	1100 0 13	16 .6 2.2	7860 72 326	6.8

Table 5 .-- Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

			eliació					Bi-			re centere			Dis	ssolved alcula		Hardi as Ca		So-	Specific con-	
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)		Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate		duct- ance (micro- mhos at 25°C)	
The state of the s	September 1					96. LIT	TLE W	ICHITA	RIVER	R NEAR R	INGGOLD	Conti	nued								
Water year 1961 Maximum, Oct. 8-15, 1960 Minimum, Oct. 16-17 Weighted average	1046	6.4 7.4 8.8		89 6.5 18	23 2.4 5.4		95 8.4 43	85 32 63		18 4.2 6.9	770 8.0 72	0.4		1340 55 187	1.82 .07 .25	94.1 155 40.7	316 26 67	247 0 15	9.7 .7 2 3	2570 99 354	6.6
Water year 1962 Maximum, Nov. 8-11, 1961 Minimum, Sept. 6-7, 1962 Weighted average	1900	8 · 4 15 12		112 5.0 17	29 2.9 5.4		64 2.6 34	64 30 66		17 .4 5.5	1100 3.0 54	.6		1870 45 168	2.54 .06 .23	42.9 231 74.4	399 24 65	346 0 12	12 .2 1.6	67	6.8 7.0 6.7
	Ų.						99	. RED	RIVER	R NEAR G	AINESVILL	Е				1987 (8) (8) (4)				9	
May to September 1944 Maximum, Sept. 21-30 Minimum, June 13-16 Weighted average	957			384 83 181	86 19 43		40 63 33	144 115 126		940 181 450	2070 252 705		1.8 2.2 3.3	4790 757 1880	6.51 1.03 2.56	3570 1960 7410	1310 285 628	1190 191 525		7530 1320 3070	
Water year 1945 Maximum, Jan. 11-20, 1945 Minimum, Sept. 30 Weighted average	42300			310 36 97	97 7.3 24		30 36 94	214 94 137		751 31 169	1740 62 335		3.5 1.2 3.5	4040 250 891	5.49 .34 1.21	5500 28600 10100	1170 120 340	997 43 228	 	6550 403 1540	
October 1945 to April 1946 Maximum, Jan. 23-31, 1946. Minimum, Oct. 1-3, 1945 Weighted average	6547			248 36 	75 7.3 		66 36 	287 94 		521 31 	1440 62 		2.5 1.2	3290 250 	=	=======================================	928 120 	692 43		5510 403 	
Water year 1953 Maximum, Apr. 1-11, 1953 Minimum, July 22-23 Weighted average	3090			450 39 169	94 8.4 38		30 66 36	153 105 127		1190 42 412	2750 101 698		7.0 4.2 4.6	6480 342 1910	8.81 .47 2.60	47940 2850 3360	1510 132 578	1380 46 474	19 2.5 7.8		7.4 8.0
Water year 1954 Maximum, Aug. 30-31 Minimum, May 12-13 Weighted average	41950			460 52 115	86 6.2 23		40 74 42	123 102 123		1190 70 246	2000 114 394		1.4	5210 412 1140	7.09 .56 1.55	11310 46670 9510	1500 155 382	1400 72 280	13 2.6 5.4	8010 698 1890	8.0
Water year 1955 Maximum, Aug. 1-6, 1955 Minimum, Sept. 26-30 Weighted average	14660			356 48 141	73 12 32		67 83 82	126 120 130		900 70 326	1400 118 462		1.7	3830 400 1370	5.21 .54 1.86	13470 15830 9730	1190 170 484	1090 72 377	11 2.8 5.6		7.4 7.7
Water year 1956 Maximum, June 21-31, 1956. Minimum, July 11 Weighted average	1220			424 45 146	103 12 37		70 89 23	134 90 136		1120 57 341	2250 155 533		3.3	5490 446 1530	7.47 .61 2.08	7860 1470 8990	1480 160 516	1370 86 405	15 3.1 6.2	8670 776 2470	8.0
Water year 1957 Maximum, Sept. 11-12, 1957 Minimum, Apr. 26-30 Weighted average	49580			384 51 107	69 7.5 23		20 49 69	118 130 136		917 58 209	1700 67 283		2.4	4260 335 917	5.79 .46 1.25	5120 44850 18530	1240 158 362	1140 52 250	13 1.7 3.9		7.6 8.2
Water year 1958 Maximum, Nov. 1-3 Minimum, Nov. 4 Weighted average	1110			344 28 164	78 3.2 44		10 6.4 34	196 84 151		793 14 383	1850 7.0 717		4.6	4680 115 1950	6.36 .16 2.65	8630 345 10520	1180 83 590	1020 14 466	14 .3 7.8	7460 176 3100	7.7

Table 5 .-- Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

								Bi-			1170					solved s		Hard as Ca	ness aCO ₃	So-	Specific	-
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	- 1
							99. R	ED RIV	ER NEA	R GAINE	SVILLEC	ontir	nued									
Water year 1959																		100				
Maximum, April 20, 1959	652			300	100]	170	116		815	1950				4690	6.38	8260	1160	1060		7150	8.
Minimum, Sept. 5	1960 1534			48 154	16 31		93 359	104 125		73 375	60 566		0.8	===	472 1640	.64 2.23	2500 6790	185 512	100 409	6.9		7.
Water year 1960																						
Maximum, July 1-8, 1960	799			348	90		190	138		990	1900				4760	6.47	10270	1240	1130		7080	8.0
Minimum, Oct. 4, 1959				36	6.3		32	108		32	45		. 4		217	.30	16350	116	28		362	7.
Weighted average	2916			147	36		364	144		342	590				1660	2.26	13070	515	397	7.0	2590	
Water year 1961	0142			416	100	,	340	132		1210	2120				5630	7.66	32580	1450	1340	15	8050	8.
Maximum, July 16-18, 1961 Minimum, Sept. 16-20	2143 5412			416	12		95	118		66	146		3.6		463	.63	6770	168	72		785	8.0
Weighted average	3044			158	43		399	148		390	644				1820	2.48	14960	571	450			
Water year 1962																						
Maximum, April 18-22, 1962.	311			252	120		923	184		718	1580			0.57	3880	5.28	3260	1120	969	12	6180	7.9
Minimum, Sept. 2	835			46	3.6		21	140		31	21				221	.30	498	130	16	. 8	294	7.8
Weighted average	2591			124	29		287	140		283	456				1340	1.82	9370	429	314	6.0	2120	
Water year 1963	i Lu.													40		7 5 0	11010	1000	1140	10	0000	0 (
Maximum, June 26-July 1, 1963	751			392	69		23	146		1150 72	2350 19		33	. 43	5580 292	7.59	11310 3330	1260	1140 52	19.7	8800 460	8.2
Minimum, Nov. 26, 1962 Weighted average	4220 1289			58 141	8.6 36		381	148 150		324	611			.26	1590	2.16	5540	498		7.0		
Water year 1967 Maximum, Mar. 1-31, 1967	179	4.6		258	86	812	8.1	232		610	1400		2.5		3310	4.50	1600	997	807	11	5350	7.6
Minimum, July 7-8	5110	7.9		56	10	86	4.4			60	146	0.2	. 2		433	. 59	5970	180		2.8		
Weighted average	1316	7.9		140	30	329	5.8	141		302	545		1.3		1430	1.94	5080	473	358	6.0	2400	7.5
						1	.02. R	ED RIV	ER AT	DENISON	DAM NEAR	DENI	SON		er case						177	
May to September 1944	χ.														1999							
Maximum, Aug. 11-20, 1944.	746			148	37		315	162		323	520		4.9		1430	1.94	2880 497	522 352	389 222		2430 1610	7.8
Minimum, June 11-20	204			98	26 32		194 255	158 161		183 255	318 424		4.8		902 1180	1.23	946	446	314	==	2040	
Weighted average	297			126	32		233	101		233	424		4.2		1100	1.00	340	110	011		2010	
Water year 1945	149			127	34		296	172		290	465		2.0		1300	1.77	523	457	316		2280	8.0
Maximum, Jan. 1-10, 1945 Minimum, July 1-10				68	19		84	149		98	146		.8		489	. 67	34200	248	9		852	8.0
Weighted average	7261			78	21		114	140		129	195		1.7		607	. 83	11900	281	166		1070	
Water year 1946																	APPENDING SERVICE					
Maximum, Aug. 11-20, 1946	2043	-		71	23		115	165		106	195		1.0		592	.91	3680	272	3		1050	- 5
Minimum, Oct. 11-20, 1945				62	17		63	134		86	115		. 5		410	. 65	47000 9000	224 245	114	==	762 874	Ξ.
Weighted average	6199			67	19		84	152		100	139		.9		486	. 13	9000	243	120		014	
Water year 1947	0247			142	34		298	145		321	490		1.5		1360	1.85	8620	497	378		2290	· .
Maximum, Oct. 20-31, 1946	2347 2528			143 70	23		116	163		120	185		1.5		644	.88	4400	269	136		1070	
Minimum, Oct. 1-10 Weighted average	7923			90	24		149	148		164	250		2.0		805	1.09	17200	323	202		1340	
Water year 1948																						
Maximum, Sept. 1-30, 1948	2124	10		92	27		170	137		184	288		.8		905	1.23	5190	340	228		1500	
Minimum, Dec. 1-31, 1947	2351			78	23		141	134		176	215		. 5		762	1.04	4840	289	179		1230	
Weighted average	3528			85	24		150	140		175	239		1.5		797	1.08	7590	310	196		1310	
Water year 1949											205				1040	1 41	0400	950	040		1710	
Maximum, July 1-31, 1949	2307	12		100	25		209	135		217	332		2.8		1040 774	1.41	6480 4530	352 310	242		1710 1340	
Minimum, Nov. 1-30, 1948		7.8		85	24		150 178	131		175 193	246 290		1.9		901	1.05	9440	334	222		1520	
Weighted average	3880	8.6		91	26		119	137		193	290		1.9		301	1.23	9440	004	222		1010	

Table 5 .-- Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

	1	, ,	Ca	lculat	ted val	lues for	sodiu	ım plus	pota	ssium ar	e centere	a bet	ween	tne t				1				
	Property and the second							Bi-								calcula		Hard as C	ness aCO ₃	So-	Specific con-	
Date of collection	Mean discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
						102. RE	D RIV	ER AT	DENISO	N DAM NI	EAR DENIS	ONCo	ontinu	ied	30 (1		500-1	1965				
Water year 1950 Maximum, Mar. 1-31, 1950 Minimum, Sept. 1-30 Weighted average	. 10330	11 12 13		96 78 88	27 21 24	1	86 56 74	129 126 130		215 165 191	300 245 276	 	3.5 2.0 3.1	 	977 790 882	1.33 1.07 1.20	4670 22000 16800	350 281 318	245 178 212	==	1600 1320 1460	==
Water year 1951 Maximum, June 1-30, 1951 Minimum, Sept. 1-30 Weighted average	. 2563	11 14 11		99 77 91	29 21 25	1	98 43 79	149 132 141		207 157 187	325 225 290	 	.8 1.0 1.2	121 14-	1010 725 913	1.37 .99 1.24	70800 5020 17200	366 278 330	244 170 214	==	1670 1220 1500	8.9 7.6
Water year 1952 Maximum, Aug. 1-31, 1952 Minimum, Oct. 1-31, 1951 Weighted average	. 1841	8.2 11 9.5		89 68 83	27 23 26	1	83 44 61	145 135 142		200 160 185	285 212 250	<u>=</u>	2.0 2.0 1.9	 52	894 722 827	1.22 .99 1.12	7580 3590 5140	333 264 3 14	214 154 198	4.3 3.9 3.9	1530 1180 1380	7.9 7.8
Water year 1953 Maximum, Aug. 1-31, 1953 Minimum, Oct. 1-31, 1952 Weighted average	. 2394	11 11 9.5		92 88 92	28 28 29	1	97 88 90	140 140 142		205 203 207	315 295 305	=======================================	1.0 1.2 1.9	==	995 912 944	1.35 1.24 1.28	7880 5890 4720	344 334 348	230 220 232		1620 1520 1570	8.0 7.9
Water year 1954 Maximum, Nov. 1-30, 1953 Minimum, July 1-31, 1954 Weighted average	. 4608	8.8 15 12		100 84 89	27 21 24	1	33 65 84	123 128 128		239 178 200	370 275 299	0.5 .3 .4	2.0	0.18 .20 .18	1040 830 908	1.41 1.13 1.23	3330 10330 9680	360 296 320	260 191 216	4.2	1750 1390 1530	
Water year 1955 Maximum, Sept. 1-30, 1955. Minimum, Oct. 1-31, 1954 Weighted average	. 1109	11 12 9.9		106 86 96	21 22 22	1	16 77 93	122 122 126		240 190 209	342 278 306	.4	1.5	.14	1000 880 937	1.36 1.20 1.27	7260 2630 6990	351 305 330	251 205 227	4.9 4.4 4.5	1720 1480 1570	7.8 7.7
Water year 1956 Maximum, Sept. 1-30, 1956. Minimum, Jan. 1-31 Weighted average	. 3627	12 11 11		128 102 106	32 21 23	1	80 98 23	126 121 122		315 228 248	448 305 346	.5	. 9	.20 .17 .17	1280 954 1030	1.74 1.30 1.40	4920 9340 9870	450 341 359	346 242 259	5.7 4.6 5.0	2190 1600 1720	
Water year 1957 Maximum, Dec. 1-31, 1956 Minimum, June 1-30, 1957 Weighted average	. 66910	12 11 11		134 78 89	32 15 18	1	11 33 67	123 107 112		342 165 195	485 202 258	==	.7 1.8 2.2		1380 696 840	1.88 .95 1.14	2520 125700 24700	465 256 296	364 168 204	3.6	2290 1130 1370	
Water year 1958 Maximum, Sept. 1-30, 1958. Minimum, Oct. 1-31, 1957 Weighted average	. 5720	9.4 15 11		100 85 91	24 20 20	1	20 44 71	148 132 136		209 173 185	345 225 268	==	.5 1.2 1.0	II.	981 733 837	1.33 1.00 1.14	4280 11320 9760	348 294 309	226 186 198	5.1 3.6 4.2	1700 1240 1400	8.0 8.2
Water year 1959 Maximum, Aug. 1-31, 1959 Minimum, Oct. 1-31, 1958 Weighted average	. 1823	10 8.8 9.4		112 99 104	27 24 28	2	60 34 52	131 138 135		259 218 246	408 365 390		.4 .5 .8	==	1140 1020 1100	1.55 1.39 1.50	14230 5020 6830	390 346 374	283 232 264	5.7 5.5 5.7		7.4 8.2
Water year 1960 Maximum, Sept. 1-30, 1960. Minimum, May 1-31 Weighted average	. 2703	12 8.6 9.5		113 99 101	28 25 26	1	64 87 22	153 155 129		266 223 238	400 280 343	.6 .3 .4			1160 900 1020	1.58 1.22 1.39	6040 6570 14330	397 350 359	272 223 254	5.8 4.3 5.1		7.5 7.5

Table 5.--Summary of Chemical Analyses of Water at Daily Stations on Streams in the Red River Basin, Texas--Continued

				Ta.					Bi-							ACCUSANCE OF SERVICE STREET	solved s		Hard as Ca		So-	Specific con-	
	Date of collection	Mean discharge (cfs)	Silica (SiO ₂)		Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	ride	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
N Company							102. H	RED RI	VER AT	DENIS	ON DAM	NEAR DENI	SON	Conti	nued								
Water year 196	1 1 20 1061	3593	11		120	37	31	00	138		312	470	0.3	2.8		1320	1.80	12810	452	338	6.2	2210	7.7
Minimum, Oct.	1-30, 1961 1-31, 1960	12040 4299	9.4		110 117	33	20	63 78	128 134		286 297	410 431	.3	$\frac{.4}{1.2}$		1170 1230	1.59 1.67	38030 14280	410 428	305 318	5.7	2010 2100	7.6
Water year 196		402.4	0.0		121	37	21	97	130		316	470	. 4	. 5		1320	1.80	17160	454	348	6.1	2220	7.2
Maximum, Oct.	1-31, 1961 . 1-30, 1962	4814 3772	9.9		100	30		38	139		256	360	. 4	1.2		1060	1.44	10800	373	259	5.4	1800	
	age	4527	8.9		111	34	2	53	136		277	403	. 4	1.4		1150	1.56	14100	420	308	5.4	1980	
Water year 196		3.503	0.0		105	31	91	25	150		249	350	4	1.8		1050	1.43	4260	390	266	5.0	1820	6.9
	1-30, 1963 1-30, 1963	1501 2862	9.2		105 98	27		99	140		236	302		1.0		941	1.28	7270	356	241	4.6	1570	
	age	3029	9.3		99	29		11	133		244	326	. 4	1.2		989	1.35	8090	366	256	4.8	1670	
Water year 196	4	1100	0.1		111	35	2	98	160		300	440	4	1.8		1270	1.73	4036	418	288	6.3	2060	7.5
Maximum, Sept	1-30, 1964 1-30, 1963	803	8.1		111	35		53	130		285	402		.2		1160	1.58	2515	421	314	5.4	2000	7.4
	age	1510	8.4		111	35		67	135		290	420	. 4	1.2		1200	1.63	4900	422	312	5.6	2040	
Water year 196	55		0.0		100	25	0	83	120		296	440	3	1.8		1230	1.67	2710	414	316	6.1	2060	7.4
	1-30, 1964	815 1939	6.2		108 96	35 30		09	135		236	325		2.2		969	1.32	5070	363		4.8	1720	7.2
	1-31, 1965	1943	5.6		101	30		38	135		251	373	.3	1.1		1070	1.46	5610	376	266	5.3	1850	
Water year 196		200			100	20	0	61	127		272	442	4	1.8		1180	1.60	4170	393	289	5.7	2040	7.0
	1-31, 1966	1310 2056	3.2		108	30 32		47	126		246	400		1.0		1090	1.48	6050	385		5.4	1940	
	1-31, 1965		2.4		110	30		55	138		264	403	. 4	.8		1130	1.54	8610	397	284	5.5	1980	
Water year 196						00	070		123		292	438	. 1	.8		1220	1.66	6360	404	303	5.9	2070	7.1
	1-28, 1967	1931 2840	4.5 2.3		114	29 25	273 220	5.5	122		228	355		1.0		996	1.35	7640	350	250	5.1	1720	
	1-31, 1967		2.5		106	28	248	5.4	123		253	404		1.0		1110	1.50	6990	380	279	5.5	1920	7.2
13000									111.	RED R	IVER AT	INDEX, AF	RK.										
Water year 196	51															1260	1.71	10590	445	204	5.3	2020	8 (
Maximum, Sept	6-10, 1961	3112			118 29	37		55 30	172 76		275 38	405 38		1.0		185	.25	12490	90		1.4	306	
Minimum, May Weighted aver	9-14, 1961	10190			75	20		39	116		161	219		.8		728	.99	26000	269		3.5	1200	7.8
Water year 196	62							45	136		268	405				1180	1.60	20950	430	318	5.1	1940	7.7
	21-31, 1961				55	====		21	92		24	24				157	.21	6190	88	12	1.0		7.5
Weighted aver	e 6-9, 1962	10930						11	116		122	177				609	. 83	18000	230	134	2.9	972	7.9
Water year 196	63							17	140		245	348				1090	1.48	6770	396	275	4.7	1740	8.
Maximum, Sep	t. 12-14, 1963.	2300						17 40	148 104		47	62				292	.40	24600	138	49	1.5	476	8.3
	1-6				55			96	121		110	149				538	.73	10120	218	115	2.6	887	7.9

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas (Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.) Dissolved solids Hardness Specific as CaCO. So-(calculated) con-Bidium Mag-Po-Carduct-Fluo- Ni-Bo-Date Calcar-Calad-Iron Sulfate Chloride Silica Sodium tasbon-Nonance pH neride trate ron Milli-Tons Discharge bonof cium Tons cium. sorp-(SO4) (C1) (SiO₂) (Fe) (Na) sium ate sium (NO.) (B) grams per carmicro-(Ca) ate (F) collection (cfs) Mag-(HCO₃) (CO₃) per tion per (Mg) (K) mhos at bonacreday neratio liter 25°C) ate foot a sium 2. BUFFALO LAKE NEAR UMBARGER 238 0 804 8.2 472 0.64 31 81 14 372 73 27 2.0 2.0 0.11 4.6 0.09 44 May 2. 1951.... PRAIRIE DOG TOWN FORK RED RIVER ABOVE STOCKTON DAM NEAR CANYON 6. 727 8.3 455 0.62 130 0 4.6 12 122 288 96 32 5.0 0.8 Apr. 26, 1965.... b0.08 13 32 0 1.0 393 7.2 . 2 227 .31 145 28 13 35 14 27 185 1.4 Aug. 24..... .20 17 7. LAKE STOCKTON NEAR CANYON 188 0 1.4 542 7.5 0.40 42 20 43 237 49 22 1.5 0.2 295 0.3 Feb. 4, 1965..... 120 0 . 6 300 6.8 6.8 .2 157 15 152 16 .8 7 34 8.5 Nov. 17..... 142 0 1.0 421 7.1 28 8.0 180 38 17 1.1 1.0 233 Aug. 10, 1966..... 3.0 34 14 157 0 1.2 466 7.5 . 2 260 17 34 9.3 203 43 20 1.2 35 Jan. 20, 1967..... .1 460 7.8 1.5 2.0 265 142 0 1.5 9.0 185 49 25 .9 29 17 41 Aug. 25..... PRAIRIE DOG TOWN FORK RED RIVER ABOVE PALO DURO PARK NEAR CANYON 9. 1.65 634 472 2.3 1550 7.3 133 197 668 42 3.1 0.0 1210 185 42 0.57 38 Dec. 1. 1961..... 10. PRAIRIE DOG TOWN FORK RED RIVER BELOW PALO DURO PARK NEAR CANYON 2.22 765 1950 7.8 36 0.5 1720 854 ___ 60 89 108 1040 Oct. 18, 1950..... -- 38 314 1460 1350 2800 7.3 2570 3.50 0.27 42 448 82 213 125 1660 58 2.4 . 0 Dec. 1, 1961..... 2540 7.8 50 2380 3.24 1580 1440 1.2 172 1540 1.8 .0 .15 26 468 100 110 Dec. 2, 1964..... 1610 1490 1.7 2900 7.3 2.7 2600 3.54 144 1690 55 .0 510 82 162 Feb. 4, 1965..... .40 27 2.6 . 2 2680 3.64 1620 1530 1.9 2920 7.0 102 1770 54 492 95 180 .21 35 Apr. 26..... 1440 1320 1.6 2660 7.3 137 144 1480 52 2.4 .2 2300 3.13 462 70 Aug. 24..... b.28 27 11. NORTH TULE DRAW AT RESERVOIR NEAR TULIA 169 0 414 8.2 17 8.0 1.6 1.5 0.05 234 0.32 8.8 233 4.8 0.17 48 12 17 May 2, 1951..... 12. TULE CREEK NEAR SILVERTON 280 6.9 164 0.22 117 0 0.6 7.8 14 147 18 4.4 0.9 0.5 13.3 12 34 Sept. 24, 1964.... 259 46 2.5 832 8.6 93 340 138 42 4.4 521 .71 .50 23 23 49 Jan. 7, 1965..... 291 6.8 125 0 .9 1.2 169 .23 . 5 165 11 2.0 37 8.0 13 4310 15 June 11..... 169 0 . 5 380 6.8 3.6 1.0 .2 221 .30 222 13 12 16 12100 18 48 June 11..... 266 6.6 154 .21 110 0 . 5 13 142 15 2.9 .7 . 5 6.1 12 34 June 13..... 212 22 1.7 646 7.8 2.6 .2 394 .54 50 21 58 232 98 2.8 .70 22 July 14..... . 52 170 21 2.3 635 6.9 .2 381 182 125 24 2.5 .72 14 40 17 69 Aug. 12..... 580 247 34 2.7 950 7.6 54 4.0 .2 ---259 190 47 31 99 14 Nov. 16..... .2 12 328 20 2.8 1090 3.0 191 53 6.6 .0 685 15 376 51 49 115 Mar. 10, 1966.... . 2 19 807 7.4 258 34 1.9 4.2 1.0 481 44 36 70 12 274 112 46 8.1 21 June 14..... 163 10 . 5 392 7.3 235 38 5.1 1.3 2.2 8.8 187 --6.3 14 49 10 15 Aug. 10..... 297 7.0 .6 1.5 129 0 . 2 3.6 166 --7.4 160 9.6 41 6.5 5.7 Aug. 25..... 185 11 16. LITTLE RED RIVER AT STATE HIGHWAY 70 NEAR TURKEY

5980

73200

2000 1190

119

115000

9900 9790

152000 7.8

See footnotes at end of table.

b0.02

Mar. 24, 1959.....

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

	1 1							Bi-								ssolved s		Hard as C	ness aCO ₃	So-	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Tiue	Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pH
						17.	PRAIR	IE DOG	TOWN	FORK RI	ED RIVER N	EAR E	STELL	INE								
July 29. 1949 Sept. 11, 1950	0.5 1000							130		1520 876	820 1110		2.0		Sec.	ETTENH MUS					4940 4910	7.
							18.	ESTE	LLINE	SPRING	NEAR ESTE	ELLINE			1 5367				in je			
Feb. 12, 1959 Mar. 24 Apr. 30, 1960 May 30, 1962	4.1 5.05	14		1510 1500 1460 1470	283 270 302 275	1770 1740 1680 1620	00	144 139 230 91		4380 4290 4220 4160	27200 27000 25800 25300				47500	64.6		4930 4840 4870 4790	4810 4730 4690 4710		61600 61500 60600 57900	7. 6.
		150)	9. B	AYLOR	CREEK	RESERVO	DIR NEAR C	CHILDR	ESS									
Aug. 19, 1949 Mar. 10, 1950		18	Major II	179 216	28 20		.1	83 98		488 541	7 8.0		0.2		765 930	1.04	er Skriger	562 621	494 540		963 1110	7.
							2	0. BA	YLOR (CREEK NE	EAR CHILDE	RESS						100				
lar. 1, 1948 Mar. 3		11 5.0	0.1	96 100	12 9.7		6.7 4.6	61 46		230 242	8.0 10		3.2		430 415			289 290	252		562	7.
						21.	SAL	T CREE	K 12	MILES NO	ORTHWEST C	F CHI	LDRES	S				164.4				
Mar. 24, 1959				1030	213	599	0	110		2860	9500				10.00			3459	3360		27200	7.
16818 B. 11						22.	PRAIR	IE DOG	TOWN	FORK RI	ED RIVER N	EAR C	HILDR	ESS			or or or other					
Sept. 23, 1948 Mar. 24, 1959 Apr. 21, 1963		15 9.8		1750 1860 1740	459 407 393	2140 2350 2060	00	106 86 96		4980 5440 4970	33700 36500 32300				62200 60100	81.8		6250 6320 5960	6250 5880		77100 77200 61300	
THE RESERVE							2	3. BU	CK CR	EEK NEAF	R WELLINGT	ON										
Sept. 17, 1945 Oct. 21, 1947 Sept. 23, 1948 Feb. 26, 1951 July 16 Aug. 15 Oct. 8, 1952	11.5	23 35		574 606 408 608 622 608	147 134 116 132 138	20 25	58 96 70 57	120 105 120 168 73 70		2000 1870 2100 1340 1960 2000	235 320 240 120 232 242		2.0 .8 1.5 3.0 4.0 3.0		3170 3420 2180 3160 3240 3210	2.96 4.30 4.41		1960 2040 2060 1500 2060 2120	1360 2000 2060 2010	 1.7	3710 3660 2530 3580 3610 3860	7. 7. 7.
May 11, 1953 Jan. 12, 1955 Jan. 12, 1956 Mar. 22, 1959 Aug. 2, 1962	3.06 2.68 3.57 b.08	15 29 19		568 568 574 390	130 125 117 187	14 13 19 12	12 37 06	73 77 157 247		1800 1770 1790 1610	225 225 242 94 42	0.5	7.0 6.6 		2930 2900 3020 982	3.98 3.94 4.11 		1950 1930 1910 1740 640	1890 1870 1780 1540 530	1.4 1.4 1.9 1.3		7. 7. 7.
								24.	RED R	IVER NEA	AR QUANAH					To Jack	100000					
Mar. 25, 1959	- 11			721	197	267	0	154		2310	4170						88343	2610	2480	23	14300	7.

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

	(Results in							Bi-		era littera					Dis	solved s	olids	Hard as Ca	ness	So- dium	Specific con-	
	Date of collection	Discharge (cfs)	Silica (SiO ₂)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
			-			25	. NOI	RTH GRO	DESBEC	K CREEK	NEAR NORT	TH GRO	ESBEC	K								
1 vr	16, 1951	5.23	21	624	127	3	82	176	1.01000000	19,00	555		5.9		3700	5.03		2080	1940		4760	
	22, 1952	4.14								1830	530		. 0					1100			4410 4550	
ec.	8, 1953	2.84			105			151		1810 1890	570 805		5.9		4070	5.54		2100	1980		5220	7.
	19, 1957	1.82 2.33		636 660	125 130		30 ·	192.	4	1920	830		8.8		4200	5.71		2180	2020	4.9	5360	8.
	3, 1961	1.90					22	188		1960	780				144		1	2160	2010		5100	7.
							26.	SOUTH	H GROE	SBECK CE	REEK NEAR	GOODI	ETT						: -			
une	7, 1962	6.25	26	32	4.4	l .	12	98		32	4.0	0.2	4.1	,1,	163	0.22		98	18	0.5	230	7.
							27	. sour	rh GRO	ESBECK C	CREEK NEAD	R ACME	4/					42003034			ery acertoses	
- 7	10 1051	6.05	22	 614	99	5	97	157		1750	292		6.5		3060	4.16		1940	1810		3630	7.
	16, 1951 22, 1952	4.00		612	102		79	203		1690	290		1.8		2990	4.07		1950	1780	1.8	3560	
	8, 1953	4.57	13				77	.==		1720	270		5.1		0050	4 01		1840	1730	2 0	3570 3470	7.
	19, 1957	2.09		600	84 97		201	137 93		1700 1730	280 235		3.8		2950 2880	4.01 3.92		1880	1810		3300	8
	20, 1958 3, 1961	4.66		595			.98	172		1750	240	-						1870			3340	7.
-	0, 1001					28. (ROESB	ECK CRI	EEK AT	STATE I	HIGHWAY: 28	33 NEA	AR QUA	NAH			# 1 TO 1 TO 1 TO 1					
			00	580	48		111	90		1790	425		1.2		3330	4.53		1640	1570		3590	7.
	21, 1950 15, 1951	11.0	29	582	122		71	150		1780	400	-22	4.5		3250	4.42		1950	1830		3810	7.
	15	10.0		564	117	2	92	102		1770	420		5.0		3230	4.39		1890	1800		3990	
	16	14.3	18	590	108		277	125		1780	398		2.8		3230 3480	4.39		1920 1970	1810 1910	52	3950 4030	
lug.	15	8.85	25	598	117		330	78		1940	428		3.5									
oct.	1	9.96		598	116		282	141		1800	418		3.0		3300 3260	4.49		1970 1980	1850 1860		4010 3 9 80	7.
	18	9.92		600	118		264	148		1760 1710	425 400		4.5		3190	4.34		1970			3900	7.
	22, 1952	9.66 83.6		254	52		17	106		714	190		3.5		1400	1.90		848		1.7	1830	7.
	13, 1953	6.50		586	117		307	84		1830	448		5.4		3350	4.56		1940	1870	3.0	4050	7.
	11		12	610	131		284	72		1870	472		8.2		3420	4.65		2060	2000	2.7	4170	7.
	8	11.0					268			1740	415		4.4								4070	
	19, 1957		14	592	102		359	132		1740	535		3.8		3410	4.64		1900 1700	1790 1620		4290 3570	7.
	21		28	532	90	231	7.8	90		1540 1690	375 415		5.2		2850 3210	3.88 4.37		1890	1810		3930	7.
Jan.	10, 1958	-	21	632	98		247	204				2477										
	20	4.72		585	109		320	75		1830	450	0.3	2.5		3360 2460	4.57 3.35		1910 1460	1850 1350		4020 3030	7.
	7, 1960	3.32		450	83 114		209	135 198		1330 1760	298 403	0.3	3.8		3260	3.33		2040			4010	7.
	8, 1965	2.3	8.3	630 660	121	328	5.3			1980	538	.4	2.2		3710	- 12		2140	2020	3.1	4510	7.
	31, 1966 5	1.6	5.0	658	124	347	8.7			1950	560	. 4	. 8	450	3700			2150	2070	3.3	4420	7.
								29.	WANDE	RERS CRI	EEK AT OD	ELL	Justil		are area o	Light		1275				Ħ,
eb.	20, 1950	4.54	8.4	104	66		26	188		415	150		7.8		1140	1.56		531	377		1600	
	17, 1951	3.98	16	94						524	170		11		205			114	27	0.9	1720 329	7.
	15, 1952	77.8	21				22	106		52	13		2.8		205	. 28		114		2.7	1779	7.
	13, 1953	2.04		139	74		158	384		446	148		10		1180	1.60		652	337			

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

								Bi-								ssolved s		Hard as Ca		So-	Specific con-	2
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	p
							29.	WANDER	ERS CI	REEK AT	ODELLCo	ntinu	ed									
t. 31, 1953	6.69							187		662	193		9.5		1	1 22		770	617		2090	8
c. 8, 1953	2.73			 114	67		77			610	195		8.3		1070	7 46					2170	8
b. 19, 1957 t. 21	2.40 6.5			186	64	131	55 5.6	353 368		399 493	128 138	0.8	11		1070 1240	1.46		560 727	270 425	2.8	1670 1720	
n. 10, 1958	6.02			263	103		48	418		906	230	. 7	10		1980	2.69		1080	738	3.3	2620	8
g. 20	1.22	37		46	38		91	165		199	81		15		616	.84		272	136	2.4	928	۶
pt. 6, 1960	1.08	28		55	36		80	228		158	64	. 6	21		555	.75	21400	285	98	2.1	860	7
		4.					3	30. CA	RROLL	CREEK N	FAR CLARE	NDON										
ly 20, 1951	0.14			36	19		32	209		43	15		1.5		294	0.40		168	0		450	۶
n. 23, 1952 n. 13, 1953	1.29	30		31	20		23	190		34 31	14		2.8		254	.35		160	4	0.8	551 488	
;. 24	. 44						30	208		48	15		1.8					177		1.0		
							32.	GREEN	BELT F	RESERVOI	R NEAR CL	AREND	ON									
15, 1967		13		56	20	43	5.2	198		97	42	1.2	0.5		375			222	60	1.3	621	
	100						33.	SALT	FORK I	RED RIVE	R NEAR CL	AREND	ON					100	E 93			
ot. 12, 1950		23		55	18		42	196		75	44		1.0		415	0.56		211	50		600	
v. 4	12.4			40	22		52	148		106	48		4.5		400	. 54		190	69		589	
1. 15, 1951	17.0 9.3	19		53 67	21 22		35 34	190 212		79 92	40 43		1.0		388 386	. 53		218 258	63 84		609 659	
2	10.6	26		61	24		36	202		106	37		1.5		402	. 55		250	83	II#	628	
. 21	12	22		53	21		32	164		79	50	-	1.0		342	. 47		218	84		626	
. 12	6.66			44	21		46	182		88	38		2.5		360	. 49		196	48		673	
y 20	1.16			56	21		72	144		137	85		1.0		498	. 68		226	108		774	
. 23, 1952	19.2	28		40	17		41	153		86 76	55 38		.0		242	47		170	44	1.4	704 526	
22													1.5		343	. 47			44			
. 8 . 13, 1953	1.25 9.26			48 42	24 21		81 48	168 146		101	108 50		.8 1.8		526 397	.72		218 192	81 72	2.4	841 669	
y 22	9.20							182		121	76	==			493	. 67		244	95		761	
. 24	1.05			50	21		78	123		123	108		. 5		536	.73		212	110	2.4	801	
. 26, 1956	2.36	34		47	21	100	69	151		121	74		. 5		440	.60		204	80	2.1	716	
19, 1960		26		61	13		42	205		68	38	0.7	2.0		352	. 48		206	38	1.3	557	
ie 2, 15, 23		30		61	17		44	191		93	43	. 8	1.2		405	. 55		222	66	1.3	609	
e 7, 9, 10 y 3, 10	==			39	6.6		18	136 172		25	16 62	. 5	3.5		196	. 27		124 224	13 83	.7	311 691	
y 6, 14, 20,27.		34		56	18		50	174		106	46	1.0	1.2		416	. 57		214	71	1.5		
. 18							-	140		23	10							120	6			
. 24	1.46			52	18		74	158 176		107 99	54 81	1.1	.2		442	. 60		198 204	68 60	2.2	638 729	
10, 1902	1.40	30		32					ADOUT							.00		204	0.5	2.2	129	
12 1051	11.7	0.4		7.4					ABOVE		RS CREEK	NORTH		LLIA		0.00		200	120		1110	
. 13, 1951	11.7	24		74	28		18	207		156	158		1.5		719	0.98		300	130		1110	
								RTON CI	REEK N	ORTHEAS	T OF CLAR	ENDON										
. 22, 1959	b3	36		112	47	2	76	228		402	332		0.9		1320	1.80		473	286	5.5	2070	

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

									Bi-			The state of					ssolved so		Hard as C		So-	Specific con-	
	Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO _s)	Sulfate (SO ₄)	Chloride (Cl)	True	Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pВ
							36. SAI	DDLERS	CREEK	8 MI	LES NORT	THEAST OF	CLARE	ENDON		100						2	50-101
	13, 1951 21	5.0			82 93	36 43	14 18		240 218		216 282	174 246		1.0		858 995	1.17 1.35		352 409	156 230		1300 1650	
								37.	SALT F	ORK R	ED RIVER	NORTH OF	FLELI	[A LA	Œ								
	21, 1951 22, 1959	b2.0	24		81 64	32 26	18		202 216		221 150	240 147		0.5		871	1.18		334 266	168 89	3.4	1490 1080	
									39. L	ELIA	LAKE CRE	EEK NEAR H	HEDLEY	Y									
Jan. Feb. Mar. Oct. Jan.	13, 1950 9, 1951 13 21 4, 1951 22, 1952	7.0 5.22 7.17	31 26 41 31		95 110 103 101 78	27 26 28 29 28 30	\$ 2 4	57 52 56 49 58	198 273 250 206 161 108		221 181 197 217 223 205 237	54 52 52 54 48 54 66	==	7.8 6.3 6.9		666 565 698 628 561 599	0.91 .77 .95 .85 .76		348 382 372 371 310	186 158 167 202 178	 1.4	890 894 921 933 858 942 864	8.
Oct. Jan.	22 8 13, 1953 12	11.1 5.58 5.79 8.22	44 40		60 73 66 51	28 28 17		54 52 32	143 105 170		215 217 89	48 51 23		7.7 9.2 3.2		601 542 372	.82 .74 .51		297 280 197	180 194 58	1.4 1.3 1.0	830 891 525 961	8.
Nov. Feb. Apr.	24	6.82 6.31 11.7 6.80	32 43		87 66 70 80	25 25 25 24 24	56	59 79 2.9 57 3.9	199 114 142 188		225 231 199 197 205	53 57 56 47 51	0.5	9.2 8.9 9.3 9.1 2.2		649 513 499 549	.88 .70 .68		320 268 273 298	174 156	1.9 1.5 1.5 1.5	956 806 767 822	7. 8. 7.
		Control of Control							41. W	HITEF	ISH CREE	EK NEAR AI	LANREI	ED		1665		400				100	
June	12, 1962								202		11		0.4			223			168	2	0.3	373	8.
	21, 1951 15, 1962		26		269	83	15		182 158	ITEFI	869 332	245 70	F MCLI	0.8 .6	0.19	1760 784	2.39		1010 430	864 300	1.4	2430 1050	
				100°s				43	. WHIT	TEFISH	CREEK N	NORTHEAST	OF HI	EDLEY							and the control		
	8, 1951 12, 1962	0.03	15		354	114		00 07	196 124		1210 1360	260 440		0.0	0.37	2250 3030	3.06		1320 1470			2820 3590	
									44.	GYP C	REEK NO	RTH OF Mcl	KNIGHT	Г						<u> Marina da da</u>		Charles Control	
Jan.	1, 1951				824	304	31:	20	200		99	4950		2.3		12100	16.3		3310	3140		16500	
							une de la	45	. SALT	FORK	RED RI	VER NORTH	OF Q	UAIL				1449					
Mar. Apr.	22, 1959 21, 1963	b0.25 b1.5		1739	355 640	98 84		68 02	169 166		1240 1800	315 570		2.0		3600	4.90			1150 1810	3.2	3050 4320	

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

				Mag		Po	Bi-	Con						AND RESIDENCE OF THE PERSONS ASSESSED.	ssolved se		Hard as C		So-	Specific con-	
Date of collection	Discharge (SiO	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (C1)	ride	Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
	100		Tien.				46. DC	ZIER	CREEK NE	AR WELLIN	NGTON										
June 14. 1950 Feb. 8. 1951 Mar. 5, 1953 Sept. 18 Nov. 25 Jan. 12, 1955	1.0 14 -23 24 -23 34 -40 24 -24 30		607 585 584 586 	84 84 82 84 	1	74 84 65 68 	144 186 110 99 		1720 1660 1660 1690 1640	75 70 78 76 73		6.8 12 8.7 9.0 9.3		2640 2600 2560 2600 2520	3.59 3.54 3.48 3.54 		1800 1790	1740 1650 1700 1730 	 0.7 .7 	2820 2780 2820 2810 2820 2720	7.7
Jan. 15, 1960							58			38			100				608	560	-11	1160	
					48	. NO	RTH FOR	RK RED	RIVER W	EST OF KE	ELLERV	ILLE						a. jaja			
Mar. 22, 1959	b6		208	112	3:	35	149		156	1010				Sec. 1			980	858	4.7	3560	7.9
					49. Me	cCLEL	LAN CRE	EK AT	STATE H	IGHWAY 70	NEAR	BOYD	STON	in the second		and Carlon					
Sept. 12, 1950							86		8.0	2.9		5.0					74			179	7.9
							50 LA	KE MC	CLELLAN	NEAR JERI	СНО										
June 28, 1951	10	0.23	23	2.8	3 1.2	1.2	83		2.7	1.7	0.1	2.2	0.27	96			69	1		148	7.6
					51. 1	McCLE	LLAN CR	EEK A	T STATE	HIGHWAY 2	73 NE	CAR Mc	LEAN		1000 Magain - 10					ales con c	
Jan. 6, 1965 June 22 Oct. 20 Nov. 22 June 8, 1966	11.4 23 12.1 29 11.2 24 10.9 24 .1 24		50 56 72 66 65	22 17 18 19 21	121	99 12 12 05 4.6	160 193 230 216 183		123 122 125 118 123	121 117 125 120 175	.8 .8 .7	0.2 .0 .2 .2 .0		886 549 590 559 624	0.70 .75 		216 210 254 244 248	52 65 67 98	2.9 3.4 3.1 2.9 3.3	929 986 961 1140	7.4 7.6
Jan. 17, 1967 July 25	2.07 25 1.00 25		78 60	20 19	124 132	3.0 4.2	168		133 136	141 171	. 8	. 0		647 631	<u> </u>		277 228	73 90	3.2	1070	7.9
Sept. 7	4.16 23		69	20	134	4.1		DODU	137	169	. 8	. 2		662			254	80	3.7	1100	7.6
						52.		FORK		ER NEAR S	10 000			1000							
Feb. 26, 1951 Apr. 9 July 25 Nov. 18 Jan. 21, 1952	162 22 12.0 32 2.19 28 .73 26 18.5 21		148 194 248 392	35 47 22 57	20 2 11	72 07 28 17	200 168 113 107		268 489 593 1040 428	315 340 41 210 308	 	1.5 1.5 2.5 1.5		1060 1390 1020 1900	1.44 1.89 1.39 2.58		514 678 710 1210	350 540 617 1120	1.5	1750 2139 1310 2530 1970	8.4
Dec. 8, 1953 Jan. 9, 1958 Feb. 13 Mar. 22, 1959 May 27, 1964	13.0 26 15.2 20 5.89 42 b6.0 - 12.9 9.		189 205 235 405	 36 45 49 44	29 23 3	02 52 10 63	178 89 159 103		445 390 530 565 1080	388 360 438 528 84	 0.5	.2 .5 1.8 		1290 1560 1740	1.75 2.12 2.37		620 696 790 1190	474 624 660 1110	3.5 4.1 4.8 .8	2320 2050 2550 2740 2010	7.5 8.0 8.0 6.5
May 28	$\begin{array}{c} .13 & 14 \\ 81.9 & 12 \\ 2080 & 18 \\ 18.3 & 7. \\ 1180 & 14 \end{array}$	5	430 164 130 218 122	55 10 21 54 20	2	52 20 95 18 55	128 104 226 76 141		1130 357 238 584 186	245 22 130 402 292	.6 .6 .6	1.0 5.3 .5 .5		2090 642 744 1520 860	2.84 .87 1.01 2.07		1300 450 411 766 387	1200 365 226 704 272	1.9 .4 2.0 3.4 3.4	2590 889 1170 2260 1510	6.5 7.1 6.8 8.1 6.8

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

					Mag		Po	Bi-	Com			170		-		ssolved s		Hard as Ca		So-	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pl
						52.	NORTH	FORK I	RED RI	VER NEAD	R SHAMROCI	KCor	ntinue	ed								
ov. 22, 1965 uly 24, 1966 ept. 19 an. 17, 1967 pr. 14	10.8 35.6 34.8 4.79 28.4	13 18 23		265 295 205 310 227	47 21 30 59 46	24 188 272 293	8.3 6.0 4.9 7.1	148		618 661 · 366 712 416	375 40 378 528 580	. 6	2.2		1600 1150 1270 2000 1680	22. 22.		855 822 635 1020 756	736 685 514 872 608	3.0 .4 3.3 3.7 4.6	2470 1480 2070 2960 2710	7
une 28 ug. 5 ug. 24	92.0 77.0 137	11		172 218 160	19 40 23	60 211 180	5.2 5.7 5.8	132		364 432 175	109 425 360	.3 .6 .6	.5 2.2 1.2		811 1420 1040	==		507 708 494	390 600 300	1.2 3.4 3.5	1170 2200 1750	7
					53.	SWEET	WATER	CREEK	AT ST	ATE HIGH	TWAY 152 W	EST O	F MOB	EETIE								
Peb. 13, 1951 Peb. 27	2.73	17 23		74 64	13 11		17 13	298 246		14 17	10 8		2.0 2.0		294 259	0.40		238 205	0		499 410	
					54.	SWEETWA	rer Ci	REEK AT	STAT	E HIGHWA	Y 152 SOU	THEAS	T OF	MOBEE	TIE							
Peb. 13, 1951 Peb. 27	8.84	18 27		68 68	14 14		41 20	308 280		26 17	24 16		3.0		368 302	0.50 .41		227 227	0		593 482	
							5	5. SWE	ETWAT	ER CREEK	NEAR WHE	ELER										
eb. 12, 1951 eb. 13eb. 27eb. 27eb. 27eb. 27		27		88 79 81 77 70	22 19 22 21 18	1	13 34 50 55 39	312 348 336 340 334		104 27 79 75 25	26 24 29 28 23	4	2.5 2.0 3.8 2.0 1.5		488 402 458 457 372	0.66 .55 .62 .62 .51		310 275 292 278 248	54 0 17 0	== == ==	718 619 718 717 589	8
uly 25	8.48 7.77 10.7 16.9 1.69	30 35 40		39 40 30 70	14 15 14 16		37 51 41 37	222 230 206 326		25 38 20 22 26	18 30 18 21 18		3.0 1.0 .8 1.0 1.5		295 318 270 362	. 40 . 43 . 37 . 49		155 162 132 240		1.7 1.6 1.0	448 560 506 426 586	8
ay 11ept. 8ec. 8an. 12, 1955pr. 24, 1957	3.83 .08 5.91 2.98 13.5	58 36 36		30 36 104	15 14 13 12		12 32 29 33	201 210 391		22 21 20 18 29	26 16 19 16 21		3.5 .5 1.0 1.0		272 286 423	. 37 . 39 . 58		137 147 308	0 0 0	1.5 1.2 .8	435 423 506 506	8
Oct. 1	.14 7.15 6.14	24		37 64 	11 13 15		33 36 2.9	200 285 		25 25 28	13 23 24		.5 .5 1.2		258 326 	.35 .44 		138 214 		1.2 1.1	385 530 526	8. 8.
							50	6. SWE	ETWAT	ER CREEK	NEAR KEI	TON										
Aug. 8, 1962 Aug. 13, 1963 Nov. 19 Jan. 21, 1964 June 11	11.9 224 10.8 15.9 7.17	12 25 27		104 74 83 88 102	21 8.6 18 18 25	;	61 18 46 45 53	267 241 287 322 274		202 44 100 87 197	32 11 28 24 28		.5 1.8 1.8		598 287 443 450 567	0.81 .39 .60 .61		346 220 281 294 358		1.4 .5 1.2 1.1	860 488 709 716 854	6. 7. 7.

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

Nov. 23 Mar. 8, 1966 June 8 July 13 Aug. 2 Sept. 12 Jan. 17, 196 Apr. 14 May 16 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 June 12 June 12 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195	964	346 (cfs) 346 .17 13.1 389 15.2 17.1 2.9 .2 1.0 .6	15 30 22 14 25 22 24 30 	(Fe)	56 175 89 44 91 96 136	7.4 46 22 5.4 19	9	Po- tas- sium (K) 6. SV 32 96 48 25	(HCO ₃)		Sulfate (SO ₄) EEK NEAR 27 528	Chloride (C1)	(F)	trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non-car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
Oct. 9 Jan. 6, 1965 Oct. 18 Nov. 23 Mar. 8, 1966 June 8 July 13 Aug. 2 Sept. 12 May 16 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 May 19 May 6 June 12 June 10, 195 Sept. 12 Sept. 12 Dec. 1 June 10, 195 Sept. 18 Sept. 28 Oct. 17 Mar. 15, 195	66 967	. 17 13.1 389 15.2 17.1 2.9 .2 1.0 .6	30 22 14 25 22 24 30 		175 89 44 91 96 136	46 22 5.4 19	3 9 4 2	32 96 18	234 196	TER CR	27	13							170		1.1	AFE	
Oct. 26	66 967	. 17 13.1 389 15.2 17.1 2.9 .2 1.0 .6	30 22 14 25 22 24 30 		175 89 44 91 96 136	46 22 5.4 19	9	96 18	196				0.4	0 0					170		1 1	455	1001237
Jan. 6, 1965 Oct. 18 Nov. 23 Mar. 8, 1966 June 8 July 13 Aug. 2 Sept. 12 Jan. 17, 196 Apr. 14 May 16 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 June 12 June 12 June 12 June 12 June 12 June 12 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195	66	13.1 389 15.2 17.1 2.9 .2 1.0 .6	22 14 25 22 24 30 		89 44 91 96 136	22 5.4 19	2	18								266	. 36		170				6.
Oct. 18 Nov. 23 Mar. 8, 1966 June 8 July 13 Aug. 2 Sept. 12 Jan. 17, 196 Apr. 14 May 16 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 May 6 June 12 June 12 July 7 7 Aug. 12 June 10, 195 Sept. 1 Sept. 28 June 10, 195 Sept. 1 Sept. 28. Oct. 17 Mar. 15, 195	967	389 15.2 17.1 2.9 .2 1.0 .6	14 25 22 24 30 		44 91 96 136	5.4 19	2				112	78 27	.6	3.0		1040 478	1.41		626 312	465 55	1.6	1430 741	7.
Nov. 23 Mar. 8, 1966 June 8 July 13 Aug. 2 Sept. 12 Jan. 17, 196 Apr. 14 May 16 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 May 9 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 July 7 Aug. 12 Dec. 1 July 7 July 7 Aug. 12 July 7 July 7 July 7 July 7 July 7 Aug. 12 July 7 Sept. 28 June 10, 195 Sept. 1 Sept. 28 Mar. 15, 195	967	15.2 17.1 2.9 .2 1.0 .6	25 22 24 30 		91 96 136	19			178		22	12	. 4	.5		211			132	0	.9		7.
June 8	967	2.9 .2 1.0 .6	24 30 		136	19		2.9	318		121	32	.8	1.0		502			306	46	1.3	798	7.
June 8	967	2.9 .2 1.0 .6	24 30 		136		51	2.3	300		127	31	.8	1.5		499			318	72	1.2	795	7.
Aug 2 Sept. 12 Jan. 17, 196 Apr. 14 May 16 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Dec. 1 Dec. 1 Dec. 1 Sept. 28 Dec. 1 Sept. 28 Oct. 21 Dec. 1 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195	967	1.0 .6 11.4	22			34	69	2.6			355	42	. 5	2.0		778			480	291	1.4	1140	7.
Sept. 12 Jan. 17, 196 Apr. 14 May 16 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195	967	.6 11.4				47	99	3.0	180		554	71	. 5	1.5		1060			608	460	1.7	1440	7.
Jan. 17, 196 Apr. 14 Aug. 28 Sept. 30, 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 May 12 May 28 May 6 June 12 June 12 June 10, 195 Sept. 18 Sept. 28 June 10, 195 Sept. 18 Sept. 28 May 6 June 10, 195 Sept. 18 Sept. 28 Oct. 17 Mar. 15, 195	967	11.4			117	21		5	210 194		433	33 55							378 526	206 367		912 1250	7.
Apr. 14 Aug. 28 Aug. 28 Aug. 28 Sept. 30. 19 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 May 6 Juc. 16 May 6 June 12 June 12 June 10, 195 Sept. 28 June 10, 195 Sept. 28 Oct. 17 Mar. 15, 195							40	0 0					7			454						715	7.
May 16		02.1			87 70	18 15	46 47	6.0	302 284		98 69	28 26	.7	1.0		454 397	151		291 236	44	1.2	637	7.
Sept. 30, 19 Oct. 26			10 E. S.			21	50	2.7	224		150	32	.8	1.8		465			276	92	1.3		7.
Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28. Oct. 17 Mar. 15, 195		1.80	26		148	33	72	2.8	218		400	45	. 7	2.5		837			505	326	1.4	1150	7.
Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28. Oct. 17 Mar. 15, 195									57.	ELM C	REEK NEA	R SHAMROO	CK										
Oct. 26 Oct. 26 Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195	1946					37		50	162		525	92	14-	2.2		1010	With the	v	681			1410	
Oct. 26 Oct. 26 Oct. 26 Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		2.54				29	10	00	241 277		403 305	98	==	5.0		923			541			1240 1120	Ē
Oct. 26 Oct. 26 Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		1.01				36	10		262		655	82 90		.0		1270			797			1620	ΠĒ
Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		.91				31		7	306		230	70				743			474			1050	-
Nov. 25 Dec. 16 Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		2.51					_	_	194		471	98		2.8						ш.		1320	-
Feb. 13, 194 Apr. 8 May 6 June 12 July 7 Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28. Oct. 17 Mar. 15, 195						31		70	260		352	94		14		998			572		112-	1320	-
Apr. 8		2.32				28		73	196		359	90	4.0	4.0		864	7.5		504		- 55	1240	-
May 6		2.42				35 28		54 31	242 248		343 342	95 92		3.8		908 891	11		576 517		II	1230 1220	_
June 12																			492			1240	_
July 7		3.3				28 31		78 38	204 264		351 310	90 83	- 11	7.4		898 853	1		579			1220	
Aug. 12 Oct. 21 Dec. 1 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		2.71										95					H 111					1210	-
Dec. 1 June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195						29		11	208		351	97		5.5		943			524			1250	
June 10, 195 Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		1.47				29	7	72	270		346	91		5.5		946			558			1270	-
Sept. 1 Sept. 28 Oct. 17 Mar. 15, 195		2.58				23		34	166		324	92	72	5.3		833	1 00		424	979		1270	
Sept. 28 Oct. 17 Mar. 15, 195		3.0 1.97				29 28		39 36	269 166		328 365	89 98		6.5		948 886	1.29		498 470	278 334		1230 1230	7.
Oct. 17 Mar. 15, 195		2.67				29		18	106		355	93		4.5		815	1.11		421	334		1120	7.
		2.39				33		52	96		354	92		4.2		801	1.09		445	366		1100	7.
	951	2.17	40		139	34	4	11	111		337	91		6.1		787	1.07		487	396		1179	8.
June 28		1.71	53			30		78	153		360	94		5.0		833	1.13		468	342		1240	7.
May 11, 1953		1.67				32		73	102 90		373 308	97 121		5.0		866	1.18		454 360	370	1.5	1200	8.
Sept. 8 Dec. 8		1.35			11				90		377	98		3.0								1300	
								78	76		393	98		2.5		863	1.17		442	380	1.6		7.
Jan. 12, 195 Jan. 9, 1958		1.67				28 25		39	248		425	100		1.5		982	1.34		594	391	1.6		7.
Mar. 22, 195	955	b2.0				33	10		161		607	112		.2		1190	1.62		697	565	1.7		7.
Feb. 21, 196 Aug. 8	955 58		25			28 27		73	274 224		370 380	103 102	0.5	3.6		998 930	1.26		600 525	376 342	1.3	1380 1280	7.

Table 6 .-- Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

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Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

							Mag-		Po-	Bi-	Car-						A CONTRACTOR OF THE PARTY OF TH	ssolved s		Hard as C		50-	Specific con-	
	Date of collection	Discharg (cfs)	e Sil		ron Fe)	Cal- cium (Ca)	ne- sium (Mg)	Sodium (Na)	tas- sium (K)	car- bon- ate (HCO ₃)	bon- ate	Sulfate (SO ₄)	Chloride (C1)	Tiue	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate		duct- ance micro- nhos at 25°C)	pН
									63.	MIDI	LE PE	ASE RIVI	ER NEAR PA	ADUCAH										
	. 11, 1950 11, 1959		22			177 1320	27 217	111	06 00	109 125	1	458 3730	152 17300		3.0		999	1.36		552 4180	463 4080		1460 44300	
		ngij				64.	SALT S	PRINGS T	RIBUTA	RY TO	MIDDL	E PEASE	RIVER 14	MILES	NORT	THEAST	OF PADI	JCAH						
	11, 1959 21)2			1430 1080	243 241	139 65		111 170		4100 3210	21700 10200							4570 3700			52000 28900	
									. 6	5. PI	EASE R	IVER NE	AR CHILDR	ESS						¥.				
Jan. Aug.	21, 1959 26, 1967 24	. 2	57 10 33 11 54 10			1200 1260 1040 705	246 229 180 115	76: 8040 5700 4110	20 18 18 15	151 144 124 108		3270 3330 2190 1730	12100 12700 9300 6600				25600 18500 13500			4000 4090 3340 2250	3880 3980 3240 2160		32600 36600 28200 20500	7.4
	12.									67. I	PEASE	RIVER NI	EAR VERNO	N									7	
Apr. Jan. Apr.	16, 1942 10, 1951 27, 1967 17	. 0.3	35 5 9	 . 5 . 9 . 1		580 500 478 385 495	115 139 132 50 41	108 16: 1360 782 1050	7.3 7.9	126 165 235 125 77		1820 1540 1560 944 1070	1610 2540 2050 1240 1640		6.7 .5 .8 2.0		5270 6410 5710 3480 4260	8.72		1920 1820 1740 1170 1180	1680 1540 1060 1120	9.9	7570 10100 8440 5310 6520	7.0
								69.	NORT	H WICH	ITA R	IVER 11	MILES SOU	TH OF	PADU	CAH								
Mar. Jan.	28, 1951 12, 1952 13, 1953 10, 1954	. 11.:	18 10 - 21			821	180 194 191 	406 438 473 482	80 30	156 150 95 128		2230 2300 2410 2460	6410 6990 7430 7660				13700 14800 15600 24000	18.6 20.1 21.2		2650 2850 277 0	2530 2720 2700 		20500 22100 22900 24000	7.7
		100						70. N	ORTH	*ICHIT	A RIVI	ER 10 MI	LES SOUTH	EAST (OF PAI	DUCAH				100	5 75			
Nov. Mar. Jan.	25, 1951 28	. 4.4 . 4.3	23 5 22 7 16 04 18			540 661	145 134 145 133	187 87 219 77 308	79 90 75	112 138 183 90 110		1860 1590 1890 1600 2110	2920 1440 3460 1250 4930		13 16		7470 4690 8450 4350	10.2 6.38 11.5 5.92		2090 1900 2240 1830	2000 1790 2100 1760	8.8 2.0 7.9	11400 6790 12800 6100 16800	7.6
July Dec. June	21, 1956 30, 1958 22, 1966 12, 1967	. 6.4 . cl200	0 6	.6 .9		1260 920 53 660	265 185 7.3 129	5720 73 3690	00 22 5.4 14	117 128 188 96 126		3870 2740 84 1880	18500 1980 9050 11 5800	0.6	 4.2		35900 18700 394 12300	47.7		4230 1280 3060 162 2180	4140 1180 2960 84 2070	2.5	27200	7.9 7.4 7.5
								7	1. SA	LT CRE	EEK 4	MILES SO	OUTHEAST (OF PAD	UCAH									
Mar. July	28, 1951 21, 1952 30, 1958 30	. 0.:	18 15 21			475	304 120 285	1230 214 1210	40	126 177 125 129		3850 1330 3890	19100 3410 18400 19700	15			37000 7580 36000			4460 1680 4260 4400	4360 1530 4160 4290		48000 12200 45700 48800	7.7 7.4

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

								Bi-							1000	ssolved se		Hard as Ca		So- dium	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pН
						72. SA	LT CI	EEK AT	MOUT	H 8 MILE	S SOUTHEA	ST OF	PADU	CAH								
ov. 16, 1939 ily 25, 1951 ov. 28 ar. 12, 1952 an. 13, 1953	2.7 2.27 2.22 1.96	7.2		546 1010 1230 1320 1270	124 271 299 319 297	39 82: 1120 1180 1170	30 00 00	186 51 95 87 91		1590 3030 3630 3730 3800	650 13000 17600 18600 18200		5.6		3400 25600 34000 35800 35300	34.3 47.6 46.9		1870 3640 4290 4600 4390	3600 4210 4530 4310		36700 45000 47000 47000	7
eb. 10, 1954 ov. 21, 1956 uly 30, 1958 uly 30 ar. 18, 1959	2.31 1.04 	14		1200 92 1300	272 15 270	116 11700 33 124	32	91 - 94 57 92 119		3890 3820 196 4050	18800 18300 7240 520 19200		 1.0		35300 1210	46.9 1.65		4110 1730 291 4360	4030 1680 216 4260	8.5	21400	7 7 7
					73.	NORTH WI	CHITA	RIVER	BELOW	SALT C	REEK 12 M	ILES S	SOUTHE	EAST (OF PADUC	АН						
ar. 12, 1952 uly 30, 1958	5.89	11		871	209	53:	20	138 108		2490	8440 4490				17400	23.7		3030 1580	2920 1490	42	25900 14300	
						-	74	. NOR	rh wic	HITA RI	VER NEAR	PADUCA	AH									
uly 30, 1958 ar. 18, 1959 ec. 7, 1965 ay 31, 1966 uly 5	4.8 2.8 2.2	10 5.0 1.2 .8		444 851 960 1040 975	94 121 220 231 207	23 53 61 6720 6300	00	106 154 164 95 72		1310 2610 2720 3100 2970	3590 8020 9680 10700 9800				7810 19800 21900 20300	10.6		1490 2620 3300 3560 3290	1410 2490 3170 3470 3220		12200 23800 29700 30200 29200	7 6
		1011192					75	. NOR	TH WIC	HITA RI	VER NEAR	TRUSCO	TTC									
eb. 10, 1954 ov. 21, 1956 ar. 19, 1959 ecc. 7, 1965 une 1, 1966 uly 6 ept. 16 ecc. 21 ug. 23, 1967	13.7 7.0 11.7 4.1 1.9 c2740 14.5 11.8	8.4 2.7 3.8 6.5 8.4 3.4 2.8	100	941 911 853 961 774 130 830 910	216 218 231 226 167 11 192 190	38 44' 444 34: 4060 2860 58 3470 3440	70 30	128 116 114 147 145 142 0 154 160 93		2590 2820 2890 2480 2770 2190 245 2580 2750	6360 7050 6930 5630 6620 4480 79 5500 5370	0.2	3.0		15600 12700 14800 10600 615 12700 12800	21.2		3240 3170 3080 3330 2620 370 2860 3060	3140 3080 2960 3210 2500 244 2730 2980	34 34 1.3	20800 21700 21500 19700 23100 15400 955 18400 18500	8 7 7 7 7
							7	6. SOI	TH WI	CHITA R	IVER AT G	UTHRI	Ξ		100						430.78	
Sept. 11, 1950 July 30, 1958 Jar. 19, 1959 Jpr. 20, 1963	5 b.04 b.25			228 570 759 675	64 227 296 260	3' 14 18 15	60	103 103 125 166		686 2020 2720 2470	605 2350 2990 2400		1.0		2040 6680 7430	2.77 9.08 10.1		832 2360 3110 2750	748 2270 3010 2620	13 14 13	3140 9530 11700 9880	7.
						78.	sou	TH WICH	HITA F	IVER 6.	5 MILES EA	AST OF	GUTH	RIE				100	eric.		1111,00	
Jan. 13, 1953 Feb. 10, 1954 Nov. 21, 1956 July 30, 1958 Mar. 19, 1959		22 15		1210 1190 833 1240	305 294 178 269	88 8000 8870 48 95	30	117 - 116 - 113 121 131		3090 3000 3140 2220 3290	14300 13300 14400 7730 15200				27800 27900 15900	37.1 37.3 21.6		4280 4180 2810 4200	4180 4090 2710 4090		38500 36800 36800 23000 39200	7. 7. 7.

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

								Bi-								solved s		Hardi as Ca		So-	Specific	
Date of collection	Discharge (cfs)	Silica (SiO ₂)		Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pН
							79.	SOUT	H WICH	ITA RIV	ER NEAR B	ENJAM	IN									
July 29, 1949 Jan. 13, 1953 Feb. 10, 1954 Nov. 21, 1956 Mar. 19, 1959	10 4.0 5.90 1.65	24 8.7		1420 1250 1420	394 303 394	889 662 6330 833	0	110 106 118 136		3410 3040 3000 3750	6080 14800 11300 10700 13600		 		28900 21700			5160 4360 5150	5060 4260 5040	 41	20100 39900 32200 28900 36300	7.' 8.
Dec. 7, 1965 June 1, 1966 July 6 Sept. 16 Jan. 25, 1967	10.2 1.7 .4 c3710 8.78	5.0 5.7 8.3 6.7 2.1		1030 1360 999 250 1050	289 396 193 18 275	449 6320 2840 184 4920	68 21 5.1 2 2	182 172 133 104 177		2690 3600 2310 546 2940	7500 10600 4820 305 7900	0.2	 0.2		16100 22500 11300 1370 17200	== == ==		3760 5020 3290 698 3750	3610 4880 3180 613 3610	3.0	2070	7.
June 13 Aug. 23	26.0 4.48	8.8		480 1140	62 279	605 4320	10 22	88 154		1170 3070	1060 7140		1.5		3440 16100			1450 3990	1380 3860		5010 23300	
							8	0. WI	СНІТА	RIVER N	EAR SEYMO	OUR										
Jan. 13, 1953 Feb. 10, 1954 July 30, 1958 Dec. 7, 1965 July 7, 1966 Sept. 1 Dec. 20 June 14, 1967 Aug. 23	8.0 17.6 44.8 8.5 2310 906 21.2	18 14 5.7 9.3 10 6.3 9.8 9.3		792 500 799 540 179 815 146 568	207 88 216 127 17 212 18 125	358 3490 123 292 1830 112 3150 140 1760	13 4. 18	85 102 75 167 140 8 129 166 7 110 127		2320 2240 1190 2030 1470 370 2320 362 1520	5710 5860 2120 4950 2920 173 5150 200 2900	0.3	 0.5 .5		12600 5180 11000 6980 921 11800 936 6950	7.04		2830 1610 2880 1870 921 2900 438 1930	2760 1550 2740 1760 388 2770 348 1829	13 18 2.2 2.9	18400 19100 8020 17000 10500 1430 17100 460 10500	7. 7. 7. 7. 7.
								81.	LAKE K	EMP NEA	R MABELLE	3					7 55 7 7 54 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					
Oct. 10, 1939 July 15, 1942 June 6, 1946 June 16, 1952 June 15, 1954 Nov. 23, 1955		7.2 7.0 7.4 4.7 6.5	0.09 .02 .02	212 198 282 240 140	44 44 65 57 28	762		86 116 104 106 6 78		594 529 774 675 373 387	685 610 1250 1100 520 470	0.3 .6 .4 .4	0.0 3.0 .8 .0 .5	0.31	2020 1824 3210 2830 1430 1380	1.94		710 675 972 834 464 468	746 400	6.5	2970 5230 4650 2430 2290	7.5
Oct. 23, 1964 June 30, 1965 Aug. 31 Dec. 7 Jan. 23, 1967		6.7 6.4 6.6 7.3	 	248 225 225 270 210	61 54 52 60 42	7777	94 14 09 40	106 116 88 132 2 102		716 648 652 740 560	1250 1110 1110 1170 820	=======================================	1.5 1.5 1.5 .5	=======================================	3130 2820 2800 3050 2210	4.26 3.84		870 784 776 920 696	783 688 704 812 613	11 11	4910 4660 4650 4890 3620	6.
				7-70					CHITA		EAR MABEL											
Dec. 12, 1965 June 1, 1966 June 2 July 7 Oct. 11	128 12.0 328 461 1740	6.1 7.3 5.4 5.6 6.3		220 603 245 250 189	46 162 49 52 35	1980 656 694 460	7. 6.	9 109 0 109 0 86		616 1670 640 672 512	898 3340 1080 1110 720		0.0 .5 1.5 1.2		2410 7850 2740 2850 1970			749 2170 813 838 616		10 10 8.0	4550 4650 3250	7.1 7.2 6.0
Dec. 20 July 21, 1967 Aug. 22	150 1340 399	6.8 6.0 6.2		213 208 204	42 40 39	526 512 477	5.	9 103 9 108 0 110		562 556 534	830 810 760		1.0		2240 2190 2080			704 684 670	595 580	8.5	3690 3540 3280	

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued
(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

									Bi-								ssolved s		Hard as Ca		So-	Specific	2
	Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	20 0399.00
		and Hilling				700		8	33. SA	NTA R	OSA LAKE	NEAR VER	NON									27H	
	5, 1966 21		6.0 9.5		34 32	7.2 7.2		3.8 4.3	138 130		22 19	5.3 7.7	0.2			159 155			114 109	1 3	0.5	281 265	
									84. B	EAVER	CREEK N	EAR ELECT	RA			100 Per 100 Pe			17.54				i i
	26, 1966	103	7.8		50	12	128	2.9			11	243	0.3	2.2	77	522		Torth 199	174	66	4.2	1020	7
	27	.7 2.10	7.2		290	135	935	7.0			54	1500 2200				3730			700 1280	626 1110	11	4760 6830	7
an.	27, 1967	3.2	5.0		330	168	1170	7.2	200		38	2750				4560			1510	1350 1290		8120	
eb.	28	. 39	1.0		330	156	1110	7.4			85	2600				4400		(d)	1460	1290	13	7740	'
33			23							BUFFAI		NEAR IOW											
	3, 1961 19, 1962	0.15 309	7.9		160	50 6.0		722 32	67 54		57 8.4	1460 64	0.3	0.0		2490 166	3.39 .23		604 75	550 30	13	4660 324	6
	19	167	9.6		21	6.0		30	68		9.8	52	. 3	1.5		163	. 22		77	21	1.5	309	6
	19	2.13	5.3		20 425	6.0	1870	6.6	91		12 137	31 3900	. 3	.8		163 6510	. 22 8.85		75 1660	1630		271 10900	6 5
	1 1988		ender.					86	. BUF	FALO (CREEK NE	AR IOWA P	ARK						1325.0	1169	139	7 1000	
eb.	12, 1964	0.16	6.4		48	18		188	88		23	355	0.3	6.2		688	0.94		194	122	5.9	1340	7
ay 4	1	.01	8.4		93 36	33 12		813	139 110		40 14	632 155	.3			1190 367	1.62		368 140	254 50	$7.1 \\ 3.1$	2270 711	6
	23	39.1			41	13		111	120		15	199	.3			448	.61		156	58	3.6	866	6
an.	26, 1965	.36	7.7		96	73		110	324		110	740	. 7			1600	2.18		540	274	7.7	2840	7
	23 12, 1967	1000	9.4		68 18	30 5.0		4.8	164 37		30 31	430 56	.3			872 177	1.19		293 65	158 35	5.6	1680 331	6
P.								87.		ITA R		WICHITA F				Page 1							
	12, 1951 14, 1966	354 1550	12 6.5	T II	230 195	61 40	480	6.4	113		629 514	1000 770		2.0		2600 2060	3.54	and a second	825 651	732 568	9.3	4320 3480	
π.	14, 1900	1330	0.0		193	40	100	88.		WICH		ICHITA FA	LLS	1.0		2000	1110		001	300	0.2	3400	_
o.t.	19, 1944		3.0	0.10	152	41		107	74		264	775	0.9	1.0		1680		2000 1 1400 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000 1000	548		100	3240	7
ov.	16		2.0	. 25	172	41	4	124	66		294	820	.9	1.5		1790			598			3370	7
	6, 1946 24, 1952		8.6	. 67	81 120	25 36	280 304	21	80		101 239	545 552	.6		0.35	1100 1310	1.78		305 448	240 362	6.2	2070 2440	7 7
	15, 1954		6.6	.37	34	8.8		5.7			36	109	. 5				. 43		121	36	2.6	592	7
	11, 1959		5.0	11	101 120	24 33		259 363	118 134		175 270	450 590	. 3	1.0		1070 1450	1.46		350 435	254 325	6.0	1890 2570	7 6
une	23, 1965		8.2		120		89			IVER		ROAD 171				1400	1.51	3000	100	323		2010)	
	21 1040	1.28 Mg 1.2	10		200	77				T T DIC 1					1.15	2540	3.45		816	696		4350	8
	21, 1949 12, 1951		10		200 248	84		615 670	146 226		486 573	1080 1160		4.4	1.15	2850	3.45		964	780	9.4	4720	7
	29, 1958							44	116			610		164_					545	450		2750	7

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

								Bi-							Residence of the second	ssolved a		Hard as Ca	THE RESERVE OF THE PARTY OF THE	So-	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)		Cal- cium (Ca)	mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (C1)		Ni- trate (NO ₃)		Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	ad- sorp- tion ratio	ance (micro-	рН
							90.	LAKE	KICK.	APOO NEA	R ARCHER	CITY										
Oct. 21, 1946 Feb. 20, 1952 June 15, 1954 Jan. 23, 1957 Oct. 23, 1964		8.2 4.3 3.1 4.0 5.7	0.25 .00 .04 .07	16 33 24 32 28	4.4 11 8.4 9.8 9.3	3	 13 16	176 139 188 159		4.8 8.4 7.7 9.8	18 15 19 31	.5 .4 .4	1.8 .5 2.8 .2 .2	0.07	115 197 153 209 200	0.27 .21 .27 .27		58 128 94 120 108	0 0 0 0		214 335 274 389 371	7.0 7.9 7.6 8.2 7.2
June 23, 1965 Aug. 30 Dec. 7 Jan. 27, 1967 May 10		5.2 9.0 9.4 4.5 5.0	===	26 34 30 29 31	8.8 9.0 10 8.0 8.0	4	19 14 15 4.5 4.5	151 181 176 5 153 5 162		13 13 14 10 10	33 38 38 30 32	. 4 . 4 . 3 . 6	.2 .0 .2 .5	===	200 237 234 194 205	.27		101 122 116 105 110			379 432 421 356 370	6.9 7.1 7.3 7.5 8.0
								92. L	AKE CI	REEK NEA	R HENRIET	TA										
Apr. 18, 1959 Apr. 21	b5 b.5	9.0 14		69 77	19 20	36 50		88 110		34 46	660 870	0.7	0.2		1200 1590	1.63 2.16		250 274	178 184		2220 2920	
						94.	DRY I	FORK LI	TTLE V	VICHITA	RIVER NEA	R HEN	RIETT	A			34 1984					
Oct. 4, 1959 Oct. 7		12 18		9.1 16	1.8 3.7		7.8 8	44 78		0.8 5.8		0.1	0.8		61 117	0.08		30 54		0.6	87 182	7.6
						95.	EAST	FORK L	ITTLE	WICHITA	RIVER NE	AR HE	NRIET	TA				3.634				
Oct. 4, 1959 Oct. 7 Mar. 10, 1964 Apr. 28 Apr. 30		12		6.5 14 22 14 16	1.9 3.4 5.1 3.9 4.9	1 2 1	8.9 8 3 4	34 66 104 60 72		0.8 3.2 10 6.8 9.2	22 20 16	0.0 .2 .3 .2	0.8 .5 .8 2.0 1.5		57 109 145 99 112	0.08 .15 .20 .13 .15		24 50 76 51 60	0 0 0 2 1	9.8 1.1 1.1 .9	85 176 258 181 206	7.3 8.1 6.7 6.2 6.3
May 8	302 641 .10 .32 909			15 10 24 12 5.0	3.5 2.9 6.8 3.9 2.2	1 2	26 .5 .2 4 . 5	40 43 109 64 5 22		4.8 6.2 12 5.8 6.6	19 26 7.9	.0 .1 .2 .3	2.8 .5 .5 .8 .5		129 84 160 84 42	.18 .11 .22 .11		52 37 88 46 22	19 2 0 0 4	1.6 1.1 1.1 .8 .3	257 146 288 141 66	6.6 6.6 6.5 6.1
Apr. 27	2730 54.0 .5 564	10 6.8 12 14 8.6		11 6.5 24 62 7.5	6.0	10 4.8 26 80 6.7	5.3 3.9 4.2 6.2 6.0	28 82		5.0 1.8 10 27 4.4	6.9 48 150	.3 .2 .1 .3	.5 .2 .2 .5 2.5		83 47 170 465 66			43 25 85 232 31	5 2 17 56 3	.7 .4 1.2 2.3 .5	145 91 313 862 105	7.6
						98.	FAR	MERS C	REEK F	ESERVOI	R NEAR NO	CONA										
Apr. 12, 1967		3.4		50	13	38		3 164		28	74		0.2		294	11.22		178	44	1.2	552	7.3
						100.			VER A		OAD 2564			ON								
June 16, 1965 Sept. 6 June 7, 1966 Jan. 17, 1967 June 21	15.0 .64 .4 .65	21		17 89 65 76 80	19 20 23 20 23			307 332 256 5 273 5 328		61 86 85 81	35 43 49 35 45	0.7 .7 .5 .8	1.2		425 488 415 476	0.58		270 304 256 272 294	18 32 46 48 25	1.2 1.4 1.1 1.4	686 786 736 668 755	

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

					Was		Po-	Bi-	Car-		Topic I					solved a		Hard as Ca		So- dium	Specific con-	
	Date of collection	Discharge (cfs)	Silica (SiO ₂)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	tas- sium (K)	car- bon- ate (HCO ₃)	bon- ate	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pH
							103	. BOI	S D'AI	RC CREEK	NEAR RAN	DOLPH			, e 1/2	Lens			N.			
Mar. Feb. Apr.	30, 1965 30, 1966 22, 1967 4	1.1 6.1 2.66 5.70 39.0	4.4	72 36 62 70 81	2.7 2.8 3.1 2.9 2.5	41 24 56 18 20	3.2 1.5 2.5 2.1 2.6	109 254 210		27 41 46 35 32	23 16 24 11 12	0.5 .5 .3 .6	0.2 .0 .2 .0 5.0		310 179 321 247 287			190 101 168 187 212	12 0 14 10	1.3 1.0 1.9 .6	580 351 550 431 486	7. 7. 7. 7.
	12	. 45 4660	5.6 9.2	53 70	3.1	36	2.4			44 5.0	24	. 6	. 8		255 200			145 179	0	1.3	438 344	7.
				11		100	1	04. S	ANDER	S CREEK	NEAR CHIC	OTA	7 - 12°		4.					100		
Feb. Apr. May	29, 1961 6, 1962 22 25 29, 1965	1.4	11	30 73 84 97 80	2.4 6.3 7.7 7.6 23	. 2	3.4 29 12 88	90 178 201 254 254		22 82 104 93 77	5.5 28 42 . 36 127	0.3 .2 .3 .3	1.0 .0 .0 .0		124 334 398 422 532	0.17 .45 .54 .57		85 208 241 274 296	11 62 76 66 88	0.4 .9 1.2 1.0 1.8	218 515 591 679 948	6.: 7.: 7.: 7.:
June Feb. Apr.	14, 1966 13 25, 1967 7 6	6.2 2.0 .23 9.71 464	17	33 66 72 34 22	6.6 11 23 5.3 2.7	23 42 72 17 5.0	2.8 3.9 9.5 3.7 3.2	193 244 102		46 58 91 32	39 64 110 20 4.9	.3 .2 .2 .4	.2 .0 .5 .5		198 353 515 172 95			110 210 274 107 66	56 52 74 23 6	1.0 1.3 1.9 .7	877	6. 7. 6. 7.
July	11	390	5.1	22	2.3	4.7	3.1	73		9.2	4.5	. 4	2.0		. 89			64	5	. 3	148	7.
								105.	RED R	IVER AT	ARTHUR CI	TY	: <u>:</u>		19 July 10 P					3.4	100	
Dec. Jan. Feb.	1, 1961 13 4, 1962 2	2560 16100 3540 5000 c4820		114 42 81 98 62	51 10 34 26 18	222 69 188 198 105		176 86 134 144 120		258 75 192 218 128	400 106 310 310 160		2.0 .1 1.1 .6 .2		1220 374 980 1020 608	1.66 .51 1.33 1.39 .83	8430 16260 9370 13770 7910	495 148 340 352 228	350 78 230 234 130	4.3 2.5 4.4 4.6 3.0	605 1550 1580	8.8.8.
May I June Aug.	2 2 19 15	13800 7180 32400 4820 10800		36 50 109 105 36	6.8 10 30 32 4.4	33 45 237 210 30		108 144 138 158 108		43 60 262 238 32	38 59 372 335 35		2.3 1.3 .2 1.0 1.1		253 351 1180 1100 226	.34 .48 1.60 1.50	9430 6800 103200 14320 6590	118 166 395 395 108	30 48 282 266 20	1.3 1.5 5.2 4.6 1.2	1850 1730	8. 8. 8.
Nov. Dec. Jan.	10	4650 4140 24200 4820 689		94	31 	193 175 143 187 116		136 146 120 154 236		225 125 210 122	308 215 295 175		.0 		1010 848 566 936 657	1.37 1.15 .77 1.27	12680 9480 36980 12180 1220	360 320 222 355 315	248 200 124 229 121	4.4 4.2 4.2 4.3 2.8	1350 936 1510	7. 8. 8. 8.
Apr. May June	26	5580 1640 1640 1560 1780				39 125 175 191 123		124 122 128 152 156		56 140 195 215 128	70 182 260 290 182		==		314 682 945 974 642	.43 .93 1.28 1.32 .87	4730 3020 4180 4100 3080	174 230 295 340 250	72 130 190 219 122	1.3 3.6 4.4 4.5 3.4	1040 1390 1500	8. 8. 8.
	12	2340 2180				210 235		140 144		225 246	322 370		==		1050 1130	1.43 1.54	6630 6650	345 386	230 268	4.9 5.2		8.
	7		1000	Managara (1987)	HIS SHOULD	16.795	AUL	106.	LAKE	CROOK N	EAR PARIS						100 100 100 100 100 100 100 100 100 100	100				
Mar.	18, 1960		3.2	14	1.3	6.3	2.9	35		19	6.0	0.3	0.2		70	0.10	Na Maria	40	12	0.4	118	6.

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basin, Texas--Continued

(Results in milligrams	per liter except as indicated.	Calculated values for sodium n	us potassium are centered	hetween the two columns.)

								Bi-	9						ssolved a		Hard as Ca	ness aCO ₃	So-	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
							10'	7. BIG	PINE	CREEK 1	NEAR MANCI	HESTE	R								
May 25, 1962 Apr. 21 Mar. 27, 1963	2.0 23.9 500	13 9.6 5.8		23 20 10	5.3 4.4 2.5	2 1 8.1		70 42 20		36 47 25	20 17 10	0.2 .2 .1	. 0	158 150 75	0.21 .20 .10		79 68 35	22 34 19	1.0 .9 .6	255 218 122	6.8
			-				1	08. PI	ECAN E	AYOU NE	AR CLARKS	VILLE									
Jan. 17, 1967 Feb. 25 Apr. 8 Apr. 27	0.2 6.1 .4 616	6.1 1.1 4.5 6.1		24 16 29 5.	2.7 2.8 3.7 8 1.5		4.8 5.0 4.7 2.9	50 44 105 20		36 23 14 7.2	21 16 15 5.4	0.0 .1 .1 .0	.2	135 98 138 44			71 51 88 21	30 15 2 4	0.8 .7 .7 .4	237 181 250 72	6.8
							10	09. RI	ED RIV	ER NEAR	NEW BOST	ON			111100	1000					
Nov. 1, 1960 Nov. 12 Dec. 23 Jan. 31, 1961 Mar. 11				101 106 94 110 55	30 33 21 26 14	225 215 184 205 96		122 140 110 144 114		249 249 187 219 99	358 350 308 340 145		1.3 .8 1.2 1.1 1.6	1110 1130 849 972 493	1.51 1.54 1.15 1.32 .67		375 400 320 380 194	275 286 230 262 100	5.1 4.7 4.5 4.6 3.0	1740 1760 1520 1700 822	7.0 7.0
Apr. 12				82 27 54 122 95	20 5.5 13 28 30	159 24 98 228 227		122 72 90 188 116		172 29 110 242 240	250 36 150 360 360		1.9 2.6 2.3 1.3	796 180 524 1200 1070	1.08 .24 .71 1.63 1.46		288 90 190 420 360	188 31 116 266 265	4.1 1.1 3.1 4.8 5.2	1280 294 836 1820 1760	8.2
Sept. 27				98 101 79 115 88	31 30 22 32 32	226 203 176 232 187		134 120 108 168 122		240 225 185 205 200	355 342 275 385 318		1.0 1.0 1.0 12	1090 996 834 1180 1000	1.48 1.35 1.13 1.60 1.36		370 375 288 420 350	260 276 200 282 250	5.1 4.6 4.5 4.7 4.4	1770 1660 1380 1900 1580	7.8 7.8 8.3
Jan. 4, 1962 Feb. 2 Mar. 2 Apr. 3 May 2				85 58 24 28 26	27 · 13 · 5.8 · 3.9 10	163 98 38 23 27		142 100 76 84 100		178 110 33 28 37	265 150 49 26 32		1.0 .7 .4 2.6	884 544 166 198 238	1.20 .74 .23 .27 .32		322 196 84 86 108	206 114 22 17 26	4.0 3.0 1.8 1.1	1400 857 277 280 355	8.6 7.8 7.8
June 19				111 101 31 	31 31 5.5 	237 210 23 172 142		136 150 102 124 130		255 238 28 192 155	385 330 25 275 220		.0 .7 1.4 .0	1220 1110 195 907 742	1.66 1.51 .27 1.23 1.01		405 380 100 315 270	294 257 16 213 163	5.1 4.7 1.0 4.2 3.8	1830 1730 296 1420 1170	7.9
Dec. 5				 	 	96 103 164 113 23		126 148 144 208 88		111 116 191 124 31	146 158 260 174 32		 	543 593 868 687 208	.74 .81 1.18 .93 .28		216 242 326 300 100	113 121 208 130 28	2.9 2.9 4.0 2.8 1.0	879 961 1390 1090 306	8.0 7.9 8.1
Apr. 24				=======================================		144 171 192 64 204		148 148 142 92 152		155 205 225 40 228	240 270 295 120 320		== == == == ==	802 922 992 380 1030	1.09 1.25 1.35 .52 1.40		308 344 359 148 370	187 223 234 73 245	3.6 4.0 4.5 2.3 4.6	1280 1450 1540 620 1640	8.3

Table 6.--Chemical Analyses of Water From Streams and Reservoirs at Sites Other Than Daily Stations in the Red River Basın, Texas--Continued (Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

					36		D-	Bi-	0							solved s		Hard as Ca		S0-	Specific	c
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	sorp-	ance (micro-	pE
							1	10. B	ARKMAN	CREEK 1	NEAR LEAR	Y										
May 23, 1961	2.3	5.7	0.00	6.4	2.4	11	2.1	35		3.8	9.0	0.2	1.0		59			26	0		108	-
may 20, 1901												0.0										h.
	. 6	11		7.5	2.2		13	33		6.0	12	1.2	2.8		72	0.10			50	1.1		6.
Oct. 23 Nov. 30	4.4	11		7.5	2.2		13 14	33 25		6.0 8.8								28 17	50 64	1.1	143 118	6.
Oct. 23 Nov. 30 Mar. 16, 1962		11 8.3			1.5						12	1.2	2.8		72	0.10		28			143	6.
Oct. 23	4.4	11		4.5	1.5	8.0	14	25		8.8	12 12	1.2	2.8 .8 .0		72 65	0.10		28 17	64	1.5	143 118	6. 5.
Oct. 23 Nov. 30 Mar. 16, 1962 May 26	$\frac{4.4}{18.2}$	11 8.3 9.8		4.5 3.5	1.5 1.1 2.4	8.0	.7	25 16		8.8 7.2	12 12 7.5	1.2 .2 .1 .1	2.8 .8 .0		72 65 44 72	0.10 .09 .06 .10		28 17 13 30	64 55 49	1.5 1.0 1.0	143 118 73 131	6. 5. 6.
May 25, 1901 Nov. 30 Mar. 16, 1962 July 7 July 31	4.4 18.2 .5	11 8.3 9.8		4.5 3.5 8.0	1.5 1.1 2.4	8.0	14 .7 13	25 16 44		8.8 7.2 2.8	12 12 7.5 14	1.2 .2 .1 .1	2.8 .8 .0		72 65 44	0.10 .09 .06		28 17 13	64 55	$\frac{1.5}{1.0}$	143 118 73	6. 5.

a Includes the equivalent of any carbonate $({\rm CO_3})$ present.

b Field estimate.

c Mean discharge.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma

								Bi-						Dia	ssolved s	olids	Hard as C	THE PARTY OF THE P	So-	Specific con-	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
						Α.	SAL	r FORK	RED F	IVER NE	AR VINSON	, OKL	Α.								
ec. 21, 1959 an. 14, 1960 an. 26 ar. 14 pr. 18				220 250 210 290 460	56 65 67 92 105	134 156 146 161 234		70 130 86 64 144	0 0 0 0	719 820 727 1050 1540	185 190 202 215 260	.7	3.8 2.3	1350 1660 1400 1970 2850	1.84 2.26 1.90 2.68 3.88		780 890 800 1100 1580	722 784 730 1050 1460	2.1 2.3 2.2 2.1 2.6	1880 2050 1970 2380 3230	8 8 8 7 7
pr. 28				112 464 404 440 320	49 122 93 93 51	100 269 201 180 181		152 204 124 180 138	0 0 0 0	402 1600 1360 1310 969	108 292 220 255 200	1.0	2.8 .0 .2 .1	950 3020 2540 2550 1890	1.29 4.11 3.45 3.47 2.57		480 1660 1390 1480 1010	356 1490 1290 1330 897	2.0 2.9 2.3 2.0 2.5	1270 3440 2820 2940 2270	8 7 8 7
et. 18				128 332 344 296 448	20 73 78 83 88	49 191 200 166 206		232 148 166 70 152	0 0 0 0	237 1090 1120 1050 1430	49 205 220 205 222	=======================================	.1 2.8 1.4 2.9	645 2110 2180 2020 2680	.88 2.87 2.96 2.75 3.64		400 1130 1180 1080 1480	210 1010 1040 1020 1360	1.1 2.5 2.5 2.2 2.3		77 77 88 77
pr. 19 ay 17une 21uly 12uly 12				412 160 388 232 352	105 73 78 44 78	205 191 157 107 146		150 402 124 128 124	0 0 0 0	1440 644 1210 670 1130	200 82 192 135 170	=======================================	2.4 .2 .2 .5	2640 1410 2370 1340 2050	3.59 1.92 3.22 1.82 2.79		1460 700 1290 760 1200	1340 370 1190 655 1100	2.3 3.1 1.9 1.7	2870 1900 2620 1710 2380	7 7 7 8 7
ept. 26				476 504 	100 103 	161 188 177 196 173		116 120 142 60 132	0 0 0 0	1510 1620 1180 1160 968	200 215 212 225 190	=======================================	1.9 .0 1.0 2.9 2.4	2660 2800 2240 2120 1860	3.62 3.81 3.05 2.88 2.53		1600 1680 1260 1150 1010	1500 1580 1140 1100 902	1.8 2.0 2.2 2.5 2.4	2860 3070 2630 2520 2280	
ar. 7ar. 28pr. 25ay 23une 27ept. 25				=======================================	=======================================	193 213 203 182 204 109		96 92 62 166 122 140	0 0 0 0 0	1250 1380 1370 772 626 1020	240 220 230 220 295 170]]]]]]	1.2	2420 2530 2550 1680 1370 2020	3.29 3.44 3.47 2.28 1.86 2.75		1300 1360 1360 855 725 1180	1220 1280 1310 719 625 1070	2.3 2.5 2.4 2.7 3.3 1.4	2850 2860 2870 2160 1710 2480	
						в.	SAL		RED I		MANGUM,	OKLA.									
ater year 1947														322	2.00						
Maximum, Aug. 11-20, 1947 Minimum, Oct.7-9,				488	158	355		118	0	1690	555		0.5	3300	4.49		1870	1770		4080	
12. 14, 1946 ater year 1948 Maximum,	1360			170	36	86		134		485	100		3.0	946	1.29	3470	572	462		1350	
Apr. 21-30, 1947 Minimum,	6.68			563	135	198		123		1840	272		2.0	3080	4.19	56	1970	1870		3500	
June 21-22,29-30 or. 7, 1949 eb. 10, 1950 ec. 2 eb. 6, 1951	1783 			243 350 432 459 401	36 83 92 97 86	83 175 152 169 162		108 136 141 161 167		687 1110 1290 1410 1240	93 232 232 220 195		1.9 1.0 1.0 2.3 3.0	1200 2020 2270 2610 2360	1.63 3.09 3.55 3.21	5780 	754 1220 1460 1540 1350	1100 1340 1410 1220		1520 2570 2870 2920 2750	

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

									Bi-							Dia	solved s	olids	Hard as Ca		So-	Specific con-	
	Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Fluo- ride (F)	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	рН
					120	В.	SALT :	FORK I	RED RIV	ER AT	MANGUM,	OKLA C	Contin	ued								Me and	
lan. leb. lune	19 16, 1952 10, 1954 3, 1955		== == ==	 	196 389 517 126 216	38 88 90 91 102	1; 1:	33 39 91 56 79	136 149 123 255 116		538 	109 212 252 70 210		0.8	10 	1220 	1.66	12 12 13	645 1330 1660 690 960	534 1210 1560 480 865	2.0 .9 2.5	1480 2610 3250 1560 2610	8.0 7.7 7.8 7.0 7.1
oct. oct. oec.	3	2160 - 2 45 300 b407	18 	=======================================	520 224 276	17 74 49 64 66	131 125 149 156	7.8	216 144 132 144 114	0 0 0	1440 677 846 933	18 188 152 192 205	0.5 .6 .6	1.2 4.3 3.6 4.2	0.42	2440 1330 1600 1720	3.32 1.77 2.18 2.34	1.32 158 1300 1890	620 1600 760 950 1020	443 1480 652 832 926	.1 1.4 2.0 2.1 2.1	1090 2830 1740 2090 2280	7.1 7.6 8.1 8.2 8.1
eb. lpr. lay	25	120 2.0 100 26 14	20 14 17	0.00	408	125 142 71 64 45	192 213 150 224 129		220 124 118 118 104	0 0 0 0	1580 1870 1020 1320 619	260 252 182 210 152	1.0 .9 .7 	3.7 .6 .1 .2 .6	 . 41 . 24	2400	3.84 4.46 2.65 3.26 1.70	914 17.7 526 168 47.2	1780 1940 1090 1280 665	1600 1840 994 1180 580	2.0 2.1 2.0 2.7 2.2	3290 3540 2280 2740 1660	7.8 7.5 7.4 8.2 7.8
uly ept ct.	19 13 11 17	11 b10 12 1890 100	 17 	=======================================	272	56 60 88 34 92	159 110 180 58 196		112 140 98 118 152	0 0 0 2	1030 784 1510 387 1260	178 165 210 78 190	.6	$\begin{array}{c} .2 \\ .7 \\ 1.0 \\ 3.7 \\ 1.4 \end{array}$. 47		2.65 2.12 3.69 1.15 3.13	57.9 42.1 87.8 4300 621	1070 925 1560 490 1280	978 810 1480 390 1160	2.1 1.6 2.0 1.1 2.4	2270 1970 2920 1120 2570	8.3 7.7 8.3 8.0
lan. lan. leb.	16, 1961 17 13 13	79 79 84 30 15	18 	=======================================	376	71 93 73 86 112	199 167 169 222 268		174 164 188 148 182	0 0 0 0	1190 1200 965 1550 1170	205 210 190 210 415	.6 	2.4 1.6 .1 2.5	.30	2280 1910	3.13 3.10 2.60 3.75 3.58	490 486 433 224 106	1240 1320 1060 1550 1370	1100 1190 906 1430 1220	2.5 2.0 2.3 2.5 3.1	2590 2610 2300 3010 3160	8.1 7.3 8.6 8.2
pr. June July	19	14 37 99 .5 6.8	16 		444 364 204 174	122 76 39 57	187 152 84 79 158		144 120 212 132 112	0 0 0 0	1550 1130 530 560 1360	200 195 90 106 222	.6 	2.0 .2 .0 .8 2.5	. 75 	2260 1120 1160	3.70 3.07 1.52 1.58 3.43	103 226 299 1.57 46.3	1610 1220 670 670 1480	1490 1120 497 562 1390	2.0 1.9 1.4 1.3	3000 2530 1440 1470 2850	7.7 7.6 7.6 7.8
ec an. lar.	4	52 75 b32 b2.5 b3.6				 	175 146 208 296 180		118 82 64 108 104	0 0 0 0	1140 887 1300 1940 1590	210 155 260 280 220	 	1.3 2.8 1.9	 	2140 1670 2560 3220 2840	2.91 2.27 3.48 4.38 3.86	300 338 221 21.7 27.6	1200 895 1320 1860 1660	1100 828 1270 1770 1580	2.2 2.1 2.5 3.0 1.9	2520 2070 2910 3500 3100	
lay June	22 22 27 24	ь28 ь84 30	==	==		==	236 124 185		128 180 148	0 0 0	1280 720 850	280 150 245	==	==	 	2500 1460 1610	3.40 1.99 2.19	189 331 130	1320 840 950	1220 692 828	2.8 1.9 2.6	2910 1850 1940	
							С.	NORT	H FORK	RED R	IVER NE	AR CARTER	, OKL	١.		21.151			Silvini	ja sa	375		
Dec. Dec. Peb.	28, 1959 10 21 25, 1960 24	b12 32.9 161 250 37	14 21 14	 0.01 .00		95 96 38 121 101	359 279 147 260 252		232 212 178 368 154	0 0 2 0 0	1060 919 371 893 1030	500 388 205 350 310	0.5 .6 1.0 .7	0.7 .6 5.8 10 1.0	0.41 .23 .52	2190 998 2210	3.37 2.98 1.36 3.01 3.05	80.4 194 434 1490 224	1220 1070 510 1170 1090	1030 896 360 868 964	4.5 3.7 2.8 3.3 3.3	3490 2910 1500 2890 2750	7.8 8.0 8.3 7.9 7.3
July Oct. Jan.	12 11 17, 1961	156 17 204 103	17 16 22 17	.00		56 93 96 95	194 261 242 251		172 170 270 200	0 0 0	635 947 856 839	252 305 305 295	.7 .4 1.0 .6	1.8 1.0 2.6 1.8	.26 .53 .28 .47	2080 1990	2.04 2.83 2.71 2.60	632 95.5 1100 531	740 990 1020 910	599 850 798 746	3.1 3.6 3.3 3.6	2100 2650 2630 2480	7.8 7.8 7.8

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

				15				Bi-	_						Dis	solved a	solids	Hard as Ca		So-	Specific	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pH
	and the second				c.	NORTH	FORK	RED R	IVER N	EAR CAR	TER, OKLA	Co	ntinu	ed								
ov. 14, 1961 bec. 5 an. 31, 1962 lar. 7 lar. 28	39 83 106 670 622					148 195 195 227 275		226 104 220 196 168	0 0 0 0	750 723 611 730 960	210 265 268 335 340		0.5 2.0 .0 		1700 1610 1510 1860 2160	2.31 2.19 2.05 2.53 2.94	179 361 432 352 128	940 790 770 900 1020	755 705 590 740 882	2.1 3.0 3.1 3.3 3.7	2190 2160 2100 2460 2780	8. 7. 8.
Apr. 25	b22 b160 b107 b.6 b86					247 229 137 128 129		96 210 232 190 232	0 0 0 0	955 625 430 610 460	315 345 185 175 172		=======================================		2040 1710 1160 1420 1180	2.77 2.33 1.58 1.93 1.60	121 739 335 2.30 274	980 810 600 760 630	902 638 410 604	3.4 3.5 2.4 2.0 2.2		7. 7. 8.
							D.	CACHE	CREEK	NEAR WA	ALTERS, OF	KLA.							1000			
Oct. 2, 1962 Oct. 23 Nov. 14 Dec. 4 Dec. 17	27.1 38.9 b38 757 49.5			1000		68 23					71 57 56 20 50							120 104		2.7 1.0	715 563 659 312 635	
Jan. 7, 1963 Jan. 30 Mar. 5 Apr. 3 Apr. 22	43.4 37.1 30.3 76.4 31.2					 51					57 56 68 54 72							 146		1.8	698 731 789 525 777	
May 21	31.7 1010 28.2 19.5 17.0					74 21 74					64 10 66 64 64							234 130 180		2.1 .8 2.4	774 346 715 677 685	
Aug. 27 Sept. 17	14.0 b18					65 65					60 54							172 170		2.2	665 638	
							E. I	EEP RE	D RUN	NEAR RA	NDLETT, C	KLA.										
Fov. 12, 1959 Jan. 6, 1960 Jan. 28 Jar. 9 Jar. 16	16 24.5 15.4 11.5 bl1			74 70 75 78 63	24 89 91 94 37	138 346 502 540 216		190 336 140 168 196	8 0 0 0	99 130	220 580 840 910 342	0.3 .7	1.8		658 886	0.89	28.4 26.3	284 540 560 580 310	115 264 446 442 150	3.6 6.5 9.2 9.8 5.3	1170 2560 3330 3610 1560	8. 8. 8. 8.
Apr. 1	9.2 b34 b4.0 24.0 b.2			66 66 54 24 45	57 19 13 6.8	314 121 107 34 85		400 208 200 120 234	0 8 6 0 4	69 47 19 36	430 175 140 30 93	=======================================	1.7 1.2 .2		655 533 182 438	.89 .72 .25	60.1 5.76 11.8 .24	400 242 188 88 184	0	6.8 3.4 3.4 1.6 2.7	2180 1020 845 310 687	8. 8. 7.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

Doto				Mag	183	Po-	Bi-	0						Di	ssolved s	solids	Hardness as CaCO ₃		So-	Specific	
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	tas- sium (K)	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
						F. V	ASHITA	RIVE	R NEAR I	DURWOOD,	OKLA.										
Water year 1951 Maximum, Jan. 21-31,1951. Minimum, May 2-3. Weighted average.	436 7850 1916	 	153 50 78	60 17 27	63 17 29	7	368 153 185	 8 	337 56 159	79 21 37	===	1.4 5.1 2.8		936 282 478	1.28 .39 .65	1100 5980 2470	628 195 306	327 56 154	==	1330 440 693	8
Water year 1952 Maximum, Dec. 11-20,1951. Minimum, Nov. 2 Weighted average.	233 1060 629	15	136 11 75	53 3.3 29	77 7 43	3.8	307 22 189	2 	337 23 169	95 4.5 46	0.3	1.3 10 5.2	0.24	912 70 499	1.24 .10 .68	574 200 847	558 41 306	302 23 151	 	1290 45 736	8 7
Water year 1953 Maximum, June 1-2, 1953 Minimum, May 13 Weighted average.	148 3410 518		136 32 65	49 4.6 19	103 12 35	1	137 111 162	 2	431 17 119	144 6.5 40	=	1.0 3.8 3.7		1050 140 390	1.43 .19 .53	422 1290 545	541 99 240	428 4	1.9 .5	1410 232 595	8 8
Water year 1954 Maximum, Sept.11-14,1954. Minimum, June 8 Weighted average.	16.2 7980 1258	Ξ	124 40 63	65 5.8 16	123 7 26	. 7	190 110 153	2 4	432 33 101	160 9.5 36		1.6 5.9 2.5		1050 160 350	1.43 .22 .48	46 3450 1190	575 124 223	416 25 98	2.2	1620 291 545	8 8
Water year 1955 Maximum, May 5-10, 1955 Minimum,	124		100	63	95		205	4	389	125		2.6		972	, 1.32	325	510		1.8	1390	8
Oct. 1, 1954 Weighted average.	4370 878		28 64	$\frac{2.4}{21}$	7 23	.9	93 157	==	16 112	5.4 33		3.9		110 362	.15	1300 858	80 246	4 118	. 4	193 562	8
Mater year 1956 Maximum, July 30, 1956 Minimum,Oct.1-4,	120	*==	144	60	139	i s	128		497	200	. 1	2.5		1160	1.58	376	605	500	2.5	1710	7.
7-10, 1955 Weighted average.	3589 440	==	51 87	15 35	14 46		126 185	<u></u>	80 211	18 62		3.9		264 573	.36	2560 681	188 361	84 210	.4 1.1	444 880	7
dater year 1957 Maximum, Jan. 23, 1957 Minimum,	188		100	49	182		164	4	269	300		2.3		1170	1.59	594	450	309	3.7	1830	8.
May 17-20 Weighted average.	64550 3555	==	36 54	7.5 16	8 20	. 5	130 162	==	18 70	8.8 26		2.5 3.2		169 303	.23	29450 2850	121 200	14 68	.3	268 465	8.
ater year 1958 Maximum, June 16-20,1958.	410		106	49	66		232	10	263	90		1.2		824	1.12	912	465	258	1.3	1090	8.
Minimum, Aug. 10-13 Weighted average.	1880 934	-55	46 84	11 31	21 43		152 229		42 149	26 62		1.4		222 522	.30	1130 1320	160 337	36	.7 1.0		7.
ater year 1959 Maximum,				ļ ;												1020	00.	100		Triangle Control	
Sept. 14-17,1959 Minimum, June 1-3. Weighted average.	224 4533 640	==	178 66 87	63 11 28	98 15 41		120 132 175	6	650 98 198	100 12 49	==	2.0 3.7 2.8		1210 280 531	1.65 .38 .72	732 3430 918	705 208 332	90	1.6 .4 1.0	1590 440 778	7. 8.

See footnotes at end of table.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

Date of collection								Bi-							Dia	solved a	solids	Hardness as CaCO ₃		So-	Specificon-	
	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	sium	car- bon- ate (HCO ₃	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (C1)		Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	pH
						F.	WASHIT	ra RIVI	ER NEA	R DURWOO	DD, OKLA	-Cont	inued									
Water year 1960																						
Maximum, Dec. 1-10, 1959. Minimum,	447			188	73	7	7	336		509	92		3.2		1200	1.63	1450	770	494	1.2	1560	8.
May 19-20, 1960. Weighted average.	17200 1594	==		46 94	11 32	1 4		142 202	6 	45 212	20 46		.2 2.4		229 572	.31 .78	10630 2460	162 366	36 40	.6	376 812	8.
Water year 1961																						
Maximum, Mar.17-24, 1961. Minimum,	953	9.0		114	41	68	2.6	248	2	250	110	0.3	1.2		1140	1.09	2930	455	248	1.4	1140	8.:
Dec. 9, 1960 Weighted average.	3040 1135	12		47 88	9.8 81	2 4		144 197	2 0	37 186	38 56	.3	5.0 2.4		262 564	.36	2150 1730	158 348	36 182	1.0	421 800	8
Water year 1962																						
Maximum, Feb. 11-20, 1962 Minimum, June 19.	498 9760			170 39	68 10	9	2	296 132	4 0	495 46	98 22		11		1190 221	1.62	1600 5820	705 140	456 32	1.5	1500 345	8.
Weighted average.	1345					4	3	199	3	213	49				581	.79	2110	363	195	1.0	840	-
Water year 1963 Maximum, July 23-28, 1963	80.8					10	9	142	0	670	130				1450	1.97	316	760	644	1.7	1730	8.
Minimum, Oct. 28-31, 1962 Weighted average.	5510 629			47 95	12 37	1 5	5	122 162	0	63 269	23 68	===	22		260 661	.35	3870 1120	166 390	66 254	.5 1.2	412 938	8.
Water year 1964																						
Maximum, May 1-3, 1964	254					10	5	124	8	695	138				1440	1.96	988	805	690	1.6	1760	8.
Minimum, Aug. 18. Weighted average.	722 340					1 4	7	124 144	0	37 207	15 54				190 552	.26	370 507	124 324	22 200	.7	328 777	8.3
weighted average.	340																77					
	A state						G. CI	LEAR BO	OGGY C	REEK NEA	AR CANEY,	OKLA.										
Oct. 19, 1961 Nov. 2	139 b65			59 51	13 21	23 39		212 214	0	21 21	38 70		0.2		273 324	0.37	102 56.9	200 212	26 37	0.7	461 573	
Dec. 7	b274			38	16	36		156	ő	23	62		. 1		274	.37	203	162	34	1.2	468	8.2
Mar. 2, 1962	102			85	26	56		320	0	30 31	106		.0		483 285	.66	133 198	320 196	58 21	1.4	871 467	8.
Apr. 4	258					26		206	4		32								30	1.4	457	
May 9	124 680			17		39 9.0		140 94	0	33 13	57 12				261 110	.35	87.4 202	144 88	11	1.4	208	7.0
Aug. 14	b14					56		190	0	11	115				388	. 53	14.7	208	52	1.7	668	8.
Sept. 11	b212					11		166	0	11	16				210	.29	120	146	10	. 4	332	8.3
Oct. 1	64					8.5		100	0	8.6	11				138	. 19	23.8	88	6	. 4	211	8.
Nov. 6	134					25		194	4	25	37				290	. 39	105	190	24	.8	463	
Dec. 4	879					18		154	2	20 34	29 60				229 301	.31	543 140	152 186	22 42	1.1	370 530	
Jan. 2, 1963	173 78				25	36 50		176 204	0	34	90				378	.51	79.6	220	53	1.5	669	
Jan. 31 Feb. 26	56					57		170	2	33	101				357	.48	54.0	196	53	1.8	643	

See footnotes at end of table.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

Date of collection ar. 26, 1963 pr. 23 ay 21 une 18 uly 16 ug. 12 ept. 9 ec. 5 ec. 30 an. 6, 1964		107 68 58 26	Iron (Fe)	Cal- cium (Ca)	Mag- ne- sium (Mg)	Sodium (Na)	Po- tas- sium (K)	Bi- car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)		trate	Bo-	Milli-	Tons	Tons	Cal-	Non-	dium ad- sorp-	duct- ance	pl
pr. 23		68 58						bon-	bon- ate (CO ₃)	(SO ₄)	(C1)	ride	trate (NO ₃)	(B)	grams per liter	per acre- foot	per day	Mag- ne- sium	bon- ate	r- tion mhore ratio 25°	mhos at 25°C)	e pE
pr. 23		68 58				G. CL	EAR BO	OGGY CI	REEK N	EAR CAN	EY, OKLA	Cont	inued	ı								
ay 21		58		44		30		196	0	30	47				290	0.39	83.8	192	31	1.0	504	8
une 18 uly 16 ug. 12 ept. 9 ec. 5 ec. 18 ec. 30 an. 6, 1964				77	100						- T-				309	. 42	56.7	194	37	. 9	504	8
uly 16 ug. 12 ept. 9 ec. 5 ec. 18 ec. 30 an. 6, 1964		26		26	22	53		136	4	30	86				319	. 43	50	156	38	1.8	537	8
ug. 12ept. 9ec. 5ec. 18ec. 30		4		27	24	63		144	4	22	112				347	. 47	24.4	168	43	2.1	605	8
ept. 9 ec. 5 ec. 18 ec. 30 an. 6, 1964		14		30	28	61		166	4	17	116				358	. 49	13.5	190	47	1.9	649	8
ec. 5ec. 18ec. 30		21		38	28	64		190	4	17	122				395	. 54	22.4	212	50	1.9	697	8
ec. 18 ec. 30 an. 6, 1964		3.5		54	28	66		228	8	15	124				443	.60	4.19	248	48	1.8	757	8
ec. 30 an. 6, 1964		8.6									52										631	
an. 6, 1964		9.9				W					58				.55		37	. 77	77		658	
		9.2				62		262	4	23	140				480	. 65	11.9	308	86	1.5	878	
		10				28		270	0	27	65				379	. 52	10.2	280	59	.7	639	
an. 20		11			55-						67										527	
an. 28		8.6				44					82						100 at t	800		9-	689	
eb. 6		20				38		196	4	21	76				322	. 44	17.4	214	47	1.1	563	
eb. 12		17									81										648	
eb. 17		14									80							100	844		675	
eb. 24		11									137						SEPARA		100		858	
ar. 2		11				46		162	4	35	95				366	.50	10.9	210	70	1.4	644	٤
ar. 16		52									46					22		1,00	44	142	507	
ar. 20		666				11		108	0	21	21				192	.26	345	116	27	. 4	284	
ar. 24		90									30										422	
ar. 31		31									80					22	Blader -				612	
pr. 6		1650			_ II	16		176	0	20	27				233	.32	1040	168	24	. 5	393	8
pr. 14		89								20	34					.52	1010				470	
pr. 21		50									86										693	
ay 1		34									64				445				70		624	
ay 7		29				32		260	0	33	80				445	.61	34.8	292	79	. 8	709	
ay 11		2670				13		136	0	15	18 17				159	.22	1150	124	12	. 5	293 341	
ay 15ay 22		326 68						400	- 11	324	26										442	
																A						
ay 28		50									37				LEE					77	497	
une 10		28				24		242	0	31	40				328	. 45	24.8	234	35	. 7	530	
une 18		2700								7.0	7.2				===	77					129	
une 19		3380		- 55		3.2		60	0	7.0	6.4				74	.10	675	58	9	. 2	123	
uly 13		6.1				25		212	0	20	43				292	.40	4.81	200	26	.8	496	{
uly 28		1.1				27		216	0	21	60				308	. 42	.91	224	47	. 8	552	8
ug. 5		. 2									70										589	
ug. 19		.1				37		222	0	23	72				337	. 46	.09	226	44	1.1	606	
ug. 25		5.1									103				775						656	
ept. 3		29									21										301	
ept. 14		3.9						-			30						122				391	
ept. 23		2940									6.0					. 10 2 1					202	
ept. 29	The San	622																THE RESERVE OF THE PARTY OF THE		COLUMN TWO IS NOT THE OWNER.	202	

See footnotes at end of table.

Table 7.--Chemical Analyses of Water From Streams at Selected Sites in the Red River Basin, Oklahoma--Continued

(Results in milligrams per liter except as indicated. Calculated values for sodium plus potassium are centered between the two columns.)

				Cal- cium (Ca)			Po-	Bi-							Dis	solved s	olids	Hard as Ca	ness aCO ₃	So-	Specific con-	2
Date of collection	Discharge (cfs)	Silica (SiO ₂)	Iron (Fe)		Mag- ne- sium (Mg)	Sodium (Na)	tas-	car- bon- ate (HCO ₃)	Car- bon- ate (CO ₃)	Sulfate (SO ₄)	Chloride (Cl)	Tiue	Ni- trate (NO ₃)	Bo- ron (B)	Milli- grams per liter	Tons per acre- foot	Tons per day	Cal- cium, Mag- ne- sium	Non- car- bon- ate	dium ad- sorp- tion ratio	duct- ance (micro- mhos at 25°C)	
							н. кі	AMICHI	RIVE	R NEAR I	BELZONI,	OKLA.										
Oct. 18, 1961				4.2	3.2	3.4		22	0	7.2	4.3		0.0		44	0.66	38.5	24	6	0.3	52	7.
Nov. 2	133			5.3	1.7	7.6		24	0	6.6	6.1		1.7		63	. 09	22.6	20	1	.7	68	7.
Dec. 14				3.2	1.7	5.8		16	0	7.0	4.6		. 8		43	.06	322	15	2	.6	45	
Mar. 2, 1962				2.4	1.9	6.9		16	0	8.2	3.5		.7		42	.06	236	14	1	.9	49	
Apr. 3	2910			2.2	1.6	5.3		16	0	6.2	2.6		. 7		51	.07	401	12	0	.7	49	7.:
May 3	1540			4.6	1.3	6.2		24	0	6.2	3.0		. 6		56	. 08	233	17	0	.7	61	7.4
June 20	264			4.2	1.6	6.0		22	0	6.6	3.6		.2		44	.06	31.4	17	0	. 6	63	
Aug. 14	5.3			6.4	1.9	9.4		36	0	5.2	6.2		1.0		61	. 08	.87	24	0	.8	90	
Sept. 13	236			3.7	2.2	8.3		26	0	6.0	5.4		1.0		51	. 07	32.5	18	0	. 8	67	
Nov. 8	427	14		4.8	1.0	6.0		18	0	6.0	5.6				41	.06	47.3	16	1	. 6	57	
Dec. 6	1785	10		4.8	1.0	5.3		16	0	6.2	5.6				46	. 06	222	16	3	. 6	54	7.0
Jan. 4, 1963	392	11		5.2	.7	6.0		16	0	6.4	6.6		==		44	.06	46.6	16	3	.6	60	
Jan. 31	194	11		4.0	1.5	8.7		20	0	7.8	7.4				41	.06	21.5	16	0	1.0	68	
Feb. 28	92	6.8		8.0	1.0	8.7		24	0	9.2	9.8				50	. 07	12.4	24	4	.8	88	
Mar. 28		17		4.0	1.5	5.5		16	0	5.8	6.3				51	. 07	420	16	3	. 6	57	
Apr. 25	210	13		7.2	. 5	8.3		24	0	7.0	7.7										77	
May 23	63	16		6.4	1.0	6.7		24	0	5.8	6.4				46 47	. 06	26.1 8.00	20 20	0	.8	70	
June 20	73	9.6		7.2	1.0	6.2		22	0	7.4	7.1				51	. 07	10.0	22	4	.6	79	
July 19	30	11		4.0	1.9	3.0		16	0	5.4	4.4				52	.07	4.21	18	5	.3	56	
Aug. 14	9.4	8.4		3.2	1.9	4.8		20	0	4.0	4.2				49	.07	1.24	16	0	. 5	54	
Sept. 11	4.9	8.4		3.2	2.4	5.3		20	0	4.3	6.0		==		48	. 07	. 64	18	2	. 5	60	

a Flow shown for maximum, minimum, and weighted average is mean discharge for the period.

b Field estimate.

EXPLANATION

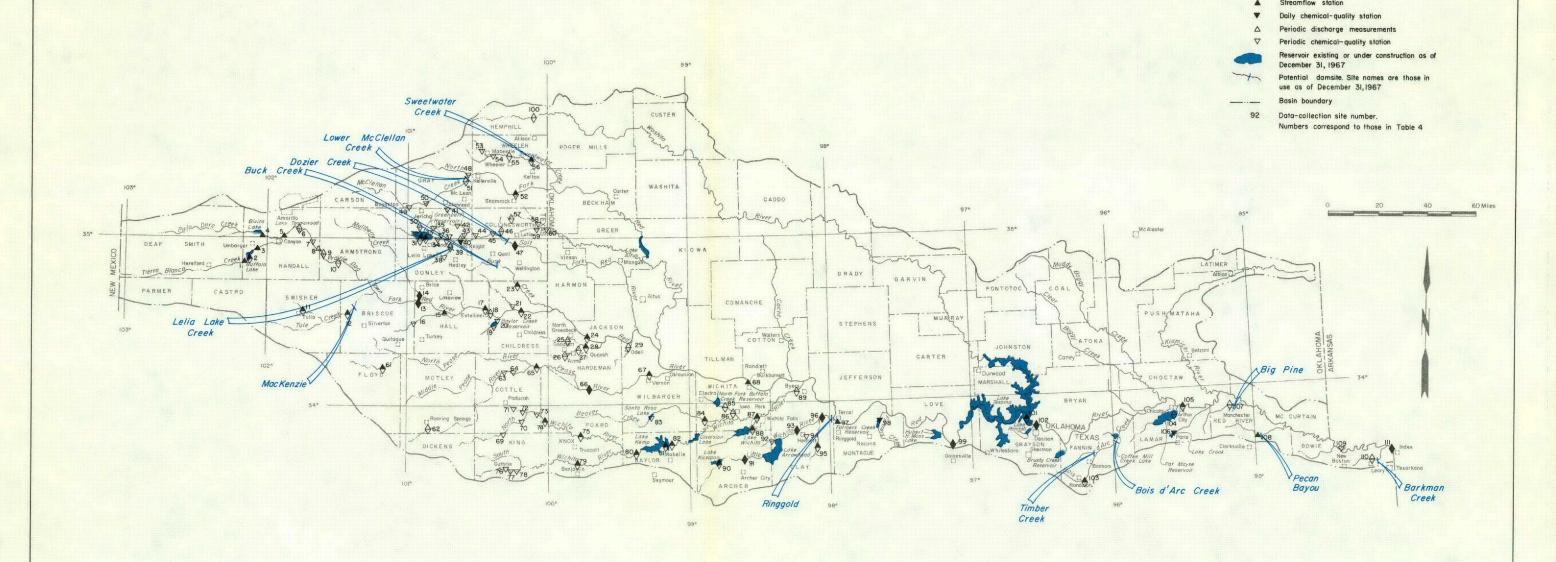
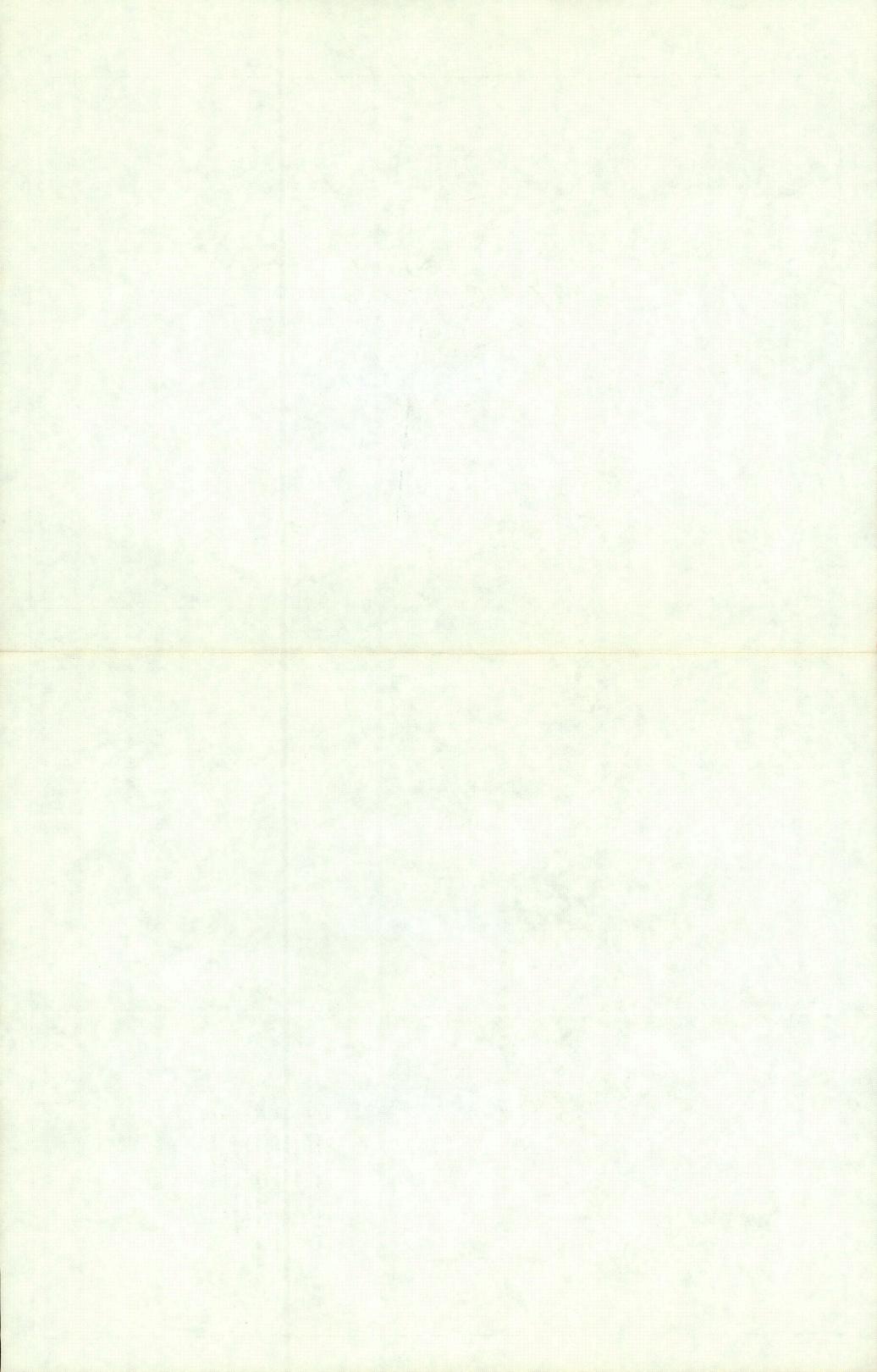
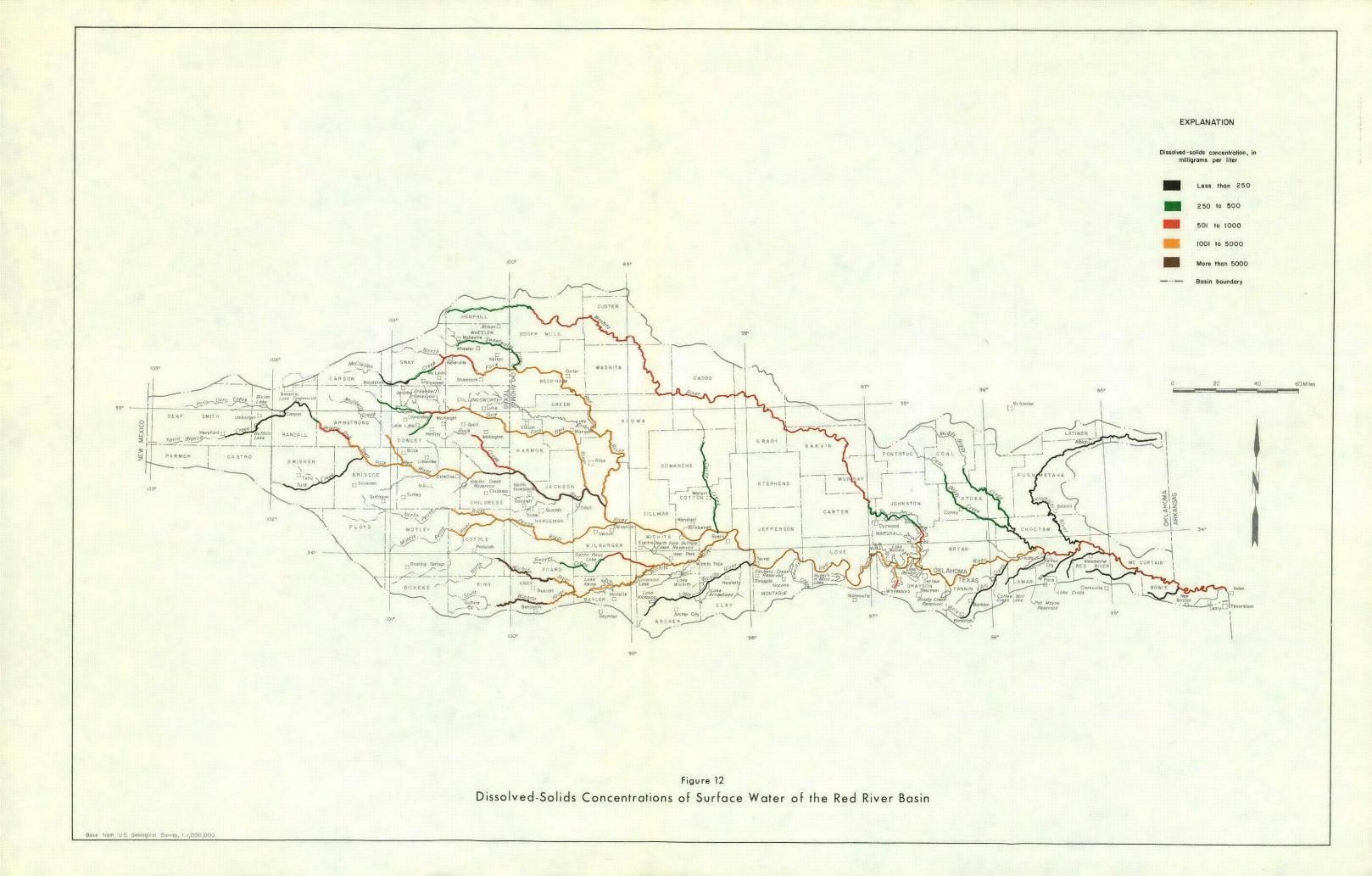
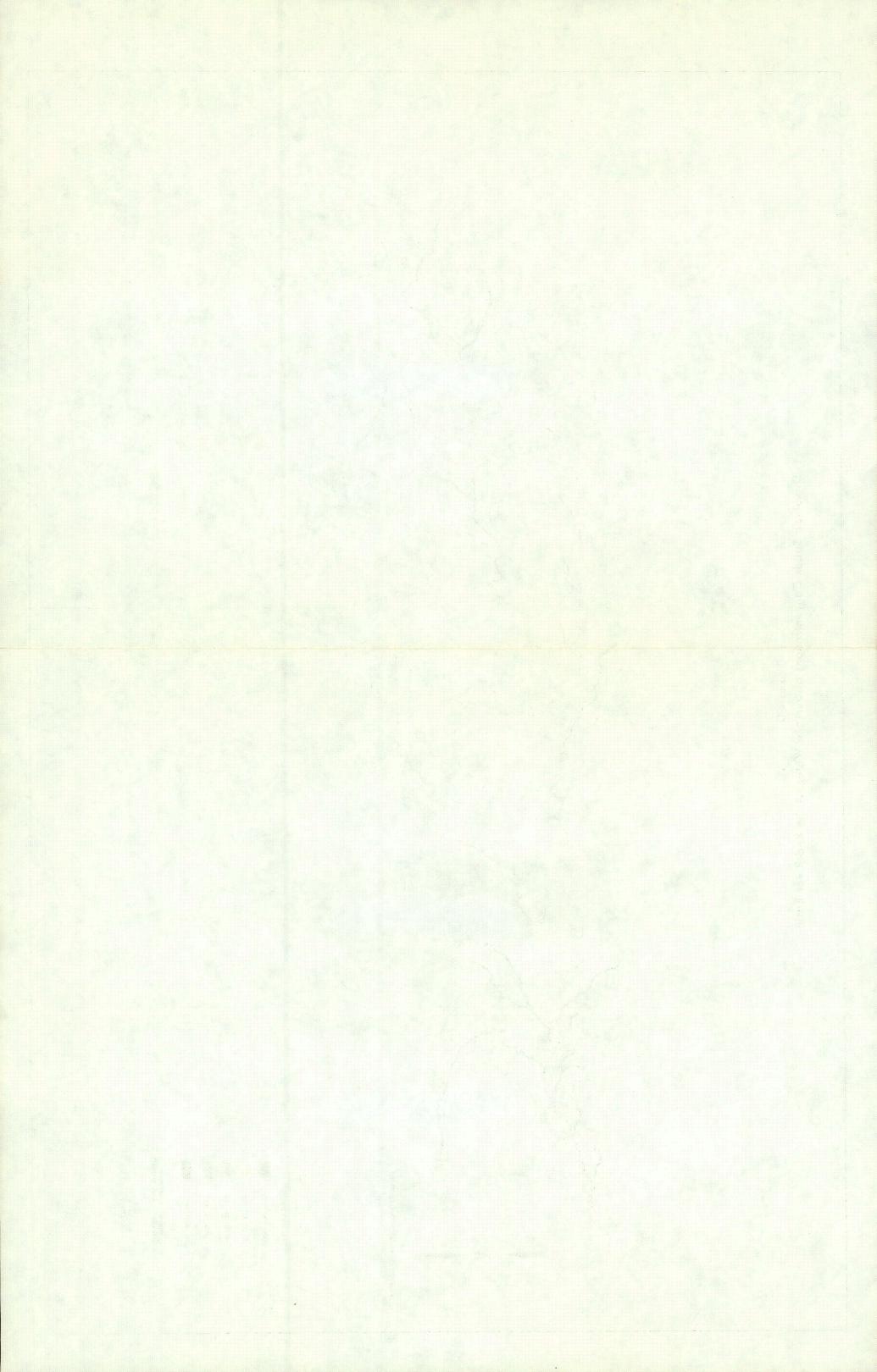


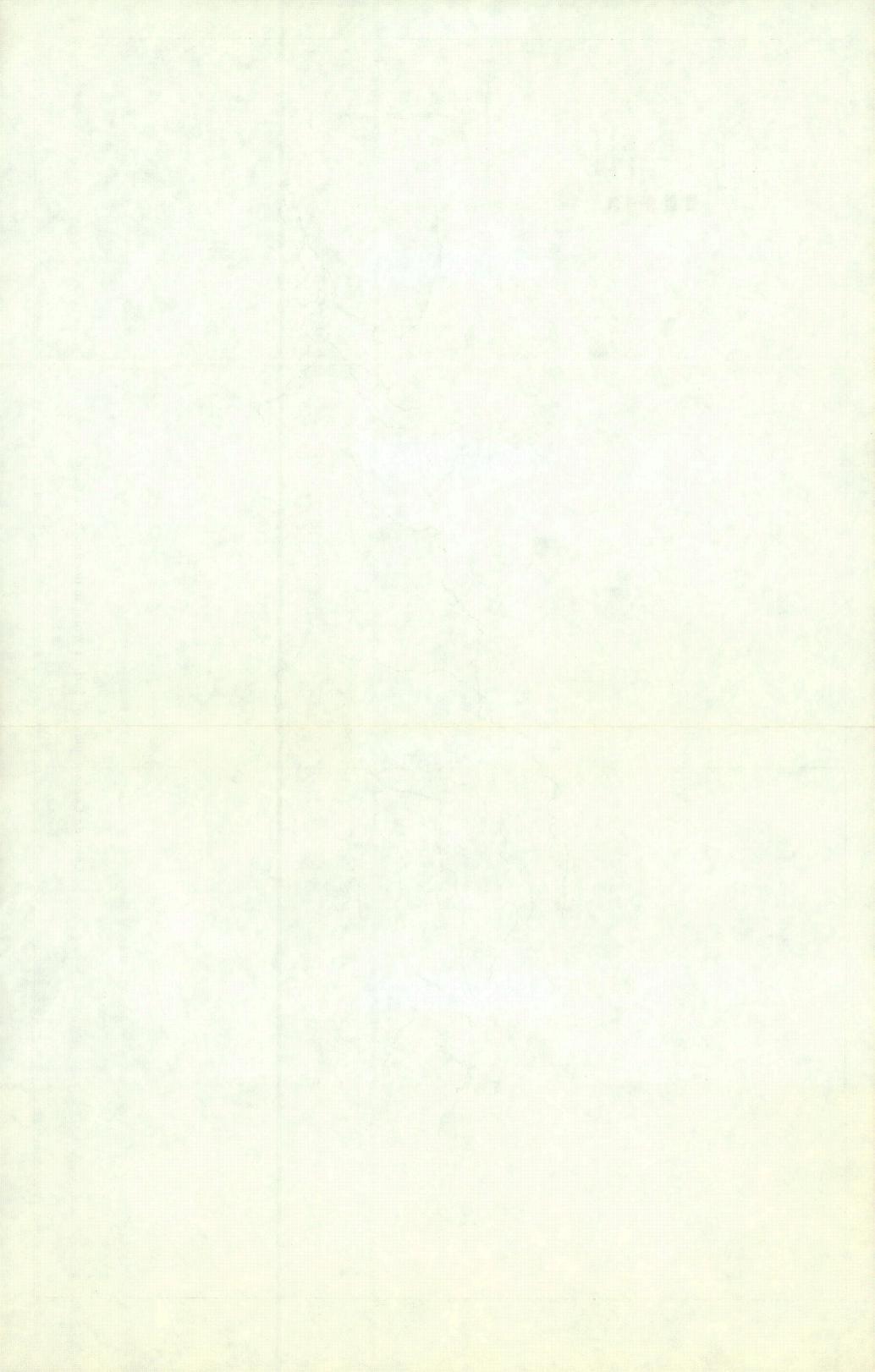
Figure 11
Location of Streamflow and Chemical-Quality Data-Collection Sites, Major Existing Reservoirs, and Potential Reservoir Sites in Texas

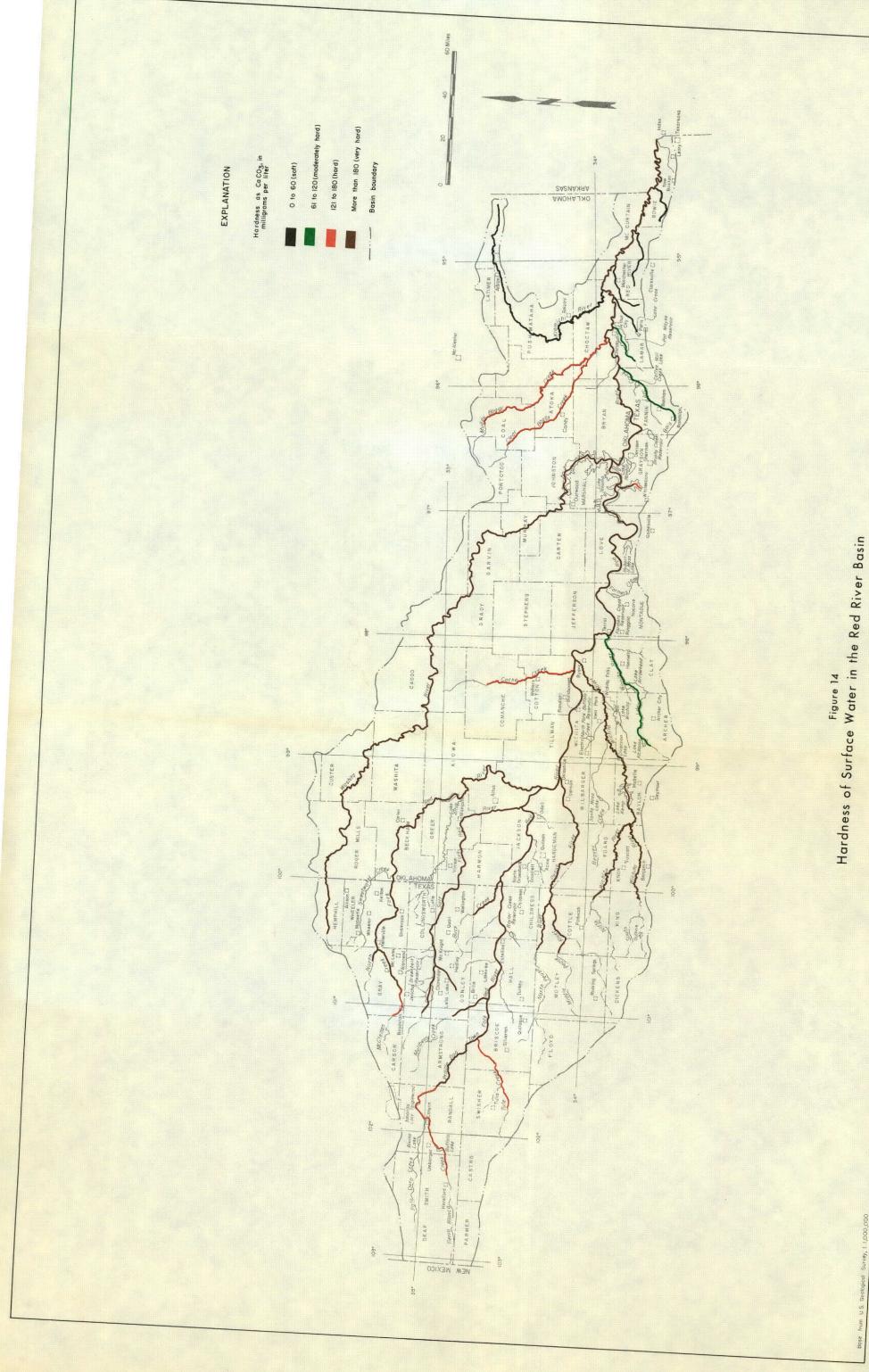


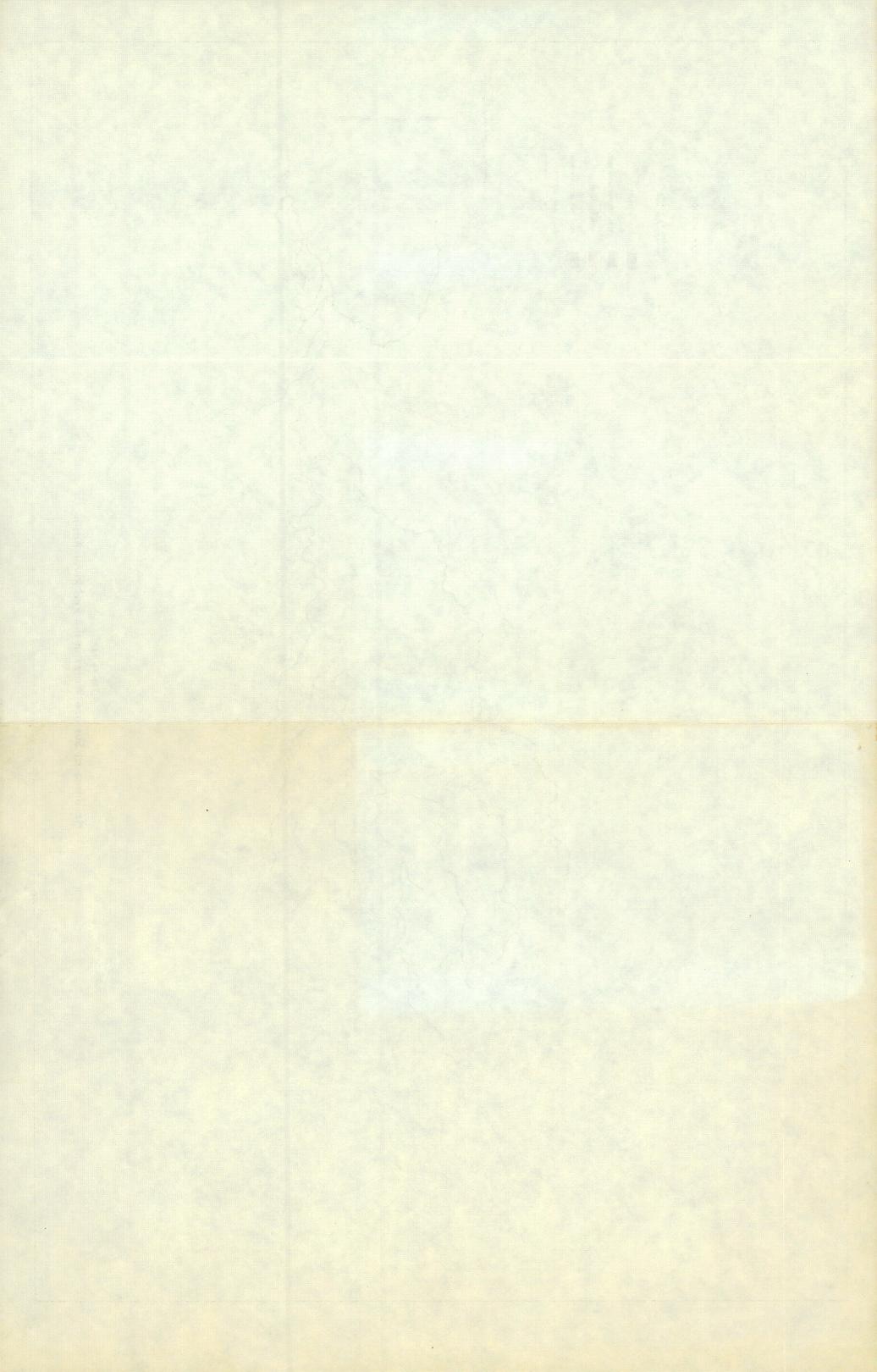












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